



# Europäisches Patentamt European Patent Office Office européen des brevets



11 Publication number:

0 604 957 A1

(12)

# **EUROPEAN PATENT APPLICATION**

21) Application number: 93120970.4

2 Date of filing: 27.12.93

(a) Int. Cl.<sup>5</sup>: **C07K** 5/06, C07K 5/08, C07K 5/10, C07K 7/06, A61K 37/02

16-1-502, Suzurandainishimachi 5-chome,

Priority: 28.12.92 JP 349062/92 16.03.93 JP 56185/93 22.07.93 JP 181735/93

- 43 Date of publication of application: 06.07.94 Bulletin 94/27
- Designated Contracting States:
  AT BE CH DE DK ES FR GB GR IE IT LI LU NL
  PT SE
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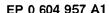
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- 2-Amino-6,7-dihydroxy-4-thiaheptanoic acid derivatives, production and use thereof.
- (57) A compound of the formula:

wherein each of R<sup>1</sup> and R<sup>2</sup> is hydrogen or aliphatic acyl, A is amino which may be protected, X is an amino acid sequence consisting of 1 to 10 amino acid residues which contain at least one amino acid residue having a water-solubility enhancing group, or a salt thereof has an activity of remarkably improving hematopoietic disorder and is useful as an immuno-stimulating agent or an agent for treating thrombocytopenia.



# FIELD OF THE INVENTION

This invention relates to 2-amino-6,7-dihydroxy-4-thiaheptanoic acid derivatives, which are useful as therapeutic agents of leukocytopenia caused by various reasons, diseases due to decrease of leukocyte, diseases requiring, from the therapeutic viewpoint, increase of bone marrow cells or leukocyte, throm-bocytopenia caused by various reasons, diseases due to decrease of thrombocyte, or diseases requiring, from the therapeutic viewpoint, increase of thrombocyte.

### BACKGROUND OF THE INVENTION

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In Hoppe-Seyler's Zeitschrift für Physiologiche Chemie 364, pp 593-606 (1983), a synthetic peptide derived from lipoprotein produced by E. coli, which is shown by the formula:

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is disclosed.

And, in JPA H4(1992)-046194, WS 1279A substance shown by the formula:

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$$\begin{array}{c} \text{O-COC}_{15}\text{H}_{31} \\ | \\ \text{CH}_2\text{-S-CH}_2\text{-CH-CH}_2\text{-O-COC}_{15}\text{H}_{31} \\ | \\ \text{C}_{15}\text{H}_{31}\text{CONHCH-CO-Asn-Ser-Gly-Ser-OH} \end{array}$$

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is disclosed.

Achiwa et al. synthesized these compounds as optical active compounds. [cf. JPA H4(1992)-099796, Chem. Pharm. Bull. 39, p 2590 (1991) and Peptide Chemistry, p. 361 (1991)].

However, the compounds of this invention are not described in these references.

Incidentally, abbreviations of amino acid, peptide or the like used in the present invention are based on those in accordance with IUPAC-IUB Commission on Biochemical Nomenclature or those conventionally used in the relevant fields, and, possible optical isomers of amino acid are, unless otherwise specified, Lisomers.

Chemotherapy or radiotherapy of cancers causes serious leukocytopenia or serious thromobocytopenia. The former induces lowering of resistance against infections or other various diseases so that sufficient therapeutic effects are not expected. The latter induces insufficiency of hemostatic mechanism so that sufficient therapeutic effects are not expected. These are being taken up as a grave concern in the field of cancer therapy. Under such circumstances, development of drugs, which mitigate the suppression of hematopoietic function caused by these therapeutic methods and are capable of promoting the recovery of leukocyte number or thrombocyte number, has been ardently desired. Further, in the therapy by bone marrow transplantation, drugs capable of promoting the proliferation of bone marrow cells then transplanted and capable of recovering the number of leukocyte promptly are desired. Furthermore, these drugs can be used for therapeutic agents of thrombocytopenia after bone marrow transplantation or autoimmunodisease accompanied by thombocytopenia, such as aplastic anemia and paroxymal thrombocytopenic purpura.

While taking the present circumstances mentioned above into consideration, the present inventors persued their studies, from a fresh viewpoint, on compounds having the action of increasing the number of leukocyte. As the result, the present inventors found that the novel 2-amino-6,7-dihydroxy-4-thiaheptanoic acid derivatives promote the proliferation of bone marrow cells of mice and increase the number of peripheral leukocytes. Further, said compounds also stimulate bone marrow cells of mice so that promote proliferation and differentiation of megakaryocytes. Based on these findings, the present inventors made further studies to complete the present invention.

# SUMMARY OF THE INVENTION

This invention is to provide:

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1) A compound of the formula (I):

$$\begin{array}{c|c}
OR^{2} \\
\mid & (I)
\end{array}$$

$$CH_{2}-S-CH_{2}-CH-CH_{2}-O-R^{1} \\
\mid & .$$

$$A-CH-CO-X-OH$$

wherein each of R<sup>1</sup> and R<sup>2</sup> is hydrogen or aliphatic acyl, A is amino which may be protected, X is an amino acid sequence consisting of 1 to 10 amino acid residues which contain at least one amino acid residue having a water-solubility enhancing group, or a salt thereof.

- 2) A compound according to 1), wherein A is amino.
- 3) A compound according to 1), wherein A is amino which may be substituted with substituted oxycarbonyl.
- 4) A compound according to 1), wherein the amino acid residue having a water-solubility enhancing group is an acidic amino acid residue.
  - 5) A compound according to 1), wherein the amino acid residue having a water-solubility enhancing group is a basic amino acid residue.
  - 6) A compound according to 1), wherein aliphatic acyl is  $C_{2-30}$  aliphatic acyl.
  - 7) A compound according to 1), wherein at least one of R1 and R2 is aliphatic acyl.
  - 8) A compound according to 1), wherein R1 is aliphatic acyl.
  - 9) A compound according to 1), wherein R2 is aliphatic acyl.
  - 10) A compound according to 1), wherein the compound is (2R,6R)-2-amino-6,7-bis(hexadecanoyloxy)-4-thiaheptanoyl-glycyl-glutamyl-glutamic acid.
  - 11) A compound according to 1), wherein the compound is (2R,6R)-2-amino-6,7-bis(hexadecanoyloxy)-4-thiaheptanoyl-glycyl-glycyl-glutamic acid.
  - 12) A compound according to 1), wherein the compound is (2R,6R)-2-amino-6,7-bis(hexadecanoyloxy)-4-thiaheptanoyl-glutamyl-glycyl-glutamic acid.
  - 13) A compound according to 1), wherein the compound is (2R,6R)-2-amino-6,7-bis(hexadecanoyloxy)-4-thiaheptanoyl-glycyl-D-glutamic acid.
  - 14) An immuno-stimulating composition having a leukocyte-increasing action, which comprises a compound or a salt thereof as defined in 1).
  - 15) An immuno-stimulating composition according to 14), wherein at least one of  $R^1$  and  $R^2$  is aliphatic acyl.
  - 16) A composition for treating thrombocytopenia, which comprises a compound or a salt thereof as defined in 1).
    - 17) A method of producing the compound or a salt thereof as defined in 1), which comprises subjecting a compound of the formula (II):

$$\begin{array}{c|c}
OR^{2} \\
| \\
CH_{2}-S-CH_{2}-CH-CH_{2}-O-R^{1} \\
| \\
A-CH-CO-X'-OR^{3}
\end{array}$$
(II)

wherein each of R<sup>1</sup> and R<sup>2</sup> is hydrogen or aliphatic acyl, R<sup>3</sup> is a protecting group, A is amino which may be protected, X' is an amino acid sequence consisting of 1 to 10 optionally protected amino acid residues which contain at least one optionally protected amino acid residue having a water-solubility enhancing group, or its salt, to a deprotection reaction.

# DETAILED DESCRIPTION OF THE INVENTION

Referring to the formulae (I) and (II), examples of the aliphatic acyl group shown by R¹ or R² include aliphatic acyl groups derived from aliphatic carboxylic acid. Examples of the aliphatic acyl groups include saturated or unsaturated aliphatic acyl groups having a maximum of 34 carbon atoms (e.g. formyl, acetyl, propionyl, butyryl, isobutyryl, valeryl, isovaleryl, hexanoyl, heptnoyl octanoyl, decanoyl, dodecanoyl, tetradecanoyl, pentadecanoyl, hexadecanoyl, heptadecanoyl, octadecanoyl, nonadecanoyl, tetracosanoyl, hexacosanoyl, icosanoyl, heneicosanoyl, docosanoyl, tetracosanoyl, hexacosanoyl, ethyltridecanoyl, methyltridecanoyl, methyltetradecanoyl, ethyltetradecanoyl, methyltetradecanoyl, methylpentadecanoyl, ethylpentadecanoyl, methyloctadecanoyl, ethylocatadecanoyl, octacosanoyl, triacontanoyl, dotriancotanyol, tetratrianiotanoyl, acryloyl, propioloyl, methacryloyl, crotonoyl, isocrotonoyl, myristoleoyl, oleoyl, palmitoleoyl, elaidoyl, cis, cis-9,12-octadecatrienoyl, 9,12,15-octadecatrienoyl, 9,11,13-octadecatrienoyl, 5,8,11,14-icosatetraenoyl, cis-15-tetracosaenoyl, etc.).

Preferable examples of aliphatic acyl groups include  $C_{2-30}$  saturated or unsaturated aliphatic acyl groups (e.g. acetyl, propionyl, butyryl, isobutyryl, valeryl, isovaleryl, hexanoyl, heptanoyl, octanoyl, decanoyl, dodecanoyl, tridecanoyl, tetradecanoyl, pentadecanoyl, hexadecanoyl, heptadecanoyl, octadecanoyl, nonadecanoyl, icosanoyl, heneicosanoyl, docosanoyl, ethyldodecanoyl, methyltridecanoyl, ethyltridecanoyl, methyltetradecanoyl, ethyltridecanoyl, ethylpentadecanoyl, ethylpentadecanoyl, methylhexadecanoyl, ethylhexadecanoyl, methylhexadecanoyl, ethylocatadecanoyl, triacontanoyl, myristoleoyl, palmitoleoyl, elaidoyl, cis, cis-9,12-octadecatrienoyl, 9,12,15-octadecatrienoyl, 9,11,13-octadecatrienoyl, 5,8,11,14-icosatetraenoyl, cis-15-tetracosaenoyl, etc.).

Especially preferable examples of the aliphatic acyl groups shown by  $R^1$  or  $R^2$  include  $C_{2-16}$  saturated or unsaturated aliphatic acyl groups.

Preferably, at least one of R1 and R2 is an aliphatic acyl group.

More preferably, both R<sup>1</sup> and R<sup>2</sup> are an aliphatic acyl group.

In the above formulae (I) and (II), examples of protective groups in the optionally protected amino groups shown by A include formyI, $C_{6-14}$  aryl carbonyl (e.g. phenyl carbonyl), substituted oxycarbonyl [e.g.  $C_{1-6}$  alkyloxy carbonyl (e.g. methoxy carbonyl, etc.),  $C_{6-14}$  aryloxy carbonyl (e.g. phenyloxy carbonyl, etc.), 9-fluorenyl methyloxy carbonyl, $C_{7-19}$  aralkyloxycarbonyl (e.g. benzyloxycarbonyl, etc.), adamantyloxy-carbonyl and so on], $C_{7-19}$  aralkylcarbonyl (e.g. benzylcarbonyl,etc.) $C_{7-19}$  aralkyl (e.g. benzylcarbonyl,trityl,etc.)These protective groups other than formyl may be substituted. Examples of the substituents include halogen atoms (e.g. fluoro, chloro, bromo and iodo),  $C_{1-6}$  alkyl carbonyl (e.g. acetyl, ethyl carbonyl, butyl carbonyl, etc.) and nitro group. The number of substituents ranges from 1 to 3.

A has preferably the above-mentioned meanings when R¹ and R² are both aliphatic acyl groups.

Preferably, A is an amino group which may be protected by the substituted oxycarbonyl. More preferably, A is an amino group.

In the formula (I), the amino acid in the amino acid sequence shown by X means a compound having in its molecule an amino group and an acidic group (e.g. carboxyl group, sulfonic acid group). Preferable examples of the amino acid include those described in Dai Yuhki Kagaku (Encyclopedia of Organic Chemistry) Vol. 21 "Natural Polymers III" compiled under the supervision of Dr. Munio Kotake, Published by Asakura Shoten in Japan, 1960 and "Amino acids and peptides" by Chapman and Hall, compiled by J.S.Davies, 1985 in USA.

To state more concretely, there may be mentioned, for example, amino acid constituting protein [e.g. aliphatic monoamino monocarboxylic acid such as glycine, alanine, valine, leucine, isoleucine or the like, aliphatic hydroxyamino acid such as serine, threonine or the like, acidic amino acid such as aspartic acid, glutamic acid or the like, acidic amino acid amide such as aspargine, glutamine or the like, aromatic amino acid such as phenylalanine, tyrosine, tryptophane or the like, iminocarboxylic acid such as proline, hydroxyproline or the like, basic amino acid such as arginine, lysine, histidine or the like, and sulfurcontaining amino acid such as methionine, cystine, cysteine or the like], amino acid obtained from natural sources as, for example, metabolites of microorganisms or components of animals and plants [e.g. aliphatic monoamino monocarboxylic acid such as L- $\alpha$ -aminobutyric acid,  $\gamma$ -aminobutyric acid,  $\beta$ -aminoisobutyric acid,  $\beta$ -alanine, homoserine,  $\alpha$ -methyl-D-serine, O-carbamyl-D-serine,  $\delta$ -hydroxy- $\gamma$ -oxo-norvaline,or the like, monoamino dicarboxylic acid such as L- $\alpha$ -aminoadipic acid, L- $\beta$ -aminoadipic acid, L-theanine, L- $\gamma$ -methylene glutamic acid, L- $\gamma$ -methyl glutamic acid or the like, diaminomonocarboxylic acid such as L-ornithine,  $\beta$ -lysine,  $\alpha$ , $\beta$ -diaminopropionic acid, L- $\alpha$ , $\gamma$ -diaminobutyric acid, or the like, diaminodicarboxylic acid such as diaminopimelic acid or the like, sulfonic acid-containing monoaminomonocarboxylic acid such

as cysteic acid or the like, sulfonic acid-containing amino acid such as taurine or the like, aromatic amino acid such as kynurenine, 3,4-dioxyphenyl-L-alanine or the like, heterocyclic amino acid such as 2,3-dicarboxyaziridine, [S]-2-amino-3-(isoxazolin-5-one-4-yl)-propionic acid, anticapsin or the like, basic amino acid such as L-4-oxalysine, L-4-oxolysine, [3R,5R]-3,6-diamino-5-hydroxyhexanoic acid, or the like, sulfur-containing amino acid such as lanthionine, S-methyl-L-cysteine or the like, cyclic amino acid such as pipecolic acid, azetidine-2-carboxylic acid, [1R, 2S]-2-aminocyclopentane-1-carboxylic acid, or the like, amino acid substituted with a specific functional group such as citrulline, alanosine, L-azaserine, or the like], and amino acids obtained by organic synthesis [e.g. 6-aminohexanoic acid, 8-aminooctanoic acid, 12-aminododecanoic acid, 4-aminobenzoic acid, 4-(aminomethyl)benzoic acid, 4-(N-(carboxymethyl)-aminomethyl)benzoic acid, etc.]

The amino acid residue means a divalent group which is derived from amino acid and has bonds to the amino group and acidic group, respectively.

Incidentally, even the group is unstable as amino acid, in the case where the amino group is acylated or the acidic group has amido-linkage, the group can be used as an amino acid residual group. Examples of such amino acid include (6-aminohexyl)carbamic acid.

Referring to the formula (I), examples of the amino acid having a water-solubility enhancing group in the amino acid sequence shown by X include the above-mentioned acidic amino acid and basic amino acid, preferably, the acidic amino acid.

Examples of the acidic amino acid include a compound having one or more acidic functional groups (e.g. carboxyl group, sulfonic acid group, etc.) in addition to a carboxyl group and an amino group. Preferable examples of the acidic amino acid include a compound having one amino group and two or more carboxyl groups. To state more concretely, there may be mentioned, for example, amino dicarboxylic acid (e.g. aspartic acid, glutamic acid, L- $\alpha$ -aminoadipic acid, L- $\beta$ -aminoadipic acid, 2,3-dicarboxyaziridine, etc.), more preferably,  $\alpha$ -aminodicarboxylic acid (e.g. aspartic acid, glutamic acid, L- $\alpha$ -aminoadipic acid, etc.), and, among them, aspartic acid and glutamic acid are especially preferable.

Examples of the basic amino acid include a compound having one or more basic functional groups (e.g. amino, a basic nitrogen containing heterocyclic group such as imidazolyl, indolyl, etc., guanidino, and so on). Preferable examples of the basic amino acid include a compound having one amino group, one carboxyl group and one or more said basic functional groups.

To state more concretely, there may be mentioned, for example, a basic amino acid constituting protein (e.g. arginine, lysine, histidine, etc.) and a basic amino acid obtained form natural sources as, for example, metabolites of microorganisms or components of animals and plants (e.g. diamino carboxylic acid such as L-ornithine,  $\beta$ -lysine,  $\alpha$ ,  $\beta$ - diaminopropionic acid, L- $\alpha$ ,  $\gamma$ -diaminobutyric acid, or the like, L-4-oxalysine, L-4-oxolysine, [3R,5R]-3,6-diamino-5-hydroxyhexanoic acid, etc.), more preferably, a basic  $\alpha$ -amino acid constituting protein (e.g. arginine, lysine, histidine, etc.) and a basic  $\alpha$ -amino acid obtained from natural sources as, for example, metabolites of microorganisms or components of animals and plants (e.g. L-ornithine,  $\beta$ -lysine,  $\alpha$ ,  $\beta$ -diaminopropionic acid, L- $\alpha$ ,  $\gamma$ -diaminobutyric acid, L-4-oxalysine, L-4-oxolysine, etc.), and among them, lysine, arginine and histidine are especially preferable.

In the formula (II), as the protecting groups shown by R<sup>3</sup>, use is preferably made of protecting groups of carboxyl group described later.

In the formula (II), the amino acid in the optionally protected amino acid residue in the amino acid sequence shown by X' has the same meaning as that in the amino acid residue in the afore-mentioned amino acid sequence shown by X.

In the formula (II), the amino acid in the optionally protected amino acid residue having a group which enhances the water-solubility of the amino acid sequence shown by X' has the same meaning as that in the amino acid residue having a group which enhances the water-solubility of the afore-mentioned amino acid sequence shown by X.

As the protecting group, mention is made of known protecting groups employed for protecting amino group, carboxyl group or hydroxyl group in peptide synthesis. These are protecting groups which can be removed by, for example, hydrolysis, hydrogenolysis, reduction, aminolysis or hydrazinolysis.

Examples of the amino-protecting groups include urethane-type protecting groups (e.g. benzyloxycarbonyl, t-butoxycarbonyl, allyloxycarbonyl, t-amyloxycarbonyl, isobornyloxycarbonyl, 4-methoxybenzyloxycarbonyl, 2-chlorobenzyloxycarbonyl, adamantyloxycarbonyl, 9-fluorenylmethyloxycarbonyl, 2,2,2-trichloroethyloxycarbonyl, or the like), acyl-type protecting groups (e.g.  $C_{1-6}$  lower fatty acid residues optionally having such a substituent as halogen, e.g. formyl, acetyl, propyl, trifluoroacetyl, chloroacetyl, or the like, phthalyl, tosyl, 2-nitrosulfenyl, 4-methoxy-2-nitrosulfenyl, benzoyl, or the like), and alkyl-type protecting groups (e.g. trityl, benzyl or the like).

Among these, urethane-type protecting groups are especially preferable.

Carboxyl groups are protected by converting them into, for example, amido group, hydrazido group or ester. Preferable amido groups or hydrazido groups are those substituted with a suitable substituent. Preferable amido groups are those substituted with a C7-19 aralkyl group optionally substituted with, for example, alkoxy group (e.g. 3,4-dimethoxybenzyl group or bis-(p-methoxyphenyl)-methyl group). Preferable hydrazido groups are those substituted with, for example, C1-6 alkyloxycarbonyl group optionally substituted with halogen (e.g. fluorine, chlorine, bromine, etc.) C<sub>6-12</sub> aryl group (e.g. phenyl, p-biphenylyl, etc.) (e.g. benzyloxycarbonyl group, trichloroethyloxycarbonyl group, tert-butyloxycarbonyl group, 2-(pbiphenylyl)-isopropyloxycarbonyl, etc.), halogenated  $C_{2-6}$  alkanoyl group (e.g. trifluoroacetyl group, etc.) and  $C_{7-19}$  aralkyl group (e.g. trityl group, etc.), among others. Further, the carboxyl groups may be 10 protected as esters with an optionally substituted lower alcohol (e.g. methanol, ethanol, cyanomethyl alcohol, benzoylmethyl alcohol, tert-butanol, etc.), aralkanol [e.g. benzyl alcohol or benzhydrols (e.g. benzhydrol, p-nitrobenzyl alcohol, p-methoxybenzyl alcohol, 2,4,6-trimethylbenzyl alcohol, etc.) optionally substituted with, for example, lower alkyl group, lower alkoxy group or or halogen atom], phenol and thiophenol optionally substituted with an electron withdrawing group (e.g. thiophenol, thiocresol, pnitrothiophenol, 2,4,5- and 2,4,6-trichlorophenol, p-cyanophenol, p-methanesulfonyl phenol, etc.), and further, with N-hydroxyimide (e.g. N-hydroxysuccinimide, N-hydroxyphthalimide, etc.), N-hydroxypiperidine, 8hydroxyquinoline or the like.

As the special protecting group of carboxyl, which can be removed under neutral reaction conditions, mention is made of a hydrocarbyl-silyl-ethyl group, for example, 2-(trimethylsilyl)-ethyl group (described in German Patent Application Laid-Open No.2,706,490).

Hydroxyl can be protected with, for example, acylation or etherification.

The especially preferable acyl group in the case of acylation is a group derived from carbonic acid (e.g. benzyloxycarbonyl group or ethyloxycarbonyl group). For etherfication, benzyl group, tetrahydropyranyl group or tert-butyl group, for example, is preferable. And, for the protection of hydroxyl group, 2,2,2-trifluoro-1-tert-butoxycarbonylamino ethyl group or 2,2,2-trifluoro-1-benzyloxycarbonylamino ethyl group [Chem. Ber., Vol. 100 (1967), pp 3838-3849] is preferably employed.

In the formula (I), in the case where the amino acid residue is possibly an optically active isomer, it can take any of L-, D- and DL-form.

The prefered examples of compound (I) or a salt thereof include

(2R,6R)-2-amino-6,7-bis(hexadecanoyloxy)-4-thiaheptanoyl-Gly-Glu-Glu-OH,

(2R,6R)-2-amino-6,7-bis(hexadecanoyloxy)-4-thiaheptanoyl-Gly-Gly-Glu-OH,

(2P,6R)-2-amino-6,7-bis(hexadecanoyloxy)-4-thiaheptanoyl-Glu-Gly-Glu-OH,

(2R,6R)-2-amino-6,7-bis(hexadecanoyloxy)-4-thiaheptanoyl-Gly-D-Glu-OH, or salts thereof.

In the following, description is made on the method of producing the above-mentioned compounds.

Protective groups and reagents frequently used are abbreviated as follows in the subsequent description.

Fmoc: 9-fluorenyl methyloxycarbonyl

Z: benzyloxy carbonyl

<sup>1</sup>Bu: t-butyl

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O<sup>t</sup>Bu : t-butoxyester
TFA: trifluoroacetic acid
TEA: triethylamine

DMF: N,N-dimethylformamide

DCC: N,N'-dicyclohexylcarbodiimide

BOP: benzotriazol-1-yloxy tris (dimethylamino) phosphonium • hexafluorophosphate

DIC: N,N'-diisopropylcarbodiimide

HONB: N-hydroxy-5-norbonen-2,3-dicarboxyimide

DEPC: diethyl phosphorocyanidate HOBT: 1-hydroxybenzotriazole

DCM: dichloromethane

MeOH: methanol THF: tetrahydrofuran

WSC: water-soluble carbodiimide • hydrochloride

DMAP: 4-dimethylaminopyridine

Boc: t-butoxycarbonyl
R-: R-configuration
S-: S-configuration

The compound, which is the basic skeleton of the compounds in this specification is 2-amino-6,7-

dihydroxy-4-thiaheptanoic acid.

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In this specification, thisheptanoyl is abbreviated as THT, thisheptanoic acid is abbreviated as THT-OH. And, octadecanoyl is abbreviated as Ste, hexadecanoyl as Pam, tetradecanoyl as Myr, and octadecanoyloxy as SteO, hexadecanoyloxy as PamO tetradecanoyloxy as MyrO.

Compounds represented by the formula (I) or salts thereof can be produced by subjecting a compound represented by the formula (II) or a salt thereof to deprotection reaction.

The deprotection reaction can be conducted by a <u>per se</u> known method, for example, a method conventionally employed in the field of peptide chemistry. [cf. Synthetic Chemistry Series, Peptide Synthesis, authors: IZUMIYA Nobuo, UNO Motonori, KATO Tetsuo & AOYAGI Haruhiko; Published by Maruzen Co. Ltd. 1975 in Japan].

Deprotection reaction on the amino group protected with a urethane-type protecting group is conducted by bringing the amino group into contact with an acid in the absence of solvent or in a solvent which gives no adverse influence on the reaction. As the solvent, use is made of halogenated hydrocarbons (e.g. dichloromethane, chloroform, 1,2-dichloroethane, etc.), alcohols (e.g. methanol, ethanol, etc.), water and a mixture of them in an appropriate ratio. This reaction is conducted by bringing the compound (II) or a salt thereof into contact with an acid. As the acid, use is made of, for example, haloacetic acid (e.g. trifluoroacetic acid, etc.), hydrohalogenic acid (e.g. hydrochloric acid, hydrobromic acid, etc.), among others. It is advantageous that N-benzyloxycarbonyl group and N-4-methoxybenzyloxycarbonyl group are removed by catalytic hydrogenation by using, for example, palladium catalyst (e.g. palladium carbon, palladium-barium sulfate, palladium black, etc.) or rhodium catalyst. In this case, a known solvent, for example, cyclic ether (e.g. tetrahydrofuran, etc.), alcohols (e.g. methanol, etc.) etc., or, depending on cases, a mixture of cyclic ether and other inert solvents [e.g. lower aliphatic acid amide (e.g. dimethylformamide, etc.) etc.] is used.

N-9-Fluorenylmethyloxycarbonyl group is removed advantageously by using an organic amine, for example, diethylamine, piperidine, morpholine, 4-dimethylaminopyridine or dicyclohexylamine. The reaction is conducted in a solvent which gives no adverse reaction on the reaction. As the solvent, use is made of, for example, amides (e.g. dimethylformamide, acetamide, etc.), alcohols (e.g. methanol, ethanol, etc.), halogenated hydrocarbons (e.g. chloroform, dichloromethane, 1,2-dichloroethane, etc.), etc., or a mixture of them in an appropriate ratio.

It is advantageous that, N-2,2,2-trichloroethyloxycarbonyl group is removed by using a metal (e.g. zinc, etc.) together with an organic carboxylic acid (e.g. acetic acid, propionic acid, etc.). The reaction is conducted in a solvent which gives no adverse influence on the reaction. As the solvent, use is made of the above-mentioned carboxylic acid, alcohols (e.g. methanol, ethanol, etc.) water or a mixture of them in an appropriate ratio.

Deprotection reaction (deacylation reaction) of acylated hydroxy group is conducted by bringing it into contact with an acid in a solvent which gives no adverse influence. As the solvent, use is made of halogenated hydrocarbons (e.g. dichloromethane, chloroform, 1,2-dichloroethane, etc.), alcohols (e.g. methanol, etc.), water and a mixture of them in an appropriate ratio. This reaction is conducted by bringing the compound (II) or a salt thereof into contact with an acid. As the acid, use is made of, for example, haloacetic acid (e.g. trifluoroacetic acid, etc.), hydrohalogenic acid (e.g. hydrochloric acid, hydrobromic acid, etc.), etc.

Elimination of O-benzyl group is performed advantageously by catalytic hydrogenation with, for example, a palladium catalyst such as palladium carbon, palladium/barium sulfate and palladium black, or a rhodium catalyst, using, in this case, a known solvent, for example, cyclic ether (e.g. tetrahydrofuran, etc.) as a mixture with other inactive solvents [e.g. lower aliphatic acid amide (dimethylformamide or the like) etc.].

Elimination of O-tetrahydropyranyl group or O-tert-butyl group can be performed, like in the above-mentioned deacylation, by hydrolysis with an acid.

Elimination of a carboxyl-protecting group can be performed, like in the above-mentioned case, by hydrolysis with an acid. And, benzyl ester, for example, can be deprotected by catalytic hydrogenation like in the case of the above-mentioned deprotection of the O-benzyl group elimination. The above-mentioned 2-(trimethylsilyl)-ethyl group can be eliminated by the action of, for example, a salt of hydrofluoric acid, for example, especially a salt of quaternary nitrogen base with hydrofluoric acid (e.g. tetraethyl ammonium fluoride, etc.) in a suitable solvent under neutral conditions.

The compound represented by the general formula (II) or a salt thereof, which is the starting compound used for the production of the afore-mentioned general formula (I) or a salt thereof, can be produced by subjecting a material having a reactive carboxyl group corresponding to one of the two fragments separated at an optional position of the peptide linkage and a material having a reactive amino group corresponding to the other fragment to condensation by a conventional means employed in peptide synthesis.

As the conventional means of peptide synthesis, mention is made of, for example, anyone of liquid-phase synthetic method and solid-phase synthetic method. Such means of peptide synthesis as mentioned above include, for example, methods described in "Peptide Synthesis" written by M. Bondosky and M. Cridetti, Interscience, New York, 1966; "The Proteins" written by F.M. Finn and K. Hoffmann, Vol.2, edited by H. Nenrath and R.L. Hill, Academic Press Inc. New York, 1976; "Base & Experiment of Peptide Synthesis" written by IZUMIYA Nobuo, Maruzen Co. Ltd., 1985; "Seikagaku Jikken Koza 1" written by YAJIMA Haruaki, SAKAKIBARA Shunpei, et al. compiled by Japan Biochemistry Society, Tokyo Kagaku Dojin, 1977, "Zoku Seikagaku Jikken Koza 2" written by KIMURA Shun et al. compiled by Japan Biochemistry Society, Tokyo Kagaku Dojin, 1987; "Solid Phase Peptide Synthesis" written by J.M. Slewart and J.D. Young, Pizs chemical company, Illinois, 1984, or methods analogous to them. Practical examples of such methods as above includes, for example, azide method, chloride method, acid anhydride method, mixed acid anhydrides method, DCC method, active ester method, method using Woodword reagent K, carbonyl imidazole method, redoc method, DCC/HONB method, DIC/HONB method, DCC/HONB method, method using BOP reagent, method using DEPC reagent or the like.

As preferable practical examples of one of the starting fragments, in the case of producing the above-mentioned compound (II) or a salt thereof, mention is made of, for example, a compound represented by the general formula (III):

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wherein Y' and R3 are of the same meaning as defined above or salts thereof.

The compound (III) or salts thereof can be produced by, far example, the above-mentioned conventional means for peptide synthesis.

Preferable examples of the compound (III) include the following compounds.

		O <sup>t</sup> Bu
	P-2	H-Gly-Gly-Gly-Glu(O <sup>t</sup> Bu)-Thr( <sup>t</sup> Bu)-O <sup>t</sup> Bu
5	P-3	H-Glu(O <sup>t</sup> Bu)-Gly-Glu(O <sup>t</sup> Bu)-Gly-D-Glu(O <sup>t</sup> Bu)-
		O <sup>t</sup> Bu
	P-4	H-Gly-Gly-Glu(O <sup>t</sup> Bu)-O <sup>t</sup> Bu
10	P-5	H-Gly-Gly-D-Glu(O <sup>t</sup> Bu)-O <sup>t</sup> Bu
	P-6	H-Glu(O <sup>t</sup> Bu)-Gly-D-Glu(O <sup>t</sup> Bu)-O <sup>t</sup> Bu
	P-7	H-Gly-Gly-Glu(O <sup>t</sup> Bu)-O <sup>t</sup> Bu
15	P-8	H-Gly-Gly-Asp(O <sup>t</sup> Bu)-O <sup>t</sup> Bu
•	P-9	H-Gly-Gly-D-Glu(O <sup>t</sup> Bu)-O <sup>t</sup> Bu
	P-10	H-Gly-D-Glu(O <sup>t</sup> Bu)-O <sup>t</sup> Bu
20	P-11	H-Gly-Glu(O <sup>t</sup> Bu)-Glu(O <sup>t</sup> Bu)-O <sup>t</sup> Bu
	P-12	H-Gly-Glu(O <sup>t</sup> Bu)-D-Glu(O <sup>t</sup> Bu)-O <sup>t</sup> Bu
	P-13	H-Glu(O <sup>t</sup> Bu)-Gly-Glu(O <sup>t</sup> Bu)-O <sup>t</sup> Bu
25	P-14	H-Glu(O <sup>t</sup> Bu)-D-Glu(O <sup>t</sup> Bu)-O <sup>t</sup> Bu
25	P-15	H-Glu(O <sup>t</sup> Bu)-Glu(O <sup>t</sup> Bu)-Glu(O <sup>t</sup> Bu)-O <sup>t</sup> Bu
	P-16	H-Glu(O <sup>t</sup> Bu)-Glu(O <sup>t</sup> Bu)-D-Glu(O <sup>t</sup> Bu)-O <sup>t</sup> Bu
	P-17	$NH_2(CH_2)_7CO-Glu(O^tBu)-O^tBu$
30	P-18	$NH_2(CH_2)_{11}CO-Glu(O^tBu)-O^tBu$
	P-19	4-aminobenzoyl-Glu(O <sup>t</sup> Bu)-O <sup>t</sup> Bu
	P-20	4-(glycylamino)benzoyl-Glu(O <sup>t</sup> Bu)-O <sup>t</sup> Bu
35	P-21	$NH_2(CH_2)_5CO-Glu(O^tBu)-O^tBu$
	P-22	$NH_2(CH_2)_6NHCO-Glu(O^tBu)-O^tBu$
	P-23	4-(aminomethyl)benzoyl-Glu(O <sup>t</sup> Bu)-O <sup>t</sup> Bu
40		hydrochloride
	P-24	4-(N-(t-
		butyloxycarbonylmethyl)aminomethyl)benzoyl-
45		Glu(O <sup>t</sup> Bu)-O <sup>t</sup> Bu
	P-25	H-Gly-Lys(Boc)-Gly-O <sup>t</sup> Bu

As the starting fragment for producing the compound (II) by combination with the above-mentioned compound (III), use is made of a compound represented by the formula (IV):

$$\begin{array}{c} \text{O-R}^2\\ & \downarrow\\ \text{CH}_2\text{-S-CH}_2\text{-CH-CH}_2\text{-O-R}^1\\ & \downarrow\\ \text{Y-HN-CH-COOH} \end{array}$$

wherein Y stands for an amino-protecting group; R1 and R2 are of the same meaning as defined above.

The amino-protecting group represented by Y in the above-mentioned general formula (IV) has the same meaning as defined for the afore-described amino-protecting group.

The compound (IV) or a salt thereof, wherein R1 and R2 stand for one and the same group, can be produced be adequately applying a per se known method [e.g. International Journal of Peptide and Protein Research, 38 1991 pp.545-554; Chemical and Pharmaceutical Bulletin, 39 pp.2590-2596].

In the case where R1 and R2 are different groups from each other, most common production route is described below.

First, the monoacyl compound of a glycerin derivative, for example, glycidol or epichlorohydrin, is prepared. On the other hand, the disulfide linkage of cystine in which amino group and carboxyl group are protected is opened reductively to give a cysteine derivative.

By subjecting this cysteine derivative to addition reaction to the above-mentioned monoacyl glycerine derivative, 2-amino-6-hydroxy-7-acyloxy-4-thiaheptanoic acid in which the amino group and carboxyl group

In this reaction, use of R-(+)-glycidol gives a (6R)-2-amino-6-hydroxy-7-acyloxy-4-thiaheptanoic acid are protected. derivative, while use of S-(-)-glycidol gives a (6S)-2-amino-6-hydroxy-7-acyloxy-4-thiaheptanoic acid deriva-

By conventional acylation of the hydroxyl group at 6-position of 2-amino-6-hydroxy-7-acyloxy-4thiaheptanoic acid obtained by the above-mentioned reaction, a 2-amino-6,7-bis(acyloxy)-4-thiaheptanoic acid derivative having a different O-acyl group can be obtained.

By directly using, for example, glycidol instead of an acyl glycerine derivative, 2-amino-6,7-dihydroxy-4thiaheptanoic acid derivative is obtained, and, acylation of this product by a conventional method conveniently gives 2-amino-6,7-bis(acyloxy)-4-thiaheptanoic acid derivative having the same O-acyl group.

By removing, in the above-mentioned manner, the protecting group of the carboxyl group of the 2amino-6,7-bis(acyloxy)-4-thiaheptanoic acid derivative thus obtained, 2-amino-6,7-bis(acyloxy)-4-thiaheptanoic acid having protected amino group can be prepared.

Preferable examples of the compound (IV) include 2-amino-6,7-bis(acyloxy)-4-thiaheptanoic acid having protected amino group.

Further examples are shown below:

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Compound No.	Structural Formula
GC-1	(2R,6S)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-OH
GC-2	(2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-OH
GC-3	(2S,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-OH
GC-4	(2S,6S)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-OH
GC-5	(2R,6R)-2-Fmoc-amino-6-hexanoyloxy-7-PamO-4-THT-OH
GC-6	(2R,6R)-2-Fmoc-amino-6,7-bis(SteO)-4-THT-OH
GC-7	(2R,6R)-2-Fmoc-amino-6,7-bis(MyrO)-4-THT-OH

The starting fragments obtained thus above are subjected to condensation in the following manner, and then, if necessary, the amino-protecting group shown by Y is removed to give the compound (II) or a salt

As the compound produced by activating the carboxyl group of the starting compound, mention is made of, for example, corresponding acid anhydrides, azides, active esters [e.g. esters with alcohol (e.g. pentachlorophenol, 2,4,5-trichlorophenol, 2,4-dinitrophenol, cyanomethyl alcohol, p-nitrophenol, N-hydroxy-5-norbornene-2,3-dicarboxyimde, N-hydroxysuccinimide, N-hydroxyphalimide, N-hydroxybenzotriazole)]. As the starting compound whose amino group is activated, mention is made of, for example, the corresponding phosphoric acid amide.

The condensation can be conducted in the presence of a solvent. The solvent can be selected from those known as being employable for peptide condensation reaction. Examples of the solvent include amides such as anhydrous or water-containing formamide, dimethylformamide, N-methyl pyrrolidone, etc., sulfoxides such as dimethyl sulfoxide, etc., aromatic amines such as pyridine, etc., halogenated hydrocarbons such as chloroform, dichloromethane, etc., ethers such as tetrahydrofuran, dioxane, etc., nitriles such as acetonitrile, etc., esters such as ethyl acetate, ethyl formate, etc., or a mixture of them in an appropriate

The reaction temperatures can be adequately selected from the range known as being employable for peptide linkage forming reaction, specifically, for example, usually from about -20 °C to 40 °C.

The reaction time can be adequately selected from the range known as being required for peptide linkage formation reaction, specifically, for example, from about several minutes to about 48 hours.

Removal of the amino-protecting group shown by Y is conducted in substantially the same manner as described above.

The compound (I) or a salt thereof thus produced can be recovered, after completion of the reaction, by means for separating peptide, for example, extraction, partition, reprecipitation, crystallization, various chromatographic processes, high performance liquid chromatography, etc.

A salt of the compound (I) of this invention with a base, especially with a pharmaceutically acceptable base, can be obtained by a per se known method. Examples of the base include an alkali metal such as sodium, potassium, etc., an alkaline earth metal such as calcium, magnesium, etc., an organic base such as triethylamine, pyridine, etc. and so on. A salt of the compound (I) with an acid, especially a pharmaceutically acceptable acid can be obtained by a per se known method. Examples of the acid include an inorganic acid (e.g. hydrochloric acid, sulfuric acid, phosphoric acid, etc.), an organic acid (e.g. acetic acid, propionic acid, citric acid, tartaric acid, malic acid, oxalic acid, etc.) and so on.

And, as salts of the compounds (II) to (IV), use is made of those substantially the same as salts of the compound (I).

The compound (I) or salts thereof have activities of remarkably improving the state of hematopoiesis-insufficiency, which can be used as therapeutic or prophylactic agents of leukocytopenia caused by radiotherapy or chemotherapy of cancers in mammals (e.g. dog, pig, cow, horse, monkey, man, etc), as hematopoietic stimulating agents in the case of bone marrow transplantation, as immuno-stimulating agents having leucocyte-increasing action, and, further, as a therapeutic agent of thrombocytopenia.

The compound (I) or salts thereof are low in toxicity and can be used safely.

In the case of administering the compound (I) or a salt thereof to, for example, man, it can be safely administered orally or non-orally alone or as a pharmaceutical composition by mixing with a suitable pharmaceutically acceptable carrier, excipient or diluent.

Examples of the above-mentioned pharmaceutical composition include injections, orally administrable compositions (e.g. powder, granules, capsules, tablets), topica (e.g. transnasal agent, transdermal agent, etc.), suppositories (e.g. rectal suppositories, vaginal suppositories).

These pharmaceutical compositions can be prepared by <u>per se</u> known methods generally employed in the processes of pharmaceutical preparation.

For example, the compound (I) or a salt thereof of this invention can be prepared into aqueous injections together with a dispersant (e.g. Tween 80 (manufactured by Atlas Powder Co., U.S.A.), HCO 60 (manufactured by Nikko Chemicals), polyethylene glycol, carboxymethyl cellulose, sodium alginate or the like), a preservative (e.g. methylparaben, propylparaben, benzyl alcohol, chlorobutanol or the like) and an isotonicating agent (e.g. sodium chloride, glycerin, sorbitol, glucose or the like), among others, or into oleagenous injections by dissolving, suspending or emulsifying in a vegetable oil such as olive oil, sesame oil, peanut oil, cotton seed oil, corn oil or the like, propylene glycol, among others.

For preparing the compound (I) or a salt thereof of this invention into compositions for oral administration, it is subjected to compression molding, in accordance with a per se known method, together with an excipient (e.g. lactose, sucrose, starch or the like), a disintegrator (e.g. starch, calcium carbonate or the like), a binding agent (e.g. starch, gum arabica, carboxymethyl cellulose, polyvinyl pyrrolidone, hydroxypropyl cellulose or the like) or a lubricant (e.g. talc, magnesium stearate, polyethylene glycol 6000), etc., followed by, upon necessity, masking the taste or coating by a per se known process for the purpose of enteric coating or of making the compositions to be of sustained-release. Examples of the coating agent include hydroxypropyl methyl cellulose, ethyl cellulose, hydroxymethyl cellulose, hdyroxypropyl cellulose, polyoxyethylene glycol, Tween 80, Brulonick F68, cellulose acetate phthalate, hdyroxypropyl methyl cellulose phthalate, hydroxymethyl cellulose acetate succinate, Eudragit (manufactured by Rohm, Germany, methacrylic acid acrylic acid copolymer) and a pigment such as titanium oxide, red iron oxide, etc. Subcoating layer may be provided between the enteric coating and the core according to per se known method.

For preparing the compound (I) or a salt thereof into, for example, solid, semi-solid or liquid compositions for topical use, a per se known method can be resorted to. For preparing solid compositions, for example, the compound (I) or a salt thereof is prepared into powdery compositions singly or in admixture with an excipient (e.g. glycol, mannitol, starch, microcrystalline cellulose or the like), a thickening agent (e.g. natural rubbers, cellulose derivatives, acrylic acid polymers or the like). As the above-mentioned liquid composition, almost like in the case of injections, mention is made of oleaginous or aqueous suspensions. As the semi-solid composition, aqueous or oleagenous gel compositions or ointments are preferably counted. These compositions may be supplemented with a pH regulator (e.g. carbonic acid,

phosphoric acid, citric acid, hydrochloric acid, sodium hydroxide, etc.), a preservative (e.g. para-hydroxybenzoic acid esters, chlorobutanol, benzalkonium chloride, etc.), among others.

In the case of preparing suppositories for example, the compound (I) or a salt thereof of the present invention can be prepared into oleagenous or water-soluble solid, semi-solid or liquid suppositories. As the oleagenous base for the above-mentioned compositions, any one which does not dissolve the compound (I) or a salt thereof can be employed, as exemplified by glyceride of higher fatty acid [e.g. cacao butter, Wittepsols (manufactured by Dynamite Nobel, Inc.), etc.], middle class fatty acid [e.g. Migliols (manufactured by Dynamite Nobel, Inc.), etc.] or vegetable oil (e.g. sesame oil, soybean oil, cotton seed oil, etc.), among others. And, examples of the water-soluble base include polyethylene glycols and propylene glycols, and, examples of the water-soluble gel base include natural rubbers, cellulose derivatives, vinyl polymers, acrylic acid polymers, etc.

While the dosage of the compound (I) or a salt thereof when administered to man as injections varies with diseases, administration routes, ages of individual patients and states of disease, it ranges from about  $0.01\mu g$  to 20 mg in terms of the effective component, preferably from about 0.1  $\mu g$  to 0.2 mg, more preferably from about 0.5  $\mu g$  to 50  $\mu g$  in the case of a common adult patient (50 kg body weight).

### [Examples]

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By the following reference examples, working examples, experimental examples and formulation examples, the present invention will be described in further detail, but they are not intended to limit the invention in any manner. The numerals showing the mixing ratio in mixed solvents mean the volume ratio of each solvent. Percent (%) means w/w%, unless specified otherwise.

 $^{1}$ H-NMR spectrum is determined by a Varian Gemini200 (200 MHz) type spectrometer using tetramethyl silane as the internal standard, expressing all the  $\delta$  values as ppm.

Symbols used in the reference examples and working examples are of the following meaning. s; singlet, d: doublet, d: triplet, q: quartet, dd: double doublet, dt: double triplet, m: multiplet, br: broad.

### Reference Example 1

Production of H-Gly-Gly-Gly-Glu(O¹Bu)-Thr(¹Bu)-Thr(¹Bu)-O¹Bu (P-1)

H-Gly-Gly-Gly-OH (10.0 g, 52.9 mmol, prepared by Peptide Research Labs) was dissolved in a 4N aqueous solution of sodium hydroxiede (13.3 ml). To the solution were added, under ice-cooling, a 4N aqueous solution of sodium hydroxide (15.9 ml) and benzyloxycarbonylchloride (9.31 ml), then the mixture was stirred overnight at 20 °C. The reaction mixture was washed with ether. To the aqueous layer was added, under ice-cooling, 5M HCl to adjust its pH to 3, which was left standing overnight at a cool place. Resulting crystalline precipitates were collected by filtration, washed with cold water and dried. The crystals thus obtained were used without purification. The yield was 13.4 g (78.5 %).

Z-Gly-Gly-Gly-OH (3.04 g, 9.41 mmol) obtained thus above was dissolved in DMF (200 ml). To the solution were added, under ice-cooling, HONB (1.86 g, 10.4 mmol) and DCC (2.14 g, 10.4 mmol). The mixture was stirred for two hours under ice-cooling, then insolubles were filtered off.

Z-Glu(O¹Bu)-Thr(¹Bu)-Thr(¹Bu)-O¹Bu (6.66 g, 9.41 mmol) was dissolved in methanol (300 ml). To the solution was added 10 % (w/w, hereinafter in the same way) palladium-carbon. The mixture was stirred for two hours in hydrogen streams at ordinary temperature under normal pressure. The catalyst was filtered off, and the solvent was distilled off. The residue was dissolved in DMF (150 ml). To the solution was added, under ice-cooling, diisopropylethylamine (1.80 ml, 10.4 mmol). The mixture was stirred, to which was added the solution prepared as above, and the mixture was stirred overnight at 20 °C, then the solvent was distilled off. To the residue were added chloroform and water, which was subjected to extraction with chloroform. The chloroform layer was washed with a 10 % (w/v) aqueous solution of citric acid, water, a saturated aqueous solution of sodium hydrogencarbonate and water, successively, which was dried over anhydrous sodium sulfate. Then, the solvent was distilled off. The residue was purified by means of a silicagel column chromatography [5 % (v/v) methanol-chloroform], followed by recrystallization from ethyl acetate-acetonitrile to afford Z-Gly-Gly-Gly-Glu(O¹Bu)-Thr(¹Bu)-Thr(¹Bu)-O¹Bu as crystals. The yield was 5.74 g (75.4 %)

m.p. 167.5-168.0 °C  $[\alpha]_D^{25} + 7.28$  ° (c = 1.03 in DMF)

Elemental Analysis for C <sub>43</sub> H <sub>70</sub> N <sub>6</sub> O <sub>13</sub> :					
Calcd.	C, 58.75;	H, 8.03;	N, 9.56		
Found	C, 58.52;	H, 7.78;	N, 9.35		

Amino acid analysis [6N HCl, 110 ° C, hydrolysis for 24 hours; values in parentheses show theoretical ones]: Glu 1.00(1); Thr 1.81(2); Gly 2.84(3) Z-Gly-Gly-Gly-Glu(O¹Bu)-Thr(¹Bu)-Thr(¹Bu)-O¹Bu obtained above (1.97 g, 2.24 mmol) was dissolved in methanol (60 ml). To the solution was added 10% palladium-carbon (120 mg). The mixture was stirred for two hours in hydrogen streams at ordinary temperature under normal pressure. The catalyst was removed, then the solvent was distilled off to leave P-1 as a solid product. The yield was 1.64 g (98 %).

FAB-MS (M + H) = 879 (theoretical value = 879)

Reference Example 2

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Production of H-Gly-Gly-Gly-Glu(O<sup>t</sup>Bu)-Thr(<sup>t</sup>Bu)-O<sup>t</sup>Bu (P-2)

a) H-Glu(O¹Bu)-Thr(¹Bu)-O¹Bu (1.27 g, 3.05 mmol) and Z-Gly-Gly-OH (0.99 g, 3.05 mmol) were dissolved in DMF (20 ml). To the solution were added HOBT (453 mg, 3.35 mmol) and WSC (643 mg, 3.35 mmol). The mixture was stirred for 8 hours at 20 °C. The reaction mixture was concentrated, which was suspended in 0.2 M aqueous solution of citric acid (70 ml), followed by extraction with ethyl acetate (100 ml, 70 ml). The ethyl acetate layers were combined and washed with a 10 % (w/v) aqueous solution of ammonium chloride, 5% (w/v) aqueous solution of sodium hydrogencarbonate, water and a saturated aqueous saline solution, successively, which was dried over anhydrous sodium sulfate, followed by concentration. To the concentrate was added ether-hexane. Resulting precipitates were collected by filtration to afford Z-Gly-Gly-Gly-Glu(O¹Bu)-Thr(¹Bu)-O¹Bu (Z-P-2) as a while powdery product (1.87 g, 85 %).

 $[\alpha]_{D}^{23} + 6.9 \cdot (c = 0.55 \text{ in chloroform})$ 

Elemental Analysis for C <sub>35</sub> H <sub>55</sub> N <sub>5</sub> O <sub>11</sub> :					
Calcd.	C, 58.24;	H, 7.68;	N, 9.70		
Found	C, 57.87;	H, 7.64;	N, 9.97		

b) This powdery product (1.09 g) was dissolved in methanol (36 ml), to which was added 10 % palladium-carbon (109 ml). The mixture was stirred for two hours in hydrogen streams at ordinary temperature under normal pressure. The catalyst was removed, and the solvent was distilled off to leave P-2 as a powdery product (861 mg).

 $[\alpha]_0^{23} + 9.1 \circ (c = 0.53 \text{ in chloroform})$ 

Elemental Analysis for C <sub>27</sub> H <sub>49</sub> N <sub>5</sub> O <sub>9</sub> :				
Calcd.	C, 55.18;	Н, 8.40;	N, 11.92	
Found	C, 54.88;	Н, 8.57;	N, 11.68	

Reference Example 3

Production of H-Glu(O¹Bu)-Gly-Glu(O¹Bu)-Gly-D-Glu(O¹Bu)-O¹Bu (P-3)

In substantially the same manner as in Reference Example 2, H-Glu(O¹Bu)-Gly-Glu(O¹Bu)-Gly-D-Glu-(O¹Bu)-O¹Bu (P-3) was produced

Compound Z-P-3:  $[\alpha]_D^{20}$  + 5.0 ° (c = 0.54, in chloroform)

Elemental Analysis for C<sub>43</sub>H<sub>67</sub>N<sub>5</sub>O<sub>14</sub>:

Calcd. C, 58.82; H, 7.69; N, 7.98

Found C, 58.67; H, 7.67; N, 8.26

Compound P-3:  $[\alpha]_D^{20}$  - 6.1 (C = 0.51, in chloroform)

Elemental Analysis for C <sub>35</sub> H <sub>61</sub> N <sub>5</sub> O <sub>12</sub> :				
Calcd.	C, 56.51;	H, 8.27;	N, 9.42	
Found	C, 56.66;	H, 8.30;	N, 9.63	

Reference Example 4

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Production of H-Gly-Gly-Glu(OtBu)-OtBu (P-4)

In substantially the same manner as in Reference Example 2, H-Gly-Gly-Gly-Glu(O¹Bu)-O¹Bu (P-4) was produced.

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40	÷	Materials	Reaction Co	onditions	Products
	a)	1) Z-Gly-Gly-Gly-OH	wsc	548 mg	Z-P-4
	·	(0.84 g)	HOBT	386 mg	(0.81 g)
45			DMF	15 ml	
		2) H-Glu(O <sup>t</sup> Bu)-O <sup>t</sup> Bu	20°C	18 h	
		(0.68 g)			
50	b)	Z-P-4	10% Pd-C	73 mg	P-4
		(735 mg)	MeOH	25 ml	(543 mg)
			20°C	2 h	

Compound Z-P-4:  $[\alpha]_D^{23} + 9.6 \, (c = 0.56)$ , in chloroform)

Elemental Analysis for C <sub>27</sub> H <sub>40</sub> N <sub>4</sub> O <sub>9</sub> :				
Calcd.	C, 57.43;	H, 7.14;	N, 9.92	
Found	C, 57.60;	H, 7.12;	N, 9.88	

Compound P-4:  $[\alpha]_{D}^{23} + 13.1 \cdot (c = 0.51, in chloroform)$ 

Elemental Analysis for C <sub>13</sub> H <sub>34</sub> N <sub>4</sub> O <sub>7</sub> • 0.5H <sub>2</sub> O:				
Calcd.	C, 51.92;	H, 8.03;	N, 12.75	
Found	C, 52.09;	H, 7.89;	N, 12.56	

# 15 Reference Example 5

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Production of H-Gly-Gly-Gly-D-Glu(O¹Bu)-O¹Bu (P-5)

In substantially the same manner as in Reference Example 2, H-Gly-Gly-Gly-D-Glu(O<sup>t</sup>Bu)-O<sup>t</sup>Bu (P-5) was produced.

Materials	Reaction Conditions	Products
a) 1) Z-Gly-Gly-Gly-OH	WSC 1.06 g	Z-P-5
(1.62 g)	HOBT 744 mg	(1.93 g)
	DMF 25 ml	
2) H-D-Glu(O <sup>t</sup> Bu)-O <sup>t</sup> Bu	20°C 16 h	

Compound Z-P-5:  $[\alpha]_D^{23}$  - 9.1 ° (c = 0.54, in chloroform)

(1.30 g)

Elemental Analysis for C <sub>27</sub> H <sub>40</sub> N <sub>4</sub> O <sub>9</sub> :				
Calcd.	C, 57.43;	H, 7.14;	N, 9.92	
Found	C, 57.38;	H, 7.06;	N, 10.02	

Compound P-5:  $[\alpha]_D^{23}$  - 13.5 • (c = 0.54, in chloroform)

Elemental Analysis for C <sub>19</sub> H <sub>34</sub> N <sub>4</sub> O <sub>7</sub> • 0.5H <sub>2</sub> O				
Calcd.	C, 51.92;	H, 8.03;	N, 12.75	
Found	C, 51.98;	H, 7.93;	N, 12.68	

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# Reference Example 6

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Production of H-Glu(O<sup>t</sup>Bu)-Gly-D-Glu(O<sup>t</sup>Bu)-O<sup>t</sup>Bu (P-6)

In substantially the same manner as in Reference Example 2, H-Glu(O¹Bu)-Gly-D-Glu(O¹Bu)-O¹Bu (P-6) was produced.

		Materials	Reaction Conditions	Products
10	a )	l) Fmoc-Glu(O <sup>t</sup> Bu)-OH	WSC 1.24 g	Fmoc-P-6
		(2.78 g)	HOBT 876 mg	(3.73 g)
			DCM 85 ml	
15		2) H-Gly-D-Glu(O <sup>t</sup> Bu)-O <sup>t</sup>	Bu 20°C 16 h	
		(1.86 g)		

 $[\alpha]_D^{23}$  - 12 • (c = 0.54, in chloroform)

Elemental Analysis for C <sub>39</sub> H <sub>53</sub> N <sub>3</sub> O <sub>10</sub> :					
Calcd.	C, 64.71;	H, 7.38;	N, 5.80		
Found	C, 64.55;	H, 7.43;	N, 6.09		

c) This powdery product (2.10 g) was dissolved in dichloromethane (63 ml), to which was added piperidine (7.0 ml), and the mixture was stirred for one hour at room temperatures. To the reaction mixture was added ethyl acetate (300 ml), which was subjected to extraction with 1N HCl (70 ml) and 0.03N HCl (100 ml x 2). The aqueous layer was adjusted to pH 6.0, which was washed with a mixture of hexane:ether (=1:1) (100 ml x 5). Then, the pH was adjusted to 8.6, followed by extraction with ethyl acetate (150 ml x 2). The extract was dried over anhydrous sodium sulfate, followed by concentration to afford P-6 as a colorless oily product (1.31 g).  $[\alpha l_0^{23} - 17^{\circ} (c = 0.47, in chloroform)$ 

Elemental Analysis for C <sub>24</sub> H <sub>43</sub> N <sub>3</sub> O <sub>8</sub> :				
Calcd.	C, 57.47;	H, 8.64;	N, 8.38	
Found	C, 57.40;	H, 8.73;	N, 8.60	

Reference Example 7

Production of H-Gly-Gly-Glu(O'Bu)-O'Bu (P-7)

In substantially the same manner as in Reference Example 2, H-Gly-Gly-Glu(O¹Bu)-O¹Bu (P-7) was produced.

Products Materials Reaction Conditions 1) Z-Gly-Gly-OH 1.63 g Z-P-7 WSC a) (2.05 g)HOBT 1.15 g (3.34 g)5 DMF 40 ml 2) H-Glu(O<sup>t</sup>Bu)-O<sup>t</sup>Bu 20°C 17 h (2.00 g)b) Z-P-7 10% Pd-C 200 mg P-7 (colorless oily product) (2.00 g)(1.45 g)MeOH 60 ml 15 20°C 3 h.

Compound Z-P-7:

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Elemental Analysis for C <sub>25</sub> H <sub>37</sub> N <sub>3</sub> O <sub>8</sub> • 0.5H <sub>2</sub> O					
Calcd.	C, 58.13;	H, 7.41;	N, 8.13		
Found	C, 58.44;	H, 7.37;	N, 8.33		

Reference Example 8

Production of H-Gly-Gly-Asp(O<sup>t</sup>Bu)-O<sup>t</sup>Bu (P-8)

In substantially the same manner as in Reference Example 2, H-Gly-Gly-Gly-Asp(O<sup>t</sup>Bu)-O<sup>t</sup>Bu (P-8) was produced.

35		Materials	Reaction C	onditions	Products
	a)	1) Z-Gly-Gly-Gly-OH	WSC	3.44 g	Z-P-8
		(6.67 g)	HOBT	2.42g	(8.30 g)
40			DMF	100 ml	
40		2) H-Asp(O <sup>t</sup> Bu)-O <sup>t</sup> Bu	20°C	15 h	
	-	(4.00 g)			
	b)	Z-P-8	10% Pd-C	187 mg	P-8
45		(1.87 g)	MeOH	60 ml	(1.35 g)
			20°C	2 h	

Compound Z-P-8:

Elemental Analysis for C <sub>39</sub> H <sub>38</sub> N <sub>4</sub> O <sub>9</sub> :					
Calcd.	C, 56.72;	H, 6.96;	N, 10.18		
Found	C, 56.48;	H, 6.88;	N, 10.23		

Compound P-8:

Elemental Analysis for C <sub>18</sub> H <sub>32</sub> N <sub>4</sub> O <sub>7</sub> :					
Calcd.	C, 51.91;	H, 7.74;	N, 13.45		
Found	C, 51.80;	H, 8.11;	N, 13.50		

Reference Example 9

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Production of H-Gly-Gly-D-Glu(O<sup>t</sup>Bu)-O<sup>t</sup>Bu (P-9)

In substantially the same manner as in Reference Example 2, H-Gly-Gly-D-Glu(O<sup>t</sup>Bu)-O<sup>t</sup>Bu (P-9) was produced.

Products Materials Reaction Conditions 15 Z-P-9 1) Z-Gly-Gly-OH WSC 569 mg a) (1.18 g)(790 mg) HOBT 401 mg DMF 23 ml 20 2) H-D-Glu(O<sup>t</sup>Bu)-O<sup>t</sup>Bu 20°C 16 h (700 mg) Z-P-9 10% Pd-C P-9 110 mg b) 25 (1.10 g)40 ml (850 mg) MeOH

20°C 2 h

Compound Z-P-9:  $[\alpha]_D^{24}$ -9.3 ° (c = 0.53, in chloroform)

Elemental Analysis for C<sub>25</sub> H<sub>37</sub> N<sub>3</sub> O<sub>8</sub> • 0.5H<sub>2</sub>O:

Calcd. C, 58.13; H, 7.41; N, 8.13

Found C, 58.31, H, 7.42; N, 8.24

Compound P-9:  $[\alpha]_0^{23}$ -11.9 ° (c = 0.37, in chloroform)

Elemental Analysis for C<sub>17</sub> H<sub>31</sub> N<sub>3</sub> O<sub>5</sub> • 0.25H<sub>2</sub>O:

Calcd. C, 54.02; H, 8.40; N, 11.12

Found C, 53.84, H, 8.58; N, 11.24

Reference Example 10

Production of H-Gly-D-Glu(O<sup>t</sup>Bu)-O<sup>t</sup>Bu (P-10)

In substantially the same manner as in Reference Example 2, H-Gly-D-Glu(O<sup>t</sup>Bu)-O<sup>t</sup>Bu (P-10) was produced.

		Materials	Reaction Co	onditions	Products
	a)	l) Z-Gly-OH	WSC	569 mg	Z-P-10
5		(620 mg)	HOBT	401 mg	(1.23 g)
			DCM	23 ml	
		2) $H-D-Glu(O^tBu)-O^tBu$	20°C	16 h	
10		(700 mg)			,
	b)	Z-P-10	10% Pd-C	120 mg	P-10
		(1.23 g)	MeOH	40 ml	(830 mg)
15			20°C	1.5 h	

Compound Z-P-10:  $[\alpha]_D^{20}$ -10.3 • (c = 0.52, in chloroform)

Elemental Analysis for C <sub>23</sub> H <sub>34</sub> N <sub>2</sub> O <sub>7</sub> • 0.25H <sub>2</sub> O:					
Calcd.	C, 60.71;	H, 7.64;	N, 6.16		
Found	C, 60.79,	H, 7.68;	N, 6.28		

Compound P-10:  $[\alpha]_0^{20}$ -17.9 ° (c = 0.52, in chloroform)

Elemental Analysis for C <sub>15</sub> H <sub>28</sub> N <sub>2</sub> O <sub>5</sub> :					
Calcd.	C, 56.94; .	H, 8.92;	N, 8.85		
Found	C, 56.81,	H, 8.95;	N, 9.04		

Reference Example 11

55 Compound Z-P-11: m.p. 92.4-92.7 °C  $[\alpha]_0^{24}$ -3.5 ° (c = 0.51, in chloroform)

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35 Production of H-Gly-Glu(O¹Bu)-Glu(O¹Bu)-O¹Bu (P-11)

In substantially the same manner as in Reference Example 2, H-Gly-Glu(O¹Bu)-Glu(O¹Bu)-O¹Bu (P-11) was produced.

40		Materials	Reaction C	onditions	Products
	a)	l) Z-Gly-OH	WSC	318 mg	Z-P-11
		(347 mg)	HOBT	224 mg	(930 mg)
45			DCM	22 ml	
		2) H-Glu(O <sup>t</sup> Bu)-Glu(O <sup>t</sup> I	Bu)-O <sup>t</sup> Bu	20°C 16	h
		(670 mg)			
50	b)	Z-P-11	10% Pd-C	90 mg	P-11
		(870 mg)	MeOH	30 ml	(650 mg)
			20°C	2 h	•

Elemental Analysis for C <sub>32</sub> H <sub>49</sub> N <sub>3</sub> O <sub>10</sub> • 0.25H <sub>2</sub> O:					
Calcd.	C, 60.03;	H, 7.79;	N, 6.56		
Found	C, 60.18,	H, 7.81;	N, 6.26		

Compound P-11:  $[\alpha]_D^{24}$ -6.8 ° (c = 0.54, in chloroform)

Elemental Analysis for C <sub>24</sub> H <sub>4 3</sub> N <sub>3</sub> O <sub>8</sub> • 0.25H <sub>2</sub> O:					
Calcd.	C, 56.95;	H, 8.66;	N, 8.30		
Found	C, 56.92,	H, 8.74;	N, 8.06		

# 75 Reference Example 12

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Production of H-Gly-Glu(O¹Bu)-D-Glu(O¹Bu)-O¹Bu (P-12)

In substantially the same manner as in Reference Example 2, H-Gly-Glu(O¹Bu)-D-Glu(O¹Bu)-O¹Bu (P-12) was produced.

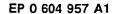
		Materials	Reaction	Conditions	Products
	a)	l) Z-Gly-OH	WSC	334 mg	Z-P-12
25					
		(364 mg)	HOBT	235 mg	(760 mg)
30			DCM	23 ml	
			20°C	16 h	
		2) H-Glu(O <sup>t</sup> Bu)-D-Glu	(O <sup>t</sup> Bu)-O <sup>t</sup> Bu		
35		(700 mg)			
	b)	Z-P-12	10% Pd-C	70 mg	P-12
		(700 mg)	MeOH	23 ml	(550 mg)
			20°C	2 h	

Compound Z-P-12:  $[\alpha]_D^{23}$ -11.5 ° (c = 0.53, in chloroform)

Elemental Analysis for C <sub>32</sub> H <sub>49</sub> N <sub>3</sub> O <sub>10</sub> :					
Calcd.	C, 60.46;	H, 7.77;	N, 6.61		
Found	C, 60.47,	H, 7.88;	N, 6.53		

Compound P-12:  $\left[\alpha\right]_{D}^{23}$ -19.8 ° (c = 0.51, in chloroform)

Elemental Analysis for C <sub>24</sub> H <sub>43</sub> N <sub>3</sub> O <sub>3</sub> •0.5H <sub>2</sub> O:					
Calcd.	C, 56.45;	H, 8.69;	N, 8.23		
Found	C, 56.52,	H, 8.77;	N, 8.25		



# Reference Example 13

Production of H-Glu(O<sup>t</sup>Bu)-Gly-Glu(O<sup>t</sup>Bu)-O<sup>t</sup>Bu (P-13)

In substantially the same manner as in Reference Example 2, H-Glu(O'Bu)-Gly-Glu(O'Bu)-O'Bu (P-13) was produced.

		Materials	Reaction Co	onditions	Products
10	a)	1) Z-Glu(O <sup>t</sup> Bu)-OH	WSC	447 mg	Z-P-13
		(785 mg)	HOBT	315 mg	(1.20 g)
			DCM	22 ml	
15		2) H-Gly-Glu(O <sup>t</sup> Bu)-O <sup>t</sup> B	u 20°C	16 h	
		(670 mg)			
	b)	Z-P-13	10% Pd-C	110 mg	P-13
20		(1.10 g)	MeOH	40 ml	(840 mg)
			20°C	2 h	

Compound Z-P-13:  $[\alpha]_D^{24}$  + 4.2 ° (c = 0.53, in chloroform)

Elemental Analysis for C<sub>32</sub>H<sub>49</sub> N<sub>3</sub>O<sub>10</sub>:

Calcd. C, 60.46; H, 7.77; N, 6.61
Found C, 60.48, H, 7.78; N, 6.88

Compound P-13:  $[\alpha]_D^{24} + 7.0$ ° (c = 0.52, in chloroform)

Elemental Analysis for C <sub>24</sub> H <sub>43</sub> N <sub>3</sub> O <sub>8</sub> • 0.25H <sub>2</sub> O:					
Calcd.	C, 56.95;	H, 8.66;	N, 8.30		
Found	C, 56.94,	H, 8.60;	N, 8.04		

40 Reference Example 14

Production of H-Glu(O<sup>t</sup>Bu)-D-Glu(O<sup>t</sup>Bu)-O<sup>t</sup>Bu (P-14)

In substantially the same manner as in Reference Example 2, H-Glu(O¹Bu)-D-Glu(O¹Bu)-O¹Bu (P-14) was produced.

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Compound Z-P-14:  $[\alpha]_D^{23}$ -7.9 (c = 0.56, in chloroform)

Elemental Analysis for C <sub>30</sub> H <sub>46</sub> N <sub>2</sub> O <sub>3</sub> :				
Calcd.	C, 62.27;	H, 8.01;	N, 4.84	
Found	C, 62.49,	H, 8.14;	N, 5.07	

Compound P-14:  $[\alpha]_D^{23} + 2.3^{\circ}$  (c = 0.61, in chloroform)

Elemental Analysis for C22 H40 N2 O7 • 1.5H2 O: C, 56.03; H, 9.19; N, 5.94 Calcd. N, 5.92 C, 56.04, H, 8.96; Found

Reference Example 15

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Production of H-Glu(O<sup>t</sup>Bu)-Glu(O<sup>t</sup>Bu)-Glu(O<sup>t</sup>Bu)-O<sup>t</sup>Bu (P-15)

In substantially the same manner as in Reference Example 2, H-Glu(O¹Bu)-Glu(O¹Bu)-Glu(O¹Bu)-O¹Bu (P-15) was produced.

40		Materials	Reaction Conditions	Products
	a)	l) Z-Glu(O <sup>t</sup> Bu)-OH	WSC 318 mg	Z-P-15
		(559 mg)	HOBT 224 mg	(1.06 g)
45			DCM 22 ml	
40			20°C 16 h	
		2) H-Glu(O <sup>t</sup> Bu)-Glu(O <sup>t</sup>	Bu)-O <sup>t</sup> Bu	
		(670 mg)		
50	b)	Z-P-15	10% Pd-C 100 mg	P-15
		(1.00 g)	MeOH 33 ml	(800 mg)
			20°C 2 h	

Compound Z-P-15:  $[\alpha]_{D}^{24}$  - 11.7 • (c = 0.53, in chloroform)

Elemental Analysis for C <sub>39</sub> H <sub>61</sub> N <sub>3</sub> O <sub>12</sub> :					
Calcd.	C, 61.32;	H, 8.05;	N, 5.50		
Found	C, 61.40,	H, 8.13;	N, 5.28		

Compound Z-15:  $[\alpha]_{0}^{24}$  - 13.9 • (c = 0.49, in chloroform)

Elemental Analysis for C <sub>31</sub> H <sub>55</sub> N <sub>3</sub> O <sub>10</sub> :					
Calcd.	C, 59.12;	H, 8.80;	N, 6.67		
Found	C, 58.96,	H, 8.91;	N, 6.80		

15 Reference Example 16

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Production of H-Glu(O¹Bu)-Glu(O¹Bu)-D-Glu(O¹Bu)-O¹Bu (P-16)

In substantially the same manner as in Reference Example 2, H-Glu(O¹Bu)-Glu(O¹Bu)-D-Glu(O¹Bu)-O¹Bu 20 (P-16) was produced.

Materials Reaction Conditions Products
a) 1) 
$$Z-Glu(O^tBu)-OH$$
 WSC 334 mg  $Z-P-16$  (586 mg) HOBT 235 mg (880 mg)

2) 
$$H-Glu(O^{t}Bu)-D-Glu(O^{t}Bu)-O^{t}Bu$$
  
(700 mg)

Compound Z-P-16:  $[\alpha]_D^{24}$ -20.6 ° (c = 0.52, in chloroform)

Elemental Analysis for C <sub>39</sub> H <sub>61</sub> N <sub>3</sub> O <sub>12</sub> :  Calcd. C, 61.32; H, 8.05; N, 5.50					
Calcd.	C, 61.32;	H, 8.05;	N, 5.50		
Found	C, 61.20,	H, 8.26;	N, 5.31		

Compound P-16:  $[\alpha]_0^{24}$ -25.5 ° (c = 0.54, in chloroform)

Elemental Analysis for C <sub>3 1</sub> H <sub>55</sub> N <sub>3</sub> O <sub>10</sub> • 0.25H <sub>2</sub> O:					
Calcd.	C, 58.70;	H, 8.82;	N, 6.63		
Found	C, 58.76,	H, 8.91;	N, 6.45		

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# Reference Example 17

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Production of NH<sub>2</sub>(CH<sub>2</sub>)<sub>7</sub>CO-Glu(O<sup>t</sup>Bu)-O<sup>t</sup>Bu (P-17)

a) To a solution of 8-(benzyloxycarbonylamino)-octanoic acid (1.55 g) and H-Glu(0¹Bu)-0¹Bu hydrochloride (1.88 g) in DMF (50 ml) were added TEA (1.76 ml) and DEPC (1.29 g), and the mixture was stirred for 24 hours at 20 °C. To the reaction mixture was added water, which was subjected to extraction with ethyl acetate. The extract was washed with a 5% (w/v, hereinafter in the same way) aqueous solution of citric acid, a saturated aqueous solution of sodium hydrogencarbonate and a saturated aqueous saline solution, which was dried over anhydrous sodium sulfate. The solvent was distilled off, and the residue was purified by means of a silica gel column chromatography (hexane:ethyl acetate = 2:1) to give Z-NH-(CH<sub>2</sub>)<sub>7</sub>CO-Glu(O¹Bu)-O¹Bu (Z-P-17) (2.40 g. yield 85%) as a colorless oily product. IR (neat) v: 3310, 1720, 1650 cm<sup>-1</sup>

<sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.10-1.37 (8H,m), 1.44 (9H,s), 1.47 (9H,s), 1.48-2.43 (8H,m), 3.18 (2H,q,J=6.6Hz), 4.41-4.54 (1H,m), 4.70-4.88 (1H,m), 5.10 (2H,s), 6.15 (1H,d,J=8.0Hz), 7.27-7.37 (5H,m)

b) To a solution of the compound Z-P-17 (1.19 g) obtained as above in ethanol (20 ml) was added 10% palladium-carbon (100 mg). The mixture was stirred for 2 hours at 20°C in hydrogen streams. The reaction mixture was subjected to filtration to obtain P-17 (739 mg, yield 83%) as a colorless oily product.

IR (neat)  $\nu$ : 3280, 1720, 1660 cm<sup>-1</sup>

<sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.20-1.40 (6H,m), 1.44 (9H,s), 1.47 (9H,s), 1.52-2.38 (10H,m), 2.38-2.65 (2H,m), 2.73 (2H,t,J=7.0Hz), 4.42-4.55 (1H,m), 6.21 (1H,d,J=8.2Hz)

Reference Example 18

Production of NH<sub>2</sub> (CH<sub>2</sub>)<sub>11</sub> CO-Glu(O<sup>t</sup>Bu)-O<sup>t</sup>Bu (P-18)

In substantially the same manner as in Reference Example 17,  $NH_2(CH_2)_{11}CO$ -Glu( $O^tBu$ )- $O^tBu$  (P-18) was synthesized.

			Materials	Rea	ction C	ondit	ions	Produc	cts
	a )	1)	$Z-NH(CH_2)_{11}COOH$		DEPC	2.4	5 g	Z-P-18	3
35			(3.49 g)		TEAT	3.5	ml	(2.32	g)
					DMF	100	ml		
		2)	H-Glu(O'Bu)-O'Bu.HC	1	20°C	24	h		
40			(2.96 g)						
	b) -		Z-P-18	10%	Pd/C	230	mg	P-18	
			(2.31 g)		EtOH	50	ml	(1.78	g)
					20°C	2	h		

Compound Z-P-18: IR (neat)  $\nu$ : 3310, 1725, 1650 cm<sup>-1</sup> <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.10-1.41 (16H,m), 1.44 (9H,s), 1.47 (9H,s), 1.41-2.44 (8H,m), 3.19 (2H,dt,J=7.0, 7.0Hz), 4.42-4.56 (1H,m), 5.10 (2H,s), 6.13 (1H,d,J=7.6Hz), 7.23-7.40 (5H,m)

Compound P-18: IR (neat)  $\nu$ : 3280, 1730, 1650 cm<sup>-1</sup> <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.05-1.40 (16H,m), 1.44 (9H,s), 1.47 (9H,s), 1.55-1.72 (4H,m), 1.72-2.50 (10H,m), 2.68 (2H,t,J=7.0Hz), 3.19 (2H,t,J=7.6Hz), 4.43-4.57 (1H,m), 6.14 (1H,d,J=7.8Hz)

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### Reference Example 19

Production of 4-aminobenzoyl-Glu(O<sup>t</sup>Bu)-O<sup>t</sup>Bu (P-19)

To a solution of 4-aminobenzoic acid (6.86 g) and H-Glu(O¹Bu)-O¹Bu hydrochloride (14.8 g) in DMF (200 ml) were added TEA (17 ml) and DEPC (12.2 g). The mixture was stirred for 48 hours at 20 °C. To the reaction mixture was added water, which was subjected to extraction with ethyl acetate. The extract was washed with a 5% aqueous solution of citric acid, a saturated aqueous solution of sodium hydrogencarbonate and a saturated aqueous saline solution, followed by drying over anhydrous sodium sulfate. The solvent was distilled off, and the residue was washed with ether to give the compound P-19 (16.2 g, yield 86%) as a white powdery product.

IR (KBr) v: 3450, 3390, 3370, 1715, 1625, 1610 cm<sup>-1</sup>

<sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.42 (9H,s), 1.49 (9H,s), 1.90-2.53 (4H,m), 3.99 (2H,br s), 4.60-4.73 (1H,m), 6.61-6.72 (2H,m), 6.76 (1H,d,J=7.6Hz), 7.60-7.71 (2H,m)

Reference Example 20

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Production of 4-(glycylamino)benzoyl-Glu(O¹Bu)-O¹Bu (P-20)

a) To a solution of the compound P-19 (568 mg) obtained in Reference Example 19 in pyridine (2 ml) was added phosphorus trichloride (0.087 ml), and the mixture was stirred for 2 hours at room temperature. To the mixture was further added Z-glycine (209 mg), which was stirred for 18 hours at 20 °C. To the reaction mixture was added water, which was subjected to extraction with ethyl acetate, followed by drying over anhydrous sodium sulfate. The solvent was distilled off, and the residue was purified by means of a silica gel column chromatography (hexane: ethyl acetate = 1:1) to give the compound Z-P-20 (414 mg, yield 73%) as a white powdery product.

IR (KBr) v: 3310, 1700, 1640, 1600 cm<sup>-1</sup>

<sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ: 1.42 (9H,s), 1.49 (9H,s), 1.95-2.55 (4H,m), 4.02 (2H,d,J=5.8Hz), 4.59-4.72 (1H,m), 5.17 (2H,s), 5.59-5.75 (1H,m), 7.09 (1H,d,J=7.8Hz), 7.30-7.42 (5H,m), 7.53 (2H,d,J=8.0Hz), 7.77 (2H,d,J=8.0Hz), 8.30-8.50 (1H,m)

b) To a solution of the compound Z-P-20 (414 mg) in ethanol (4 ml) was added 10% palladium-carbon (40 mg). The mixture was stirred for 2 hours at 20 °C in hydrogen streams. The reaction mixture was subjected to filtration to give the compound P-20 (296 mg, yield 94%) as a white powdery product. IR (KBr)  $\nu$ : 3425, 3400, 1740, 1720, 1680, 1640, 1610 cm<sup>-1</sup>

<sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.41 (9H,s), 1.48 (9H,s), 1.90-2.55 (4H,m), 3.59 (2H,br s), 4.58-4.72 (1H,m), 7.02-7.15 (1H,m), 7.65 (2H,d,J=8.8Hz), 7.77 (2H,d,J=8.8Hz), 9.65-9.77 (1H,m)

Reference Example 21

40 Production of NH<sub>2</sub>(CH<sub>2</sub>)<sub>5</sub> CO-Glu(O<sup>t</sup>Bu)-O<sup>t</sup>Bu (P-21)

In substantially the same procedure as in Reference Example 17, NH<sub>2</sub>(CH<sub>2</sub>)<sub>5</sub>CO-Glu(O<sup>t</sup>Bu)-O<sup>t</sup>Bu (P-21) was synthesized.

	Materials	Reaction Conditions	Products
a)	1) Z-NH(CH <sub>2</sub> ) <sub>5</sub> COOH	DEPC 2.45 g	Z-P-21
	(2.92 g)	TEA 4.9 ml	(5.30 g)
	•	DMF 50 ml	
	2) H-Glu(O <sup>t</sup> Bu)-O <sup>t</sup> Bu.HC	0°C 30 min	

Compound P-21:  $^{1}$ H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.389 (2H,m), 1.444 (9H,s), 1.50-1.75 (4H,m), 1.75-2.15 (2H,m), 2.15-2.38 (4H,m), 2.796 (2H,t,J=6.6Hz), 3.320 (2H,bs), 4.460 (1H,m), 6.400 (1H,d,J=7.4Hz)

10 IR (neat)  $\nu$ : 3270, 3040, 2970, 2920, 2855, 1725, 1645. 1540, 1470, 1450, 1390, 1360, 1320, 1290, 1250,

Reference Example 22

1220, 1150 cm<sup>-1</sup>

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Production of NH2 (CH2) NHCO-Glu(O'Bu)-O'Bu (P-22)

a) Hexamethylenediamine (2.32 g, 20 mmol) was dissolved in methanol (20 ml). To the solution were added a 1N aqueous solution of sodium hydroxide (22 ml) and a solution of carbobenzoxy chloride (3.4 g, 20 mmol) in THF (20 ml) simultaneously. The mixture was stirred for one hour at 20 °C. The reaction mixture was concentrated to dryness, and the concentrate was dissolved in chloroform (100 ml). The solution was washed with water and a saturated aqueous saline solution. The organic layer was dried over anhydrous sodium sulfate, which was concentrated to dryness under reduced pressure. The concentrate was allowed to be adsorbed on a silica gel column (20 g) processed with ammonia. Elution was then conducted with chloroform-methanol-water (65:25:4) to give a colorless oily product. This product was dissolved in TEA (1.34 ml, 9.64 mmol) and DCM (10 ml). To the solution was added, under ice-cooling, 4-nitrophenyl chloroformate (971 mg, 4.82 mmol), and the mixture was stirred for one hour. To the reaction mixture was added chloroform (20 ml), which was washed with water and a saturated aqueous saline solution. The organic layer was dried over anhydrous sodium sulfate, which was concentrated to dryness under reduced pressure. The concentrate was washed with ethyl ether to give 6-benzyloxy-carbonylamino-1-(p-nitrophenyloxycarbonyl)aminohexane as colorless crystals (1.16 g, yield 14.5%)

<sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ: 1.25-2.70 (8H,m), 3.225 (4H,m), 4.742 (1H,br s), 5.099 (2H,s), 5.25 (1H,br s), 7.284 (2H,d,J = 9.0Hz), 8.237 (2H,d,J = 9.0Hz)

b) The compound obtained as above (1.16 g, 2.79 mmol) and H-Glu(O¹Bu)-O¹Bu hydrochloride (825 mg, 2.79 mmol) were dissolved DCM (10 ml). To the solution were added DMAP (680 mg, 5.58 mmol) and TEA (0.77 ml, 5.58 mmol), and the mixture was stirred for 2 hours at 20 °C. To the reaction mixture was added chloroform (20 ml). The mixture was washed with water at pH 3.5, followed by washing with a saturated aqueous solution of sodium hydrogencarbonate. The organic layer was dried over anhydrous sodium sulfate, which was concentrated under reduced pressure. The concentrate was purified by means of a silica gel column [20 g, ethyl acetate - n-hexane (2:3)] to give Z-NH(CH<sub>2</sub>)<sub>6</sub> NHCO-Glu(O¹Bu)-O¹Bu (Z-P-22) as-a white solid product (1.50 g, yield 100%).

<sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ: 1.20-1.38(8H,m), 1.436 (9H,s), 1.453 (9H,s), 1.70-2.22 (2H,m), 2.326 (2H,m), 3.169 (4H,m), 4.340 (1H,m), 4.626 (1H,br s), 4.869 (1H,br s), 5.045 (1H,d,J=7.6Hz), 5.103 (2H,s), 7.352 (5H,s) IR (neat)  $\nu$ ; 3350, 2980, 2940, 2855, 1730, 1640, 1560, 1455, 1395, 1370, 1330, 1255, 1150, 1100, 1030, 850, 750, 735, 700 cm<sup>-1</sup>

c) 10% Palladium-carbon (250 mg) was suspended in methanol (20 ml). The suspension was stirred for 30 minutes in hydrogen streams, to which was added the compound Z-P-22 (1.50 g, 2.79 mmol), followed by stirring for 1.5 hour at 20°C in hydrogen streams. From the reaction mixture was removed insolubles by filtration. The filtrate was concentrated to dryness under reduced pressure to give P-22 as a colorless solid product (1.12 g, yield 100%).

<sup>1</sup>H-NMR (CDCl<sub>2</sub>) δ: 1.391 (10H,m), 1.431 (9H,s), 1.451 (9H,s), 1.747 (2H,m), 1.60-2.15 (2H,m), 2.90-3.20 (4H,m), 4.321 (1H,m), 6.013 (2H,m)

IR (neat) v: 3350, 2980, 2930, 2860, 1730, 1640, 1560, 1500, 1475, 1455, 1390, 1365, 1250, 1150 cm<sup>-1</sup>

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### Reference Example 23

Production of 4-(aminomethyl)benzoyl-Glu(O¹Bu)-O¹Bu hydrochloride (P-23)

- a) To a solution of 4-(aminomethyl)benzoic acid (25 g) in a 2N aqueous solution of sodium hydroxide (100 ml) was added dropwise, under ice-cooling, a solution of benzyloxycarbonyl chloride (33.8 g) in THF (50 ml), and the mixture was stirred for 2 hours. Resultant precipitate was collected by filtration, which was washed with water, 1N HCl and ether, followed by drying under reduced pressure to give 4-(Z-aminomethyl)benzoic acid as a white powdery product (10.5 g, yield 22%). IR (neat) ν: 3313, 1684, 1612, 1529, 1430, 1322, 1292, 1253, 1054, 761, 696 cm<sup>-1</sup>
  - <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 4.26 (2H,d,J=6.2Hz), 5.05 (2H,s), 7.31 (2H,d,J=8.0Hz), 7.36 (5H,s), 7.87 (2H,d,J=8.0Hz)
  - b) To a solution of 4-(Z-aminomethyl)benzoic acid (1.0 g) synthesized as above, H-Glu(O¹Bu)-O¹Bu hydrochloride (1.15 g) and DEPC (860 mg) in DMF (20 ml) was added dropwise TEA (1.06 g). The mixture was stirred for one hour at 20 °C. The solvent was distilled off under reduced pressure, and the residue was purified by means of a silica gel column (n-hexane:ethyl acetate = 2:1) to give 4-(Z-aminomethyl)benzoyl-Glu(O¹Bu)-O¹Bu (1.86 g, yield 100%) as a colorless waxy product.
  - IR (neat) v: 1720, 1700, 1640, 1530, 1500, 1360, 1250, 1145 cm<sup>-1</sup>
  - <sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ: 1.42 (9H,s), 1.49 (9H,s), 1.90-2.55 (4H,m), 4.43 (2H,d,J=6.0Hz), 4.66 (1H,m), 5.15 (3H,s), 7.01 (1H,d,J=7.0Hz), 7.30-7.40 (7H,m), 7.79 (2H,d,J=8.4Hz)
  - c) A suspension of the compound obtained as above (1.85 g) and 10% palladium-carbon (200 mg) in methanol (13 ml) was subjected to catalytic reduction to consume hydrogen (80 ml). The catalyst was filtered off, and to the filtrate was added a 4N HCl ethyl acetate solution (0.88 ml). The solution was concentrated to give 4-(aminomethyl)benzoyl-Glu(O¹Bu)-O¹Bu hydrochloride (P-23) as an amorphous product.
  - IR (neat)  $\nu$ : 3400, 3000, 1731, 1650, 1540, 1506, 1369, 1235, 1151 cm<sup>-1</sup> ¹H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.41 (9H,s), 1.48 (9H,s), 1.90-2.40 (4H,m), 4.13 (2H,s), 4.59 (1H,m), 7.48 (2H,d,J=8.0Hz), 7.69 (3H,d,J=8.0Hz)

### 30 Reference Example 24

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Production of 4-(N-(t-butyloxycarbonylmethyl)-aminomethyl)benzoyl-Glu(O<sup>t</sup>Bu)-O<sup>t</sup>Bu (P-24)

- a) To a solution of methyl terephthalaldehydate (3.0 g), H-Gly-O¹Bu hydrochloride (3.0 g), TEA (1.81 g), and acetic acid (1.08 g) was added, under ice-cooling, sodium cyanoborohydride (1.15 g), and the mixture was stirred for one hour at 20 °C. The reaction mixture was concentrated, which was dissolved in ethyl acetate. The solution was washed with a saturated aqueous solution of sodium hydrogencarbonate, followed by drying. The resultant mixture was concentrated, and the concentrate was purified by means of a silica gel column (n-hexane:ethyl acetate = 9:1 5:1) to give methyl 4-(N-t-butyloxycarbonylmethyl)-aminomethyl)benzoate (1.98 g, yield 39%).
  - IR (neat)  $\nu$ : 2975, 1720, 1605, 1455, 1430, 1410, 1390, 1360, 1280, 1225, 1150, 1100 cm<sup>-1</sup> <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.47 (9H,s), 3.30 (2H,s), 3.85 (2H,s), 3.91 (3H,s), 7.41 (2H,d,J=8.4Hz), 8.00 (2H,d,J=8.4Hz)
  - b) To a solution of the compound synthesized as above (1.03 g) and TEA (445 mg) in DCM (20 ml) was added dropwise, under ice-cooling, benzyloxycarbonyl chloride (690 mg). The mixture was stirred for 2 hours. The solvent was distilled off under reduced pressure. The residue was dissolved in a mixture of THF (6 ml)-MeOH(10 ml)-H<sub>2</sub>O (1 ml). To the solution was added LiOH•H<sub>2</sub>O (400 mg), and the mixture was stirred for 4 hours at 20 °C. The solvent was distilled off under reduced pressure. The residue was made acidic with a 5% aqueous solution of KHSO<sub>4</sub>, followed by extraction with chloroform. The extract solution was dried, then the solvent was distilled off under reduced pressure. The residue was purified by means of a silica gel column (n-hexane:ethyl acetate = 1:1) to give 4-(N-benzyloxycarbonyl-N-(t-butyloxycarbonylmethyl)aminomethyl)benzoic acid (337 mg, yield 23%).
    - IR (neat) v: 1735, 1705, 1700, 1450, 1410, 1360, 1290, 1230, 1220, 1150 cm<sup>-1</sup>
    - <sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ: 1.37 (1/2x9H,s), 1.45 (1/2x9H,s), 3.81 (1/2x2H,s), 3.92 (1/2x2H,s), 4.60-4.75 (2H,m), 5.20 (2H,s), 7.25-7.45 (7H,m), 8.05 (1/2x2H,d,J=8.0Hz), 8.08 (1/2x2H,d,J=8.0Hz)
  - c) A solution of the compound obtained as above (335 mg), H-Glu(O¹Bu)-O¹Bu hydrochloride (273 mg), DEPC (205 mg) and TEA (255 mg) in DMF (8 ml) was stirred for one hour at 20 °C. The solvent was distilled off under reduced pressure, and the residue was purified by means of a silica gel column (n-

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hexane:ethyl acetate = 4:1) to give 4-(N-benzyloxycarbonyl-N-(t-butyloxycarbonylmethyl)aminomethyl)-benzoyl-Glu(O¹Bu-O¹Bu (504 mg, yield 94%).

IR (neat)  $\nu$ : 3350, 2970, 1730, 1710, 1660, 1530, 1495, 1450, 1360, 1240, 1220, 1150 cm<sup>-1</sup>

 $^{1}$ H-NMR (CDCl<sub>3</sub>) δ: 1.37 (1/2x9H,s), 1.42 (9H,s), 1.45 (1/2x9H,s), 1.50 (9H,s), 1.95-2.50 (4H,m), 3.77 (1/2x2H,s), 3.88 (1/2x2H,s), 4.60-4.75 (3H,m), 5.20 (2H,s), 7.02 (1H,d,J=7.4Hz), 7.20-7.40 (7H,m), 7.77 (1/2x2H,d,J=8.0Hz), 7.80 (1/2x2H,d,J=8.0Hz)

d) A solution of the compound obtained as above (500 mg) and 10% Pd-C (200 mg) in MeOH (8 ml) was subjected to catalytic reduction to allow 20 ml of hydrogen to be consumed. The catalyst was removed, then the solvent was distilled off under reduced pressure. The residue was pulverized from n-hexane-ethyl acetate (3:1) to give the compound P-24 (265 mg, yield 67%) as a white powdery product.

IR (neat) v: 3400, 2970, 1730, 1650, 1535, 1525, 1360, 1245, 1150 cm<sup>-1</sup>

<sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 1.43 (9H,s), 1.47 (9H,s), 1.49 (9H,s), 1.90-2.50 (4H,m), 3.53 (2H,s), 4.41 (2H,s), 4.65 (1H,m), 7.18 (1H,d,J=7.2Hz), 7.70 (2H,d,J=8.0Hz), 7.88 (2H,d,J=8.0Hz)

### 15 Reference Example 25

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Production of H-Gly-Lys(Boc)-Gly-O<sup>t</sup>Bu (P-25)

In substantially the same manner as in Reference Example 2, H-Gly-Lys(Boc)-Gly-O<sup>t</sup>Bu (P-25) was produced.

		Materials	Reaction C	onditions	Products
25	a)	l) Z-Gly-OH	WSC	745 mg	Z-P-25
		(813 mg)	HOBT	525 mg	(1.90 g)
			DMF	20 ml	
30		2) H-Lys(Boc)-Gly-O'Bu	20°C	13 h	
		(1.27 g)			
	b)	Z-P-25	10% Pd/C	154 mg	P-25
		(1.54 g)	MeOH	40 ml	(1.15 g)

20°C 5 h

Compound Z-P-25:  $[\alpha]_{D}^{24} - 13.4^{\circ}$  (c = 0.58, in chloroform)

Elemental Analysis for C<sub>27</sub>H<sub>42</sub>N<sub>4</sub>O<sub>8</sub> • 0.5H<sub>2</sub>O:

Calcd. C, 57.95; H, 7.74; N, 10.01

Found C, 57.73, H, 7.44; N, 10.03

Compound P-25:  $[\alpha]_{0}^{24}$  - 23.8 • (c = 0.58, in chloroform)

Elemental Analysis for C <sub>19</sub> H <sub>36</sub> N <sub>4</sub> O <sub>6</sub> • H <sub>2</sub> O:					
Calcd.	C, 52.52;	H, 8.81;	N, 12.89		
Found	C, 52.72,	H, 8.81;	N, 12.86		

### Example 1

Production of (Fmoc-(S)-Cys-O<sup>t</sup>Bu)<sub>2</sub>

D-cystine (5.00 g, 20.8 mmol) was dissolved in 60% perchloric acid (10.2 ml), to which was added, under ice-cooling, t-butyl acetate (117 ml), and the mixture was stirred for two days at 20 °C. The reaction mixture was subjected to filtration to collect crystals. The crystals were washed with ether (150 ml), then suspended in a saturated aqueous solution of sodium hydrogencarbonate (100 ml). The suspension was subjected to extraction with ethyl acetate (2 x 150 ml). Ethyl acetate layers were combined, washed with water and dried over anhydrous sodium sulfate, followed by concentration to given a colorless oily product (4.16 g). This oily product was dissolved in THF (60 ml). To the solution were added, under ice-cooling, N-(9-fluorenylmethyloxycarbonyloxy)succinimide (8.16 g, 24.2 mmol) and N-ethylmorpholine (3.08 ml, 24.2 mmol) dissolved in THF (10 ml), and the mixture was stirred for 2.5 hours at room temperature. The reaction mixture was concentrated, and the concentrate was suspended in a 10% aqueous solution of citric acid (200 ml). The suspension was subjected to extraction with chloroform (2 x 200 ml). Chloroform layers were combined, washed with water, and then dried over anhydrous sodium sulfate, followed by concentration. The concentrate was crystallized from ethyl acetate to afford (Fmoc-(S)-Cys-O¹Bu)<sub>2</sub> as colorless crystals (8.68 g, yield 52%), m.p. 149.5-150 °C.

 $[\alpha]_D^{23}$  + 5.9 ° (c = 0.52, in chloroform)

Elemental Analysis for C <sub>44</sub> H <sub>48</sub> N <sub>2</sub> O <sub>8</sub> S <sub>2</sub> :						
Calcd.	C, 66.31;	H, 6.07;	N, 3.51;	S, 8.05		
Found	C, 66.55;	H, 6.13;	N, 3.43;	S, 7.97		

# Example 2

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Production of (2R,6S)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-OH (GC-1)

a) (Fmoc-(R)-Cys-O¹Bu)₂ (10.0 g, 12.5 mmol) prepared by the method described in a literature reference (J.W. Metzger et al., Int. J. Peptide Protein Res. 38, p.545, 1991) was dissolved in DCM (80 ml). To the solution were added, under ice-cooling, powdery zinc (3.27 g, 50.0 mmol) and amixture of MeOH-36%HCl-conc.H₂SO₄ (100:7:1) (hereinafter simply referred to as "acid mixture solution" (40 ml). The resultant mixture was stirred for 30 minutes at 20°C. To the reaction mixture was added (S)-(-)-glycidol (8.29 ml, 125 mmol), which was stirred for two hours at 40°C. The reaction mixture was concentrated until its volume is reduced to 40 ml, then insolubles were filtered off. To the filtrate was added a saturated aqueous saline solution (200 ml), and the mixture was subjected to extraction with DCM (2 x 300 ml). DCM layers were combined, dried over anhydrous sodium sulfate, and then concentrated. The concentrate was subjected to a silica-gel column chromatography, eluting with ethyl acetate-hexane (1:1, 2:1). Fractions containing the object compound were combined and concentrated to give (2R,6S)-2-Fmoc-amino-6,7-dihydroxy-4-THT-O¹Bu (GC-1a) as a white powdery product (10.9 g, yield 92%).  $\{\alpha\}_{D}^{12} + 6.7^{\circ}$  (c = 0.56, in chloroform)

Elemental Analysis for C <sub>25</sub> H <sub>31</sub> NO <sub>6</sub> • H <sub>2</sub> O:					
, 6.52 . 6.31					

b) The compound, GC-1a (11.0 g, 23.2 mmol) was dissolved in THF (200 ml), to which were added palmitic acid (19.1 g, 74.3 mmol), DIC (11.6 ml, 74.3 mmol) and 4-dimethylaminopyridine (DMAP, 1.13 g, 9.26 mmol). The mixture was stirred for 12 hours at 20 °C. The reaction mixture was concentrated, and the concentrate was suspended in 10% (w/v) aqueous solution of citric acid (400 ml). The suspension was subjected to extraction with ethyl acetate (800 ml). The ethyl acetate layer was washed with water, and then concentrated. The concentrate was subjected to a silica-gel column chromatography, eluting with ethyl acetate-hexane (1:20, 1:10), successively. Fractions containing the object compound were combined and concentrated. The concentrate was crystallized from hexane to give (2R,6S)-2-Fmoc-

amino-6,7-bis(PamO)-4-THT-O'Bu (GC-1b) as colorless crystals. (14.4 g, yield 65%). m.p. 62.5-63.5 °C

Elemental Analysis for C <sub>57</sub> H <sub>91</sub> NO <sub>8</sub> S:					
Calcd.	C, 72.03;	H, 9.65;	N, 1.47;	S, 3.37	
Found	C, 71.94;	H, 9.85;	N, 1.51;	S, 3.23	

c) The compound GC-1b (14.4 g, 15.2 mmol) was dissolved in TFA (200 ml), and the solution was left standing for 30 minutes at 20 °C. The reaction mixture was concentrated, and the concentrate was dissolved in ethyl acetate (500 ml). This solution was washed with water, dried over anhydrous sodium sulfate, followed by concentration. Resulting crystals were recrystallized from ethyl acetate-hexane to afford GC-1 as colorless crystals (12.4 g, yield 91%)

m.p. 82.5-83.5 ° C

 $[\alpha]_{D}^{23} + 14.9 \cdot (c = 0.55 \text{ in chloroform})$ 

Elemental Analysis for C <sub>5 3</sub> H <sub>83</sub> NO <sub>8</sub> S:					
	C, 71.18; C, 70.96;				

### Example 3

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Production of (2R,6R)-2-Fmoc-amino-6,7-dihydroxy-4-THT-O¹Bu (GC-2a), (2R,6R)-2-Fmoc-amino-6,7-bis-(PamO)-4-THT-O¹Bu (GC-2b), and (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-OH (GC-2)

In substantially the same manner as in Example 2, (2R,6R)-2-Fmoc-amino-6,7-dihydroxy-4-THT-O<sup>t</sup>Bu (GC-2a), (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-O<sup>t</sup>Bu (GC-2b), and (2R,6R)-2-Fmoc-amino-6,7-bis-(PamO)-4-THT-OH (GC-2) were produced.

		Materials (g)	Reac	tion Conditions	Products (g)
35	a)	(Fmoc-(R)-Cys-O <sup>t</sup> Bu) <sub>2</sub>	1)	Zinc 3.27 g	GC-2a
		(10.0)		acid mixt. 40 m	nl (10.9)
				soln.	
40				DCM 80 ml	
	-			20°C 20 min	
			2)	(R)-(+)-glycide	1
				supplemented	
45				8.29 ml	
				40°C 2.5 h	
	b)	GC-2a	palmi	tic acid 6.33 g	GC-2b
50		(3.90)	DIC	3.87 ml	(4.76)
			DMAP	402 mg	
			THF	70 ml	
55			20°C	13 h	
- <del>-</del>	c)	GC-2b	TFA	50 ml	GC-2
		(2.50)	20°C	1.5 h	(2.20)

Compound GC-2a:  $[\alpha]_D^{21}$  - 8.8 ° (c = 0.65, in chloroform)

Elemental Analysis for C <sub>25</sub> H <sub>31</sub> NO <sub>5</sub> S•0.5H <sub>2</sub> O					
Calcd.	C, 62.22;	H, 6.68;	N, 2.90;	S, 6.64	
Found	C, 62.14;	H, 6.66;	N, 2.81;	S, 6.54	

Compound GC-2b: m.p. 58.0 °C

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Elemental Analysis for C <sub>57</sub> H <sub>91</sub> NO <sub>8</sub> S:					
Calcd.	C, 72.03;	H, 9.65;	N, 1.47;	S, 3.37	
Found	C, 71.94;	H, 9.58;	N, 1.43;	S, 3.36	

Compound GC-2: m.p.  $90.0 \,^{\circ}$  C  $[\alpha]_{D}^{20} + 12.9 \,^{\circ}$  (c = 0.73, in chloroform)

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Elemental Analysis for C <sub>53</sub> H <sub>83</sub> NO <sub>8</sub> S:					
Calcd.	C, 71.18;	H, 9.35;	N, 1.57;	S, 3.59	
Found	C, 71.23;	H, 9.12;	N, 1.54;	S, 3.47	

# 25 Example 4

Production of (2S,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-O'Bu (GC-3a), (2S,6R)-2-Fmoc-amino-6,7-bis-(PamO)-4-THT-O'Bu (GC-3b), and (2S,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-OH (GC-3)

In substantially the same manner as in Example 2, (2S,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-O¹Bu (GC-3a), (2S,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-O¹Bu (GC-3b), and (2S,6R)-2-Fmoc-amino-6,7-bis-(PamO)-4-THT-OH (GC-3) were produced.

Products (g) Materials (q) Reaction Conditions 35 (Fmoc-(S)-Cys-O<sup>t</sup>Bu)<sub>2</sub> GC-3a .656 mg a) 1) Zinc (2.00)acid mixt.8.0 ml (2.18)soln. 40 DCM 16 ml 20°C 30 min 2) (S)-(-)-glycidol45 supplemented 0.67 ml 40°C 5 h 50 palmitic acid 2.60 g GC-3b b) GC-3a 1.59 ml (2.44)DIC (1.50)DMAP 154 mg 30 ml THF 55 20°C 13 h GC-3 TFA 40 ml C) GC-3b

(2.00)

20°C

1.5 h

(1.80)

5 Compound GC-3a:  $[\alpha]_{0}^{23}$  - 7.6 • (c = 0.67, in chloroform)

Elemental Analysis for C <sub>25</sub> H <sub>31</sub> NO <sub>6</sub> :				
	C, 63.40; C, 63.12;			

Compound GC-3b: m.p. 62.5-63.0 ° C

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Elemental Analysis for C <sub>57</sub> H <sub>91</sub> NO <sub>8</sub> S:					
Calcd.	C, 72.03;	H, 9.65;	N, 1.47;	S, 3.37	
Found	C, 72.04;	H, 9.78;	N, 1.49;	S, 3.38	

20 Compound GC-3: m.p. 82.5-83.0 ° C  $[\alpha]_{c}^{23}$  - 16.0 ° (c = 0.51, in chloroform)

Elemental Analysis for C <sub>53</sub> H <sub>83</sub> NO <sub>8</sub> S:					
	C, 71.18; C, 71.20;			r e	

Example 5

Production of (2S,6S)-2-Fmoc-amino-6,7-dihydroxy-4-THT-O<sup>t</sup>Bu (GC-4a), (2S,6S)-2-Fmoc-amino-6,7-bis-(PamO)-4-THT-O<sup>t</sup>Bu (GC-4b), and (2S,6S)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-OH (GC-4)

In substantially the same manner as in Example 2, (2S,6S)-2-Fmoc-amino-6,7-dihydroxy-4-THT-O<sup>1</sup>Bu (GC-4a), (2S,6S)-2-Fmoc-amino-6,7-bis-(PamO)-4-THT-OH (GC-4) were produced.

40 Materials (g) Reaction Conditions Products (g)

a) 
$$(Fmoc-(S)-Cys-O^{t}Bu)_{2}$$
 1) Z  
(4.00) a

Zinc 1.3

1.31 g GC-4a

acid mixt.soln.

(4.01)

16 ml

DCM 32 ml 20°C 20 min

2) (S)-(-)-glycidol

supplemented

3.33 ml

40°C 2.5 h

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Compound GC-4a:  $[\alpha]_D^{23}$  + 8.4 ° (c = 0.67, in chloroform)

Elemental Analysis for C<sub>25</sub>H<sub>31</sub>NO<sub>6</sub>S

Calcd. C, 63.40; H, 6.60; N, 2.96; S, 6.77

Found C, 63.15; H, 6.47; N, 2.88; S, 6.67

Compound GC-4b: m.p. 58.0 °C

Elemental Analysis for C<sub>57</sub>H<sub>91</sub>NO<sub>8</sub>S:

Calcd. C, 72.03; H, 9.65; N, 1.47; S, 3.37

Found C, 72.01; H, 9.55; N, 1.34; S, 3.36

Compound GC-4: m.p. 88.5-89.0 °C  $[\alpha]_D^{23}$  - 13.1 ° (c = 0.56, in chloroform)

Elemental Analysis for C <sub>53</sub> H <sub>83</sub> NO <sub>8</sub> S:					
1	C, 71.18; C, 71.20;				

# Example 6

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Production of (2R,6S)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-Glu-Thr-Thr-OH (Compound 1)

a) The compound GC-1 (1.79 g) synthesized in Example 2 was dissolved in DMF (20 ml), to which were added, under ice-cooling, HONB (394 mg), DIC (344  $\mu$ 1) and H-Gly-Gly-Gly-Glu(O¹Bu)-Thr(¹Bu)-Thr(¹Bu)-O¹Bu (P-1) (1.64 g) obtained in Reference Example 1. The mixture was stirred for 24 hours at 20 °C. The reaction mixture was concentrated, which was then dissolved in chloroform. The solution was washed with a 10% (w/v) aqueous solution of citric acid, water, a saturated aqueous solution of sodium hydrogencarbonate and water, successively. The chloroform layer was dried over anhydrous sodium sulfate, which was then concentrated. To the concentrate was added acetonitrile. Resulting precipitations were collected by filtration to obtain (2R,6S)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Glu-(O¹Bu)-Thr(¹Bu)-O¹Bu (1a) as a white powdery product (3.16 g, yield 97%). [ $\alpha$ ] $^{18}$  + 6.5 ° (c = 0.54, in chloroform)

Elemental Analysis for C <sub>88</sub> H <sub>145</sub> N <sub>7</sub> O <sub>18</sub> S:					
Calcd.	C, 65.20;	H, 9.02;	N, 6.05;	S, 1.98	
Found	C, 64.90;	H, 8.83;	N, 5.86;	S, 1.78	

Amino acid analysis [6N HCl,  $110 \,^{\circ}$  C, hydrolysis for 20 hours; Values in parentheses show theoretical ones.]: Glu 1.00 (1); Thr 1.93 (2); Gly 2.96 (3) FAB-MS (M + Na) = 1643 (theoretical value = 1643)

b) The compound 1a (2.70 g) was dissolved in DMF (27 ml). To the solution was added piperidine (2.7 ml), and the mixture was stirred for one hour at 20 °C. The reaction mixture was concentrated and subjected to a silica-gel column chromatography, eluting with chloroform-methanol (50:1, 20:1), successively. Fractions containing the object compound were combined and concentrated to leave (2R,6S)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Glu(O¹Bu)-Thr(¹Bu)-Thr(¹Bu)-O¹Bu (1b) as a white powdery product (2.10 g, yield 90%).

 $[\alpha]_0^{18} + 7.1 \cdot (c = 0.49, in chloroform)$ 

Elemental Analysis for C <sub>73</sub> H <sub>135</sub> N <sub>7</sub> O <sub>16</sub> S:					
Calcd.	C, 62.67;	H, 9.73;	N, 7.01;	S, 2.29	
Found	C, 62.55;	H, 9.85;	N, 6.95;	S, 2.21	

Amino acid analysis [6N HCl, 110 °C, hydrolysis for 20 hours; Values in parentheses show theoretical ones.]: Glu 1.00 (1); Thr 1.92 (2); Gly 2.95 (3) FAB-MS (M + Na) = 1399 (theoretical value = 1399)

c) The compound 1b (200 mg) was dissolved in TFA (2.0 ml), which was left standing for 1.5 hour at 20°C. The reaction mixture was concentrated, to which was added acetonitrile. Resulting precipitates were collected by filtration to obtain the compound 1 as a white powdery product (164 mg).

Elemental Analysis for C <sub>57</sub> H <sub>103</sub> N <sub>7</sub> O <sub>16</sub> S•1.5H <sub>2</sub> O:					
Calcd.	C, 56.98;	H, 8.89;	N, 8.16;	S, 2.67	
Found	C, 57.04;	H, 8.80;	N, 8.11;	S, 2.72	

### Example 7

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Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-Glu-Thr-Thr-OH (Compound 2)

In substantially the same manner as in Example 6, (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-Gly-Glu(O¹Bu)-Thr(¹Bu)-O¹Bu (2a), (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-Glu-(O¹Bu)-Thr(¹Bu)-Thr(¹Bu)-O¹Bu(2b), and (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-Glu-Thr-Thr-OH (Compound 2) were produced.

		Materials (mg)	Reaction	Conditions	Products (mg)
	a)	GC-2	P-1	458 mg	2a
5		(500)	HONB	110 mg	(873)
			DIC	96 µl	
			DMF	5.0 ml	
			20°C	15 h	
10	b)	2a ·	piperidi	ne 0.70 ml	2b
		(770)	DMF	7.0 ml	(623)
			20°C	1 h	
15			silica-g	eì	
			(chlorof	orm-methanol	L)
			50:1, 20	:1	
20	c)	2b	TFA	1.5 ml	2
20		(150)	20°C	1.5 h	(125)

Compound 2a:  $[a]_{5}^{20} + 3.4^{\circ}$  (c = 0.66, in chloroform)

Elemental Analysis for C <sub>88</sub> H <sub>145</sub> N <sub>7</sub> O <sub>18</sub> S•0.5H <sub>2</sub> O:					
Calcd.	C, 64.84;	H, 9.03;	N, 6.01;	S, 1.97	
Found	C, 64.88;	H, 9.18;	N, 6.08;	S, 1.95	

Compound 2b:  $[\alpha]_{b}^{20}$  + 4.9 • (c = 0.55, in chloroform)

Elemental Analysis for C <sub>73</sub> H <sub>135</sub> N <sub>7</sub> O <sub>16</sub> S•0.5H <sub>2</sub> O:					
Calcd.	C, 62.27;	H, 9.74;	N, 6.96;	S, 2.28	
Found	C, 62.31;	H, 9.73;	N, 6.99;	S, 2.19	

Compound 2:  $[\alpha]_D^{21}$  - 2.3 ° (c = 0.58, in 5% TFA-chloroform)

Elemental Analysis for C <sub>53</sub> H <sub>103</sub> N <sub>7</sub> O <sub>16</sub> S•1.5H <sub>2</sub> O:					
Calcd.	C, 56.98;	H, 8.89;	N, 8.16;	S, 2.67	
Found	C, 56.72;	H, 8.62;	N, 8.11;	S, 2.63	

Example 8

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Production of (2S,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-Glu-Thr-Thr-OH (Compound 3)

In substantially the same manner as in Example 6, (2S,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-Glu(O¹Bu)-Thr(¹Bu)-O¹Bu (3a), (2S,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-Glu-(O¹Bu)-Thr(¹Bu)-Thr(¹Bu)-O¹Bu (3b), and (2S,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-Gly-Thr-Thr-OH (Compound 3) were produced.

		Materials (mg)	Reaction	Conditions	Products (mg)
	a)	GC-3	P-1	458 mg	3a
5		(500)	HONB	110 mg	(885)
3			DIC	96 µl	
			DMF	5.0 ml	
			20°C	15 h	
10	p)	3a	piperidi	ne 0.80 ml	3b
		(830)	DMF	8.0 ml	(624)
			20°C	1.5 h	
15			silica-g	el	
			(chlorof	orm-methanol	L)
			50:1, 20	:1	
20	c)	3b	TFA	1.6 ml	3
		(160)	20°C	1.5 h	(130)

Compound 3a:  $[\alpha]_D^{20} + 15.9$ ° (c=0.51, in chloroform)

Elemental Analysis for C<sub>88</sub>H<sub>145</sub>N<sub>7</sub>O<sub>18</sub>S•0.5H<sub>2</sub>O:

Calcd. C, 64.84; H, 9.03; N, 6.01; S, 1.97

Found C, 64.63; H, 9.07; N, 5.80; S, 2.21

Compound 3b:  $[\alpha]_D^{20}$  + 21.4° (c=0.64, in chloroform)

Elemental Analysis for C<sub>73</sub>H<sub>135</sub>N<sub>7</sub>O<sub>16</sub>S:

Calcd. C, 62.68; H, 9.73; N, 7.01; S, 2.29
Found C, 62.75; H, 9.41; N, 7.05; S, 2.40

Compound 2:  $[\alpha]_{D}^{21}$  - 21.7° (c = 0.63, in 5% TFA-chloroform)

Elemental Analysis for C<sub>57</sub>H<sub>103</sub>N<sub>7</sub>O<sub>16</sub>S•H<sub>2</sub>O:

Calcd. C, 57.41; H, 8.88; N, 8.22; S, 2.69

Found C, 57.38; H, 8.66; N, 8.27; S, 2.59

Example 9

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Production of (2S,6S)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-Glu-Thr-Thr-OH (Compound 4)

In substantially the same manner as in Example 6, (2S,6S)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-Glu(O¹Bu)-Thr(¹Bu)-Thr(¹Bu)-O¹Bu (4a), (2S,6S)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-Glu-(O¹Bu)-Thr(¹Bu)-Thr(¹Bu)-O¹Bu (4b), and (2S,6S)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-Glu-Thr-Thr-OH (Compound 4) were produced.

		Materials (mg)	Reaction	Conditions	Products (mg)
	a)	GC-4	P-1	458 mg	4a
5		(500)	HONB	110 mg	(879)
			DIC	· 96 μ1	
			DMF	5.0 ml	
10			20°C	15 h	
70	b)	4a	piperidi	ne 0.80 ml	4b
		(820)	DMF	8.0 ml	(651)
			20°C	1 h	
15			silica-g	el	
			(chlorof	orm-methano	1)
			50:1, 20	:1	
20					
	۵)	4 h			
	c)	4b	TFA	1.6 ml	4
25		(160)	20°C	1.5 h	(131)

Compound 4a:  $[\alpha]_D^{20}$  + 15.3° (c = 0.65, in chloroform)

Elemental Analysis for C <sub>88</sub> H <sub>145</sub> N <sub>7</sub> O <sub>18</sub> S:						
Calcd.	C, 65.20;	H, 9.02;	N, 6.05;	S, 1.98		
Found	C, 65.05;	H, 9.05;	N, 6.07;	S, 1.86		

Compound 4b:  $[\alpha]_{0}^{20}$  + 24.4 ° (c = 0.62, in chloroform)

Element	Elemental Analysis for C <sub>73</sub> H <sub>135</sub> N <sub>7</sub> O <sub>16</sub> S:						
Calcd.	C, 62.68;	H, 9.73;	N, 7.01;	S, 2.29			
Found	C, 62.53;	H, 9.48;	N, 6.93;	S, 2.31			

Compound 4:  $[\alpha]_D^{21}$  - 15.6 ° (c = 0.5, in 5% TFA-chloroform)

Elementa	Elemental Analysis for C <sub>53</sub> H <sub>103</sub> N <sub>7</sub> O <sub>16</sub> S•1.5H <sub>2</sub> O:			
Calcd.	C, 56.98;	H, 8.89;	N, 8.16;	S, 2.67
Found	C, 56.74;	H, 8.57;	N, 8.03;	S, 2.72

Example 10

Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-Gly-Glu-Thr-OH (Compound 5)

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		Materials (g)	Reacti	on Conditions	Products (g)
	a)	GC-2	P-2	809 mg	5a
5		(1.23)	HONB	271 mg	(1.91)
			DIC	0.24 ml	
			DMF	20 ml	
10			20°C	. 19 h	
	b)	5a	piperi	dine 1.3 ml	5b
		(1.30)	DCM	13 ml	(0.99)
15			20°C	1.3 h	
15			silica	-gel	
			(chlor	oform-methano	1)
20					
			49:1,	19:1, 14:1	
	c)	5b	TFA	2 ml	5
25		(0.49)	20°C	2 h	(0.43)

Compound 5a:  $[\alpha]_D^{23} - 6.9$  (c = 0.51, in chloroform)

Element	Elemental Analysis for C <sub>80</sub> H <sub>130</sub> N <sub>6</sub> O <sub>16</sub> S:					
	C, 65.63; C, 65.59;					

Compound 5b:  $[\alpha]_0^{23}$  - 6.9 ° (c = 0.50, in chloroform)

Elemental Analysis for C <sub>65</sub> H <sub>120</sub> N <sub>6</sub> O <sub>14</sub> S:						
	C, 62.87; C, 62.96;					

Compound 5:  $[\alpha]_D^{23} + 4.9 \cdot (c = 0.49 \text{ in } 5\% \text{ TFA-chloroform})$ 

45	Elemental Analysis for C <sub>53</sub> H <sub>96</sub> N <sub>6</sub> O <sub>14</sub> S⋅0.5H <sub>2</sub> O:					
	Calcd.	C, 58.81;	H, 9.03;	N, 7.76;	S, 2.96	
	Found	C, 58.94;	H, 8.82;	N, 7.70;	S, 2.99	

50 Example 11

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Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu-Gly-Glu-Gly-D-Glu-OH (Compound 6)

In substantially the same manner as in Example 6, (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-Glu-(O¹Bu)-Gly-Glu(O¹Bu)-Gly-D-Glu(O¹Bu)-O¹Bu (6a), (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu-Gly-Gly-Gly-Glu-Gly-D-Glu-OH (Compound 6) were produced.

	M	Materials (g)	Reaction	on Conditions	Products (g)
	a)	GC-2	P-3	470 mg	6a
5		(0.51)	HONB	113 mg	(0.71)
			DIC	0.10 ml	
			DMF	5 ml	
10			20°C	16 h	
	<b>b)</b> .	6a	piperio	dine 0.7 ml	6b
		(0.66)	DCM	6.3 ml	(0.56)
			20°C	1.0 h	
15					
			silica-	-gel	
20			(chlore	oform-methano	1)

TFA

25 (0.25) 20°C 3 h

Compound 6a:  $[\alpha]_0^{23} + 1.8^{\circ}$  (c = 0.50, in chloroform)

c)

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6b

Elemental Analysis for C<sub>88</sub> H<sub>142</sub> N<sub>6</sub> O<sub>19</sub> S:

Calcd. C, 65.24; H, 8.83; N, 5.19; S, 1.98
Found C, 65.05; H, 8.87; N, 5.15; S, 1.91

50:1, 20:1

3.0 ml

(0.20)

Compound 6b:  $\left[\alpha\right]_{D}^{23} + 1.0^{\circ}$  (c = 0.50, in chloroform)

Elemental Analysis for C<sub>73</sub>H<sub>132</sub>N<sub>6</sub>O<sub>17</sub>S⋅0.5H<sub>2</sub>O:

Calcd. C, 62.32; H, 9.53; N, 5.97; S, 2.28

Found C, 62.30; H, 9.53; N, 5.77; S, 2.19

Compound 6:  $[\alpha]_D^{23} + 3.8^{\circ}$  (c = 0.53 in 5% TFA-chloroform)

Elemental Analysis for C<sub>50</sub>H<sub>90</sub>N<sub>4</sub>O<sub>13</sub>S•H<sub>2</sub>O:

Calcd. C, 57.46; H, 8.63; N, 7.05; S, 2.69
Found C, 57.56; H, 8.60; N, 7.24; S, 2.54

# Example 12

Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-Glu-OH (Compound 7)

In substantially the same manner as in Example 6, (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-Glu(O¹Bu)-O¹Bu (7a), (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-Glu-OH (Compound 7) were produced.

	Ma	aterials (g)	Reaction	Conditions	Products (g)
	a)	GC-2	P-4	630 mg	7a
5		(1.31)	HONB	288 mg	(1.64)
			DIC	0.25 ml	
			DCM	19 ml	
10			20°C	15 h	
	b)	7a	piperidi	ne 1.2 ml	7b
		(1.18)	DCM	12 ml	(0.92)
15					
			20°C	1.3 h	
			silica-g	el	
20			(chlorof	orm-methano	1) .
			49:1, 19	:1, 14:1	
	c)	7b	TFA	2 ml	7
25		(0.45)	20°C	2 h	(0.38)

Compound 7a:  $[\alpha]_D^{23}$  - 6.4° (c = 0.50, in chloroform)

Elemental Analysis for C<sub>72</sub>H<sub>115</sub>N<sub>5</sub>O<sub>14</sub>S:

Calcd. C, 66.18; H, 8.87; N, 5.36; S, 2.45

Found C, 66.03; H, 8.87; N, 5.31; S, 2.22

Compound 7b:  $[\alpha]_D^{23}$  - 6.7 ° (c = 0.63, in chloroform)

Elementa	Elemental Analysis for C <sub>57</sub> H <sub>105</sub> N <sub>5</sub> O <sub>12</sub> S•H <sub>2</sub> O:					
Calcd.	C, 62.09;	Н, 9.78;	N, 6.35;	S, 2.91		
Found	C, 62.12;	Н, 9.59;	N, 6.36;	S, 2.87		

Compound 7:  $[\alpha]_D^{23}$  + 8.3 ° (c = 0.60 in 5% TFA-chloroform)

Elemental Analysis for C <sub>49</sub> H <sub>89</sub> N <sub>5</sub> O <sub>12</sub> S • H <sub>2</sub> O:					
Calcd.	C, 59.43;	H, 9.26;	N, 7.07;	S, 3.24	
Found	C, 59.19;	H. 8.96;	N. 6.96;	S. 3.26	

50 Example 13

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Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-D-Glu-OH (Compound 8)

In substantially the same manner as in Example 6, (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-D-Glu(O¹Bu)-O¹Bu (8a), (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-Gly-Gly-Gly-Gly-Gly-Gly-O-Glu-OH (Compound 8) were produced.

Materials (g) Reaction Condition  a) GC-2 P-5 1.53 (1.92) HONB 684 DIC 0.59 DMF 33 20 °C 15 b) 8a piperidine 0.82  (0.82) DCM 8.2 20 °C 2	g 8a l mg (3.93) ml
10 (1.92) HONB 684 DIC 0.59 DMF 33 20 °C 15 piperidine 0.82	mg (3.93)
DIC 0.59  DMF 33  20 °C 15  b) 8a piperidine 0.82	ml
DMF 33 20 °C 15 b) 8a piperidine 0.82	•
20 °C 15 piperidine 0.82	
b) 8a piperidine 0.82	m1
(0.82) DCM 8.2	h
(0.82) DCM 8.2	ml 8b
(0.82) DCM 8.2	
• •	ml (0.64)
20 - 6 2	•
-111	n
silica-gel	•
20 (chloroform-met	hanol)
49:1, 19:1, 14:	1
c) 8b TFA 2	ml 8
<sub>25</sub> (0.58) 20 °C 2	h (0.45)

Compound 8a:  $[\alpha]_0^{23}$  - 7.1 ° (c = 0.49, in chloroform)

Elemental Analysis for C<sub>72</sub>H<sub>115</sub>N<sub>5</sub>O<sub>14</sub>S:

Calcd. C, 66.18; H, 8.87; N, 5.36; S, 2.45

Found C, 66.12; H, 8.77; N, 5.52; S, 2.45

Compound 8b:  $[\alpha]_D^{23}$  - 14.5 ° (c = 0.53, in chloroform)

Elemental Analysis for C <sub>57</sub> H <sub>105</sub> N <sub>5</sub> O <sub>12</sub> S:						
Calcd.	C, 63.13;	H, 9.76;	N, 6.46;	S, 2.96		
Found	C, 63.10;	H, 9.83;	N, 6.44;	S, 2.77		

Compound 8:  $\left[\alpha\right]_{D}^{23}$  + 8.1 ° (c = 0.62 in 5% TFA-chloroform)

Elemental Analysis for C <sub>49</sub> H <sub>89</sub> N <sub>5</sub> O <sub>12</sub> S•0.5H <sub>2</sub> O:				
Calcd.	C, 59.97;	H, 9.24;	N, 7.14;	S, 3.27
Found	C, 59.82;	Н, 9.14;	N, 6.96;	S, 3.38

50 Example 14

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Production of (2R,6S)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Glu-OH (Compound 9)

		Materials (g)	Reaction	on Conditions	Products (g)
	a)	GC-1	P-4	490 mg	9a
_		(1.02)	HONB	229 mg	(1.25)
5			DIC	0.20 ml	
			DMF	15 ml	
			20 °C	18 h	
10					

9b b) 9a piperidine 1.10 ml 15 (1.10)(0.777)DCM 11 ml 20 °C 1.5 h silica-gel (chloroform-methanol) 20 49:1, 19:1 3.0 ml 9b TFA C)

20 °C

Compound 9a:  $[\alpha]_D^{24}$  - 3.3° (c = 0.51, in chloroform)

(0.300)

Eleme	Elemental Analysis for C <sub>72</sub> H <sub>115</sub> N <sub>5</sub> O <sub>14</sub> S• H <sub>2</sub> O:						
Calcd.	-, ,	H, 8.87;	N, 5.36;	S, 2.45			
Found		H, 8.89;	N, 5.46;	S, 2.58			

1.5 h

(0.266)

Compound 9b:  $[\alpha]_D^{21}$  - 5.5° (c = 0.73, in chloroform)

Elemental Analysis for C <sub>57</sub> H <sub>105</sub> N <sub>5</sub> O <sub>12</sub> S:					
	C, 63.13;	H, 9.76;	N, 6.46;	S, 2.96	
	C, 62.84;	H, 9.61;	N, 6.42;	S, 2.96	

Compound 9:  $[\alpha]_D^{21} + 13.5 \circ (c = 0.67 \text{ in } 5\% \text{ TFA-chloroform})$ 

Elemental Analysis for C <sub>49</sub> H <sub>89</sub> N <sub>5</sub> O <sub>12</sub> S•2.5H <sub>2</sub> O:					
Calcd.	C, 57.85;	H, 9.31;	N, 6.88;	S, 3.15	
Found	C, 57.89;	H, 8.82;	N, 6.88;	S, 3.05	

Example 15

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Production of (2R,6S)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-D-Glu-OH (Compound 10)

In substantially the same manner as in Example 6, (2R,6S)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-Gly- $Gly-Gly-D-Glu(O^tBu)-O^tBu \qquad (10a), \qquad (2R,6S)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-D-Glu(O^tBu)-O^tBu \qquad (2R,6S)-2-amino-6,7-bis(PamO)-4-amino-6,7-bis(PamO)-6,7-bis(PamO)-6,7-bis(PamO)-6,7-bis(PamO)-6,7-bis(PamO)-6,7-bis(PamO)-6,7-bis(PamO)-6,7-bis(PamO)-6,7-bis(PamO)-6,7-bis(PamO)-6,7-bis(PamO)-6,7-bis(PamO)-6,7-bis(PamO)-6,7-bis(PamO)-6,7-bis(PamO)-6,7-bis(PamO)-6,7-bis(PamO)-6,7-bis(PamO)-6,7-bis(PamO)-6,$ (10b) and (2R,6S)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-D-Glu-OH (Compound 10) were produced.

	Mater	ials (g)	Reaction (	Conditions	Products (g)
a	.)	GC-1	P-5	1.61 g	10a
5		(2.41)	HONB	541 mg	(3.45)
			DIC	0.47 ml	
			DMF	25 ml	
10					
			20 °C	19 h	
b 15	)	10a	piperidin	e 1.10 ml	10b
· ·		(1.10)	DMF	11 ml	(0.783)
			20 °C	1.5 h	
			silica-ge	1	
20			(chlorofo	rm-methanol	.)
			49:1, 19:	1	
c	)	10b	TFA	3.0 ml	10
25		(0.300)	20 °C	1.5 h	(0.262)

Compound 10a:  $\left[\alpha\right]_{D}^{24}$  - 5.1 (c = 0.57, in chloroform)

Elemental Analysis for C<sub>72</sub>H<sub>115</sub>N<sub>5</sub>O<sub>14</sub>S•H<sub>2</sub>O:

Calcd. C, 66.18; H, 8.87; N, 5.36; S, 2.45

Found C, 66.07; H, 8.94; N, 5.48; S, 2.49

Compound 10b:  $[\alpha]_D^{21}$  - 10.7 • (c = 0.57, in chloroform)

Elemental Analysis for C<sub>57</sub> H<sub>105</sub> N<sub>5</sub> O<sub>12</sub> S:

Calcd. C, 63.13; H, 9.76; N, 6.46; S, 2.96

Found C, 63.13; H, 9.51; N, 6.46; S, 2.96

Compound 10:  $[\alpha]_{D}^{21} + 13.2^{\circ}$  (c = 0.67 in 5% TFA-chloroform)

Elemental Analysis for C<sub>49</sub>H<sub>89</sub>N<sub>5</sub>O<sub>12</sub>S•2H<sub>2</sub>O:

Calcd. C, 58.36; H, 9.29; N, 6.95; S, 3.18

Found C, 58.30; H, 8.90; N, 6.76; S, 3.25

Example 16

Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu-Gly-D-Glu-OH (Compound 11)

In substantially the same manner as in Example 6, (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-Glu-(O¹Bu)-Gly-D-Glu(O¹Bu)-O¹Bu (11a), (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu(O¹Bu)-Gly-D-Glu(O¹Bu)-O¹Bu (11b) and (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu-Gly-D-Glu-OH (Compound 11) were produced.

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		Materials (g)	Reaction Co	nditions	Products (g)
	a)	GC-2	P-6	1.31 g	11a
5		(2.04)	HONB	453 mg	(2.72)
			DIC	0.40 ml	
10					
			DMF	20 ml	
			20 °C	16 h	
	b)	11a	piperidine	2.0 ml	11b
15		(2.00)	DCM	18 ml	(1.75)
			20 °C	1 h	
			silica-gel		
20			(ethylaceta	te-methar	nol)
			10:0, 9:1		
	c)	11b	TFA	8.0 ml	
25		(0.80)	20 °C	2 h	(0.56)

Compound 11a:  $[\alpha]_{D}^{23}$  - 7.0 ° (c = 0.56, in chloroform)

Elemental Analysis for C<sub>77</sub>H<sub>124</sub>N<sub>4</sub>O<sub>15</sub>S:

Calcd. C, 67.12; H, 9.07; N, 4.07; S, 2.33
Found C, 67.01; H, 9.19; N, 4.01; S, 2.31

Compound 11b:  $[\alpha]_0^{23}$  - 18° (c = 0.53, in chloroform)

Elemental Analysis for C<sub>62</sub>H<sub>114</sub>N<sub>4</sub>O<sub>13</sub>S:

Calcd. C, 64.44; H, 9.94; N, 4.85; s, 2.77

Found C, 64.23; H, 9.95; N, 4.94; S, 2.64

Compound 11:  $[\alpha]_D^{21} + 2.1^{\circ}$  (c = 0.53 in 5% TFA-chloroform)

Elemental Analysis for C<sub>50</sub> H<sub>90</sub> N<sub>4</sub> O<sub>13</sub> S • H<sub>2</sub> O:

Calcd. C, 59.73; H, 9.22; N, 5.57; s, 3.19

Found C, 59.56; H, 9.05; N, 5.48; s, 3.15

Example 17

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Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Glu-OH (Compound 12)

In substantially the same manner as in Example 6, (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Glu(O¹Bu)-O¹Bu (12a), (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Glu(O¹Bu)-O¹Bu (12b) and (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Glu-OH (Compound 12) were produced.

		Materials (g)	Reacti	on Conditions	Products (g)
	a)	GC-2	P-7	918 mg	12a
5		(2.00)	HONB	441 mg	(1.81)
			DIC	0.39 ml	
10			DMF	20 ml	
			20 °C	15 h	
	b)	12a	piperio	dine 1.8 ml	12b
15		(1.75)	DCM	18 ml	(1.32)
			20 °C	1.5 h	
			silica-	-gel	
			(chlore	form-methano	1)
20			50:1, 2	20:1	
	c)	12b	TFA	5 ml	12
		(0.60)	20 °C	2 h	(0.52)
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Compound 12a:  $[\alpha]_D^{23}$  - 5.2 • (c = 0.58, in chloroform)

Elemental Analysis for C <sub>70</sub> H <sub>112</sub> N <sub>4</sub> O <sub>13</sub> S:						
	Calcd. Found	C, 67.28; C, 67.22;	H, 9.03; H, 8.84;	N, 4.48; N, 4.55;		

Compound 12b:  $[\alpha]_{D}^{23}$  - 7.9 ° (c = 0.60, in chloroform)

Elemental Analysis for C <sub>55</sub> H <sub>102</sub> N <sub>4</sub> O <sub>11</sub> S•0.5H <sub>2</sub> O:						
Calcd.	C, 63.73;	H, 10.02;	N, 5.41;	S, 3.09		
Found	C, 63.88;	H, 10.22;	N, 5.48;	S, 3.09		

Compound 12:  $[\alpha]_{D}^{23} + 14.8^{\circ}$  (c = 0.68 in 5% TFA-chloroform)

Ele	Elemental Analysis for C <sub>47</sub> H <sub>86</sub> N <sub>4</sub> O <sub>11</sub> S•2.5H <sub>2</sub> O:							
Cald		C, 58.78;	H, 9.55;	N, 5.83;	S, 3.34			
Fou		C, 58.91;	H, 8.83;	N, 5.67;	S, 3.06			

## Example 18

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Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-OH (Compound 13)

In substantially the same manner as in Example 6, (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-O¹Bu (13a), (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-O¹Bu (13b) and (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-OH (Compound 13) were produced.

Materials (g) Reaction Conditions Products (g) GC-2 H-Gly-Gly-Gly-O<sup>t</sup>Bu a) 660 mg 13a (2.18)HONB 482 mg (2.50)5 DIC 0.42 ml .10 DMF 20 ml 20 °C 17 h b) 13a piperidine 1.8 ml 13b (1.80)DCM 15 18 ml (1.30)20°C 1.3 h silica-qel (chloroform-methanol) 20 50:1, 20:1 C) 13b TFA 5 ml 13 (0.60)20 °C 2 h (0.55)

Compound 13a:  $[\alpha]_{D}^{23}$  - 11.0 • (c = 0.60, in chloroform)

Elemental Analysis for C<sub>63</sub>H<sub>100</sub>N<sub>4</sub>O<sub>11</sub>S:

Calcd. C, 67.47; H, 8.99; N, 5.00; S, 2.86
Found C, 67.47; H, 8.86; N, 4.92; S, 2.89

Compound 13b:  $[\alpha]_{D}^{23}$  - 14.3 • (c = 0.48, in chloroform)

Elemental Analysis for C<sub>48</sub>H<sub>90</sub>N<sub>4</sub>O<sub>9</sub>S:

Calcd. C, 64.11; H, 10.09; N, 6.23; S, 3.57
Found C, 63.97; H, 10.01; N, 6.21; S, 3.44

Compound 13:  $[\alpha]_{10}^{123}$  + 12.2 • (c = 0.63 in 5% TFA-chloroform)

Elemental Analysis for C44H82N4O9S•2.5H2O:

Calcd. C, 59.50; H, 9.87; N, 6.31; S, 3.61
Found C, 59.21; H, 9.17; N, 6.16; S, 3.34

Example 19

Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Glu-OH (Compound 14)

In substantially the same manner as in Example 6, (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-Gly-Glu(O¹Bu)-O¹Bu (14a), (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Glu(O¹Bu)-O¹Bu (14b) and (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Glu-OH (Compound 14) were produced.

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		Materials (g)	Reaction Con	ditions Produc	cts (g)
	a)	GC-2 H-Gly-C	Glu(O <sup>t</sup> Bu)-O <sup>t</sup> B	u 777 mg	14a
5		(2.03)	HONB 4	41 mg	(2.20)
10			DIC 0	.39 ml	
			DMF	20 ml	
			20 °C	16 h	
15	b)	14a	piperidine	1.4 ml	14b
73		(1.40)	DCM	14 ml	(0.90)
			20 °C	1.0 h	
			silica-gel		
20			(chloroform	-methanol)	
			50:1, 20:1		
	c)	14b	TFA	4.0 ml	14
25		(0.40)	20 °C	2 h	(0.29)

Compound 14a:  $[\alpha]_0^{23}$  - 2.0 ° (c = 0.50, in chloroform)

Elemental Analysis for C<sub>68</sub> H<sub>109</sub> N<sub>3</sub> O<sub>12</sub> S:

Calcd. C, 68.48; H, 9.21; N, 3.52; S, 2.69
Found C, 68.62; H, 9.26; N, 3.60; S, 2.68

Compound 14b:  $[\alpha]_D^{23}$  - 5.6 ° (c = 0.57, in chloroform)

Elemental Analysis for C<sub>53</sub>H<sub>99</sub>N<sub>3</sub>O<sub>10</sub>S:

Calcd. C, 65.60; H, 10.28; N, 4.33; S, 3.30
Found C, 65.51; H, 10.31; N, 4.20; S, 3.25

Compound 14:  $[\alpha]_0^{21} + 2.1^{\circ}$  (c = 0.53 in 5% TFA-chloroform)

Elemental Analysis for C<sub>45</sub>H<sub>83</sub>N<sub>3</sub>O<sub>10</sub>S•0.5H<sub>2</sub>O:

Calcd. C, 62.32; H, 9.76; N, 4.85; S, 3.70
Found C, 62.16; H, 9.64; N, 4.61; S, 3.67

Example 20

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Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu-OH (Compound 15)

In substantially the same manner as in Example 6, (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-Glu-(O'Bu)-O'Bu (15a), (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu(O'Bu)-O'Bu (15b) and (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu-OH (Compound 15) were produced.

Materials (g) Reaction Conditions Products (g)
a) GC-2  $H-Glu(O^{t}Bu)-O^{t}Bu$  1.27 g 15a (4.00) HONB 882 mg (4.80)

0.77 ml DIC 10 DMF 30 ml 20 °C 17 h b) 15a piperidine 2.3 ml 15b 15 (2.30)DCM 23 ml (1.14)20 °C 1.3 h silica-gel 20 (chloroform-methanol) 50:1, 20:1 C) 15b **TFA** 5 ml 15 (0.48)20 °C 2 h (0.41)25

Compound 15a:  $[\alpha]_{D}^{23}$  - 2.0° (c = 0.58, in chloroform)

Elemental Analysis for C<sub>66</sub> H<sub>106</sub> N<sub>2</sub> O<sub>11</sub> S:

Calcd. C, 69.81; H, 9.41; N, 2.47; S, 2.82
Found C, 69.79; H, 9.26; N, 2.44; S, 2.74

Compound 15b:  $[\alpha]_D^{23}$  - 10.1 • (c = 0.56, in chloroform)

Elemental Analysis for C<sub>5 1</sub> H<sub>96</sub> N<sub>2</sub>O<sub>9</sub>S:

Calcd. C, 67.06; H, 10.59; N, 3.07; S, 3.51

Found C, 67.04; H, 10.36; N, 3.09; S, 3.45

Compound 15:  $[\alpha]_D^{23} + 3.7^{\circ}$  (c = 0.60 in 5% TFA-chloroform)

Elemental Analysis for C<sub>4.3</sub>H<sub>8.0</sub>N<sub>2</sub>O<sub>9</sub>S • 0.5H<sub>2</sub>O:

Calcd. C, 63.75; H, 10.08; N, 3.46; S, 3.96

Found C, 63.70; H, 9.75; N, 3.31; S, 3.88

## Example 21

Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-Asp-OH (Compound 16)

In substantially the same manner as in Example 6, (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-Asp(O<sup>t</sup>Bu)-O<sup>t</sup>Bu (16a), (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-Gly-Asp-OH (Compound 16) were produced.

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		Materials	(g)	React	ion (	Condition	s Pr	oducts	(g)
	a)	GC-2		P-8		1.17 g		16	a
5									
		(2.28)		HONB		502 mg		(3.	13)
10				DIC		0.44 ml			
				DMF		20 ml			
				20 °C		15 h			
	þ)	16a		piperi	ldine	3.0 ml		16b	)
15		(2.94)		DCM		30 ml		(1.	83)
				20 °C		2 h			
				silica	-gel	•			
20				(chlor	ofor	m-metha	nol)		
				50:1,	20:1				
	c)	16b		TFA		9 ml		16	
25		(0.83)		20 °C		2 h		(0.	74)

Compound 16a:  $\left[\alpha\right]_{D}^{22}$  + 3.7° (c = 0.61, in chloroform)

Elemental Analysis for C <sub>71</sub> H <sub>113</sub> N <sub>5</sub> O <sub>14</sub> S:					
Calcd. Found	C, 65.97; C, 65.61;	, ,		S, 2.48 S, 2.24	

Compound 16b:  $[\alpha]_{0}^{22}$  - 1.5 • (c = 0.55, in chloroform)

Element	Elemental Analysis for C <sub>56</sub> H <sub>103</sub> N <sub>5</sub> O <sub>12</sub> S:						
Calcd.	C, 62.83;	H, 9.70;		S, 3.00			
Found	C, 62.77;	H, 9.92;		S, 2.94			

Compound 16:  $[\alpha]_{D}^{25} + 19.2 \cdot (c = 0.64 \text{ in } 5\% \text{ TFA-chloroform})$ 

15	Elementa	Elemental Analysis for C <sub>48</sub> H <sub>87</sub> N <sub>5</sub> O <sub>12</sub> S•1.5H <sub>2</sub> O:						
	Calcd. Found	C, 58.51; C, 58.68;	Н, 9.21; Н, 9.00;	N, 6.87; N, 6.87;	S, 3.25 S, 3.25			

50 Example 22

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Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-D-Glu-OH (Compound 17)

In substantially the same manner as in Example 6, (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-Gly-Gly-D-Glu(O¹Bu)-O¹Bu (17a), (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-D-Glu(O¹Bu)-O¹Bu (17b) and (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-D-Glu-OH (Compound 17) were produced.

Materials (g) Reaction Conditions Products (g) 5 17a 780 mg a ) GC-2 P-9374 mg (1.97)(1.68)HONB DIC  $327 \mu l$ 10 . DMF 18 ml 20 °C 16 h 17b piperidine 2.0 ml b) 17a 15 DCM 18 ml (1.52)(1.80)20 °C 1 h silica-qel (chloroform-methanol) 20 25:1 TFA 10.0 ml 17 C) 17b (0.76)20 °C 3 h (1.00)25

Compound 17a:  $[\alpha]_D^{24}$  - 10.0 ° (c = 0.54, in chloroform)

Elemental Analysis for C<sub>70</sub>H<sub>112</sub>N<sub>4</sub>O<sub>13</sub>S•0.5H<sub>2</sub>O:

Calcd. C, 66.79; H, 9.05; N, 4.45; S, 2.55

Found C, 66.59; H, 8.75; N, 4.75; S, 2.53

Compound 17b:  $[\alpha]_{D}^{24}$  - 15.7 • (c = 0.49, in chloroform)

Elemental Analysis for C<sub>55</sub>H<sub>102</sub>N<sub>4</sub>O<sub>11</sub>S•0.5H<sub>2</sub>O:

Calcd. C, 63.73; H, 10.02; N, 5.41; S, 3.09
Found C, 63.67; H, 10.23; N, 5.28; S, 3.07

Compound 17:  $[\alpha]_D^{24} + 6.0^{\circ}$  (c = 0.51, in 5% TFA-chloroform)

Elemental Analysis for C<sub>47</sub>H<sub>86</sub>N<sub>4</sub>O<sub>11</sub>S•2H<sub>2</sub>O:

Calcd. C, 59.34; H, 9.54; N, 5.89; S, 3.37

Found C, 59.01; H, 9.17; N, 5.92; S, 3.08

Example 23

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Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-OH (Compound 18)

In substantially the same manner as in Example 6, (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-Gly-Gly-O¹Bu (18a), (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Gly-O¹Bu (18b) and (2R,6R)-2-amino-6,7-bis-(PamO)-4-THT-Gly-Gly-OH (Compound 18) were produced.

Materials (g) Reaction Conditions Products (g)

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	a)	GC-2	H-Gly-Gly-0 <sup>t</sup> Bu	580 mg	18a
		(2.50)	HONB	552 mg	(2.42)
			DIC	482 µl	
10			DMF	20 ml	
			20 °C	13 h	
	b)	18a	piperidi	ine 2.2 ml	18b
15		(2.20)	DCM	22 ml	(1.59)
			20 °C	1.5 h	
			silica-c	gel	
			(chlorof	orm-methanol)	
20			20:1	·	
	c)	18b	TFA	6.0 ml	18
		(0.60)	20 °C	1 h	(0.56)

Compound 18a:  $[\alpha]_D^{24}$  - 7.6 • (c = 0.52 in chloroform)

Elemental Analysis for C <sub>61</sub> H <sub>97</sub> N <sub>3</sub> O <sub>10</sub> S:								
	C, 68.83; C, 69.06;	1						

Compound 18b:  $[\alpha]_D^{24}$  - 13.4° (c = 0.52, in chloroform)

Elemental Analysis for C <sub>46</sub> H <sub>87</sub> N <sub>3</sub> O <sub>8</sub> S•0.5H <sub>2</sub> O:							
Calcd.	C, 64.90;	H, 10.42;	N, 4.94;	S, 3.77			
Found	C, 64.99;	H, 10.45;	N, 4.82;	S, 3.49			

Compound 18:  $[\alpha]_0^{25} + 10.7^{\circ}$  (c = 0.51 in 5% TFA-chloroform)

Elemental Analysis for C <sub>42</sub> H <sub>79</sub> N <sub>3</sub> O <sub>8</sub> S•1.5H <sub>2</sub> O:							
Calcd.	C, 62.03;	Н, 10.16;	N, 5.17;	S, 3.94			
Found	C, 61.90;	Н, 9.97;	N, 4.85;	S, 4.04			

Example 24

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Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-D-Glu-OH (Compound 19)

In substantially the same manner as in Example 6, (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-Gly-D-Glu(O'Bu)-O'Bu (19a), (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-D-Glu(O'Bu)-O'Bu (19b) and (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-D-Glu-OH (Compound 19) were produced.

		Materials (g)	Reaction Co	nditions	Products (g)
	a)	GC-2	P-10	830 mg	19a
5	·	(2.12)	HONB	471 mg	(2.18)
			DIC	412 µl	
			DMF	21 ml	
10 .			20 °C	16 h	
	b)	19a	piperidine	2.0 ml	19b
		(2.00)	DCM	18 ml	(1.72)
			20 °C	1 h	
15			silica-gel		
			(chlorofor	m-methanol	)
			97:3		
20	C)	19b	TFA	10 ml	19
		(1.00)	20 °C	3 h	(0.70)

Compound 19a:  $[\alpha]_{D}^{24}$  - 12.5° (c = 0.55 in chloroform)

Elemental Analysis for C<sub>68</sub> H<sub>109</sub> N<sub>3</sub> O<sub>12</sub> S:

Calcd. C, 68.43; H, 9.21; N, 3.52; S, 2.69

Found C, 68.31; H, 9.25; N, 3.72; S, 2.48

Compound 19b:  $[\alpha]_D^{24}$  - 16.4° (c=0.53, in chloroform)

Elemental Analysis for C<sub>5.3</sub>H<sub>9.9</sub>N<sub>3</sub>O<sub>10</sub>S • 0.8H<sub>2</sub>O:

Calcd. C, 64.64; H, 10.30; N, 4.27; S, 3.26

Found C, 64.62; H, 10.25; N, 4.08; S, 3.22

Compound 19:  $[\alpha]_D^{24}$  + 14.2° (c = 0.52 in 5% TFA-chloroform)

Elemental Analysis for C<sub>45</sub>H<sub>83</sub>N<sub>3</sub>O<sub>10</sub>S•2.2H<sub>2</sub>O:

Calcd. C, 60.19; H, 9.81; N, 4.68; S, 3.57

Found C, 60.05; H, 9.44; N, 4.35; S, 3.61

Example 25

Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-D-Glu-OH (Compound 20)

In substantially the same manner as in Example 6, (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-D-Glu-(O¹Bu)-O¹Bu (20a), (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-D-Glu(O¹Bu)-O¹Bu (20b) and (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-D-Glu-OH (Compound 20) were produced.

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		Materials (g)	Reaction Conditi	ons Produ	cts (g)
	a)	GC-2 H-D-G1	u (O <sup>t</sup> Bu) -O <sup>t</sup> Bu	0.75 g	20a
5		(2.35)	HONB	518 mg	(2.35)
			DIC	0.45 ml	
			DMF	20 ml	
10			20 °C	17 h	
	b)	20a	piperidine	2.0 ml	20b
		(2.23)	DCM	20 ml	(1.76)
15			20 °C		
15	-		silica-gel		
			(hexane-ethylad	cetate)	
			4:1, 3:2		
20	c)	20ъ	TFA	10 ml	20
		(0.77)	20 °C	2 h	(0.64)

Compound 20a:  $[\alpha]_0^{22}$  - 7.4 • (c = 0.60 in chloroform)

Elemental Analysis for C<sub>66</sub> H<sub>106</sub> N<sub>2</sub> O<sub>11</sub> S:

Calcd. C, 69.81; H, 9.41; N, 2.47; S, 2.82

Found C, 69.66; H, 9.52; N, 2.85; S, 2.53

Compound 20b:  $[\alpha]_D^{22}$  - 15.9 ° (c = 0.66, in chloroform)

Elemental Analysis for C<sub>51</sub>H<sub>96</sub>N<sub>2</sub>O<sub>9</sub>S•0.5H<sub>2</sub>O:

Calcd. C, 66.41; H, 10.60; N, 3.04; S, 3.48

Found C, 66.42; H, 10.42; N, 3.25; S, 3.21

Compound 20:  $[\alpha]_0^{25}$  - 1.5° (c = 0.55 in 5% TFA-chloroform)

Elemental Analysis for C<sub>43</sub>H<sub>80</sub>N<sub>2</sub>O<sub>9</sub>S:

Calcd. C, 64.46; H, 10.06; N, 3.50; S, 4.00
Found C, 64.56; H, 9.92; N, 3.42; S, 3.90

Example 26

Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Asp-OH (Compound 21)

In substantially the same manner as in Example 6, (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-Asp-(O¹Bu)-O¹Bu (21a), (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Asp(O¹Bu)-O¹Bu (21b) and (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Asp-OH (Compound 21) were produced.

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		Materia	ls (g)	Reaction	Conditions	Products (g)
	a)	GC-2	H-Asp(	O <sup>t</sup> Bu ) –O <sup>t</sup> Bu	0.69 g	21a
5		(2.30)		HONB	507 mg	(2.67)
				DIC	0.44 ml	
				DMF	20 ml	
10 .				20 °C	17 h	
	b)	21a		piperidi	ne 2.5 ml	21b
		(2.56)		DCM	25 ml	(1.80)
				20 °C	1 h	
15				silica-g	el	
				(hexane-	ethyl acetat	ce)
				4:1, 3:2		
20	c),	21b		TFA	10 ml	21
		(0.86)		20 °C	2 h	(0.73)

Compound 21a:  $[\alpha]_D^{22} + 5.2 \cdot (c = 0.65 \text{ in chloroform})$ 

Elemental Analysis for C<sub>65</sub> H<sub>104</sub> N<sub>2</sub> O<sub>11</sub> S:

Calcd. C, 69.61; H, 9.35; N, 2.50; S, 2.86

Found C, 69.38; H, 9.49; N, 2.77; S, 2.69

Compound 21b:  $[\alpha]_D^{22}$  - 1.1 • (c=0.70, in chloroform)

Elemental Analysis for C<sub>50</sub>H<sub>94</sub>N<sub>2</sub>O<sub>9</sub>S:

Calcd. C, 66.76; H, 10.54; N, 3.11; S, 3.57

Found C, 66.87; H, 10.55; N, 3.36; S, 3.30

Compound 21:  $[\alpha]_D^{25}$  + 5.5 • (c = 0.63 in 5% TFA-chloroform)

Elemental Analysis for C<sub>42</sub>H<sub>78</sub>N<sub>2</sub>O<sub>9</sub>S:

Calcd. C, 64.09; H, 9.99; N, 3.56; S, 4.07
Found C, 64.23; H, 9.74; N, 3.44; S, 4.02

Example 27

Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-D-Asp-OH (Compound 22)

In substantially the same manner as in Example 6, (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-D-Asp-(O¹Bu)-O¹Bu (22a), (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-D-Asp-OH (Compound 22) were produced.

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		Materia	ls (g)	Reaction	n Conditi	ons	Produc	cts (g)
	a)	GC-2	H-D-As	o (O <sup>t</sup> Bu) - (	) <sup>t</sup> Bu	0.70	g	22a
5		(2.32)		HONB	511 n	ng		(2.84)
				DIC	0.45	ml		
				DMF	20	ml		
10				20 °C	17	h		
	b)	22a		piperid.	ine 2.5	ml		22b
		(2.64)	٠	DCM	25	ml		(1.87)
46				20 °C	1	h		
15				silica-	gel			
				(hexane	ethylac	cetate	∍)	
				4:1, 3:	2			
20	c)	22b		TFA	10	ml		22
		(0.95)		20 °C	2	h		(0.82)

Compound 22a:  $[\alpha]_0^{22}$  - 12.7° (c = 0.51 in chloroform)

Elemental Analysis for C<sub>65</sub> H<sub>104</sub> N<sub>2</sub> O<sub>11</sub> S:

Calcd. C, 69.61; H, 9.35; N, 2.50; S, 2.86
Found C, 69.82; H, 9.48; N, 2.98; S, 2.53

Compound 22b:  $[\alpha]_{D}^{22}$  - 22.7 ° (c = 0.67, in chloroform)

Elemental Analysis for C<sub>50</sub>H<sub>94</sub>N<sub>2</sub>O<sub>9</sub>S•0.5H<sub>2</sub>O:

Calcd. C, 66.11; H, 10.54; N, 3.08; S, 3.53

Found C, 65.83; H, 10.38; N, 2.93; S, 3.40

Compound 22:  $[\alpha]_{b}^{25}$  - 10.8° (c = 0.62 in 5% TFA-chloroform)

Elemental Analysis for C<sub>4.2</sub>H<sub>78</sub> N<sub>2</sub>O<sub>9</sub>S:

Calcd. C, 64.09; H, 9.99; N, 3.56; S, 4.07

Found C, 64.13; H, 9.60; N, 3.47; S, 4.06

Example 28

Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Glu-OH (Compound 23)

In substantially the same manner as in Example 6, (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-Gly-Glu(O¹Bu)-Glu(O¹Bu)-O¹Bu (23a), (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Glu(O¹Bu)-Glu(O¹Bu)-O¹Bu (23b) and (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Glu-OH (Compound 23) were produced.

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		Materials (g)	Reaction Co	nditions	Products (g)
	a)	GC-2	P-11	590 mg	23a
5		(0.95)	HONB	210 mg	(1.22)
			DIC	185 µl	
			DMF	9.5 ml	
10			20 °C	16 h	
	b)	23a	piperidine	1.1 ml	23b
		(1.10)	DCM	9.9 ml	(0.97)
			20 °C	1 h	
15			silica-gel		
			(chlorofor	m-methano	1)
			97:3		
20	c)	23b	TFA	4.8 ml	23
		(0.48)	20 °C	3 h	(0.36)

Compound 23a:  $[\alpha]_D^{24}$  - 7.0 ° (c = 0.50 in chloroform)

Elemental Analysis for C<sub>77</sub>H<sub>124</sub>N<sub>4</sub>O<sub>15</sub>S:

Calcd. C, 67.12; H, 9.07; N, 4.07; S, 2.33

Found C, 67.16; H, 9.05; N, 4.10; S, 2.40

Compound 23b:  $[\alpha]_{D}^{24}$  - 10.6 ° (c = 0.50, in chloroform)

Elemental Analysis for C<sub>62</sub>H<sub>114</sub>N<sub>4</sub>O<sub>13</sub>S•0.5H<sub>2</sub>O:

Calcd. C, 63.94; H, 9.95; N, 4.81; S, 2.75

Found C, 63.91; H, 9.80; N, 4.74; S, 2.71

Compound 23:  $[\alpha]_D^{21} + 13.0 \circ (c = 0.52 \text{ in } 5\% \text{ TFA-chloroform})$ 

Elemental Analysis for C<sub>50</sub>H<sub>90</sub>N<sub>4</sub>O<sub>13</sub>S•1.5H<sub>2</sub>O:

Calcd. C, 59.20; H, 9.24; N, 5.52; S, 3.16
Found C, 59.15; H, 9.09; N, 5.50; S, 3.41

Example 29

Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Glu-D-Glu-OH (Compound 24)

In substantially the same manner as in Example 6, (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-Gly-Glu(O¹Bu)-D-Glu(O¹Bu)-O¹Bu(24a), (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Glu(O¹Bu)-D-Glu(O¹Bu)-D-Glu(O¹Bu)-D-Glu(O¹Bu)-A-THT-Gly-Glu-D-Glu-OH (Compound 24) were produced.

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		Materials (g)	Reaction Con	ditions Pro	oducts (g)
	<b>a</b> )	GC-2	P-12 5	00 mg	24a
5		(0.81)	HONB 1	78 mg	(0.98)
			DIC 1	57 µl	
			DMF	8.0 ml	
10			20 °C	16 h	
	b)	24a	piperidine	0.9 ml	24b
		(0.90)	DCM	8.1 ml	(0.80)
			20 °C	1 h	
15		•	silica-gel		
			(chloroform-	-methanol)	
			97:3		
20	c)	24b	TFA	4.0 ml	24
		(0.40)	20 °C	3 h	(0.29)

Compound 24a:  $[\alpha]_D^{24}$  - 12.3 ° (c = 0.52 in chloroform)

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Elemental Analysis for C <sub>77</sub> H <sub>124</sub> N <sub>4</sub> O <sub>15</sub> S•0.5H <sub>2</sub> O:					
Calcd.	C, 66.68;	H, 9.08;	N, 4.04;	S, 2.31	
Found	C, 66.82;	H, 8.83;	N, 4.34;	S, 2.33	

Compound 24b:  $[\alpha]_{D}^{24}$  - 15.9 • (c = 0.49, in chloroform)

Elemental Analysis for C<sub>62</sub>H<sub>114</sub>N<sub>4</sub>O<sub>13</sub>S • 0.5H<sub>2</sub>O:

Calcd. C, 63.94; H, 9.95; N, 4.81; S, 2.75
Found C, 64.04; H, 9.84; N, 4.86; S, 2.76

Compound 24:  $[\alpha]_D^{24} + 5.0 \cdot (c = 0.47 \text{ in } 5\% \text{ TFA-chloroform})$ 

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Elemental Analysis for C <sub>50</sub> H <sub>90</sub> N <sub>4</sub> O <sub>13</sub> S•H <sub>2</sub> O:					
Calcd.	C, 59.73;	H, 9.22;	N, 5.57;	S, 3.19	
Found	C, 59.73;	H, 9.25;	N, 5.48;	S, 3.18	

Example 30

Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu-Gly-Glu-OH (Compound 25)

In substantially the same manner as in Example 6, (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-Glu-(O¹Bu)-Gly-Glu(O¹Bu)-O¹Bu (25a), (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu(O¹Bu)-O¹Bu (25b) and (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu-Gly-Glu-OH (Compound 25) were produced.

		Materials (g)	Reaction Co	onditions	Products (g)
	a)	GC-2	P-13	770 mg	25a
_		(1.24)	HONB	274 mg	(1.63)
5			DIC	241 μ1	
			DMF	12.0 ml	
			20 °C	16 h	
10	b)	25a	piperidine	e 1.5 ml	25b
		(1.50)	DCM	13.5 ml	(1.34)
			20 °C	1 h	
15			silica-gel	-	
			(chlorofor	m-methano	L)
			97:3		
	c)	25b	TFA	6.7 ml	25
20		(0.67)	20 °C	3 h	(0.50)

Compound 25a:  $[\alpha]_D^{24}$  - 1.7 • (c = 0.51 in chloroform)

Elemental Analysis for C<sub>77</sub> H<sub>124</sub> N<sub>4</sub> O<sub>15</sub> S:

Calcd. C, 67.12; H, 9.07; N, 4.07; S, 2.33

Found C, 66.97; H, 9.12; N, 4.06; S, 2.25

30 Compound 25b:  $[\alpha]_D^{24}$  - 9.6 • (c = 0.53, in chloroform)

Elemental Analysis for C<sub>62</sub>H<sub>114</sub>N<sub>4</sub>O<sub>13</sub>S•H<sub>2</sub>O:

Calcd. C, 63.45; H, 9.96; N, 4.77; S, 2.73

Found C, 63.31; H, 9.81; N, 4.82; S, 2.65

Compound 25:  $[\alpha]_D^{24}$  + 8.7° (c = 0.52 in 5% TFA-chloroform)

Elemental Analysis for C<sub>50</sub>H<sub>90</sub>N<sub>4</sub>O<sub>13</sub>S•1.5H<sub>2</sub>O:

Calcd. C, 59.20; H, 9.24; N, 5.52; S, 3.16
Found C, 59.34; H, 9.13; N, 5.53; S, 3.21

Example 31

Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu-Glu-OH (Compound 26)

In substantially the same manner as in Example 6, (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-Glu-(O¹Bu)-Glu(O¹Bu)-O¹Bu (26a), (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu(O¹Bu)-Glu(O¹Bu)-O¹Bu (26b) and (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu-Glu-OH (Compound 26) were produced.

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		Material	Ls (g)	Reaction	Con	ditions	Produc	cts	(g)
	a)	GC-2	H-Glu(	O <sup>t</sup> Bu)-Glu	(O <sup>t</sup> B	u)-O <sup>t</sup> Bu		26	a.
5		(1.20)			6	60 mg		(1.	63)
· ·				HONB	2	65 mg			
				DIC	2	33 μl			
				DMF		12.0 m	L		
10				20 °C		16 h			
	b)	26a		piperidi	ne	1.5 ml		26t	)
		(1.50)		DCM		13.5 ml	L	(1.	32)
15				20 °C		1 h			
				silica-g	el				
				(chlorof	orm.	-methano	)		
20				97:3					
	c)	26b		TFA		6.6 ml	-	26	
		(0.66)		20 °C		3 h		(0.	47)

25 Compound 26a:  $[\alpha]_{0}^{24}$  - 7.4 • (c = 0.51 in chloroform)

Elemental Analysis for C <sub>75</sub> H <sub>121</sub> N <sub>3</sub> O <sub>14</sub> S:					
Calcd.	C, 68.20;		N, 3.18;	S, 2.43	
Found	C, 68.26;		N, 2.95;	S, 2.57	

Compound 26b:  $[\alpha]_D^{24}$  - 17.7° (c = 0.53, in chloroform)

Elemental Analysis for C <sub>60</sub> H <sub>111</sub> N <sub>3</sub> O <sub>12</sub> S•0.5H <sub>2</sub> O:					
Calcd.	C, 65.06;	H, 10.19;	N, 3.79;	S, 2.89	
Found	C, 64.88;	H, 10.29;	N, 3.81;	S, 2.95	

40 Compound 26:  $[\alpha]_{b}^{24}$  - 0.9 ° (c = 0.52 in 5% TFA-chloroform)

Elemental Analysis for C <sub>48</sub> H <sub>87</sub> N <sub>3</sub> O <sub>12</sub> S•H <sub>2</sub> O:					
Calcd.	C, 60.79;	H, 9.46;	N, 4.43;	S, 3.38	
Found	C, 60.89;	H, 9.40;	N, 4.41;	S, 3.38	

Example 32

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Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu-D-Glu-OH (Compound 27)

In substantially the same manner as in Example 6, (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-Glu-(O¹Bu)-D-Glu(O¹Bu)-O¹Bu(27a), (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu(O¹Bu)-D-Glu(O¹Bu)-O¹Bu(27b) and (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu-D-Glu-OH (Compound 27) were produced.

		Materials (g)	Reaction Con	ditions Prod	ucts (g)
	a)	GC-2	P-14	700 mg	27a
5		(1.27)	HONB	282 mg	(1.45)
			DIC	247 µl	
			DMF	12.7 ml	
10			20 °C	16 h	
	b)	27a	piperidine	1.4 ml	27b
		(1.33)	DCM	12.6 ml	(1.12)
			20 °C	1 h	
15			silica-gel		
			(chloroform	-methanol)	
			97:3		
20	C)	27b	TFA	5.5 ml	27
		(0.55)	20 °C	3 h	(0.45)

Compound 27a:  $[\alpha]_{D}^{24}$  - 12.6 ° (c = 0.52 in chloroform)

Elemental Analysis for C<sub>75</sub>H<sub>121</sub>N<sub>3</sub>O<sub>14</sub>S:

Calcd. C, 68.20; H, 9.23; N, 3.18; S, 2.43

Found C, 68.27; H, 9.46; N, 3.14; S, 2.30

Compound 27b:  $[\alpha]_0^{24}$  - 21.3 • (c = 0.51, in chloroform)

Elemental Analysis for C<sub>60</sub>H<sub>111</sub>N<sub>3</sub>O<sub>12</sub>S•0.5H<sub>2</sub>O:

Calcd. C, 65.06; H, 10.19; N, 3.79; S, 2.89
Found C, 65.05; H, 10.33; N, 3.65; S, 2.88

Compound 27:  $[\alpha]_D^{24}$  - 5.9 ° (c = 0.53 in 5% TFA-chloroform)

Elemental Analysis for C<sub>48</sub>H<sub>87</sub>N<sub>3</sub>O<sub>12</sub>S•H<sub>2</sub>O:

Calcd. C, 60.79; H, 9.46; N, 4.43; S, 3.38
Found C, 60.88; H, 9.25; N, 4.43; S, 3.19

Example 33

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Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu-Glu-Glu-OH (Compound 28)

In substantially the same manner as in Example 6, (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-Glu-(O¹Bu)-Glu(O¹Bu)-Glu(O¹Bu)-O¹Bu (28a), (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu(O¹Bu)-Glu(O¹Bu)-Glu(O¹Bu)-O¹Bu (2Bb) and (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu-Glu-Glu-OH (Compound 28) were produced.

		Materials (g)	Reaction Condi	itions Produ	cts (g)
	a)	GC-2	P-15 71	5 mg	28a
5		(0.91)	HONB 20	1 mg	(1.48)
			DIC 17	7 μl	
			DMF	9.1 ml	
10			20 °C	16 h	
	b)	28a	piperidine	1.3 ml	28b
		(1.35)	DCM	11.7 ml	(1.20)
			20 °C	1 h	
15			silica-gel	•	
			(chloroform-	methanol)	
			97:3		
20	c)	28b	TFA	6.0 ml	28
		(0.60)	20 °C	3 h	(0.44)

Compound 28a:  $[\alpha]_D^{24}$  - 13.2 ° (c = 0.52 in chloroform)

Elemental Analysis for C<sub>84</sub> H<sub>136</sub> N<sub>4</sub> O<sub>17</sub> S:

Calcd. C, 66.99; H, 9.10; N, 3.72; S, 2.18

Found C, 67.08; H, 9.14; N, 3.78; S, 2.16

Compound 28b:  $[\alpha]_{D}^{24}$  - 19.5 ° (c = 0.53, in chloroform)

Elemental Analysis for C<sub>69</sub> H<sub>126</sub> N<sub>4</sub> O<sub>15</sub> S • 0.5H<sub>2</sub> O:

Calcd. C, 64.10; H, 9.90; N, 4.33; S, 2.48

Found C, 64.28; H, 9.92; N, 4.21; S, 2.47

Compound 28:  $[\alpha]_{6}^{24}$  - 5.1 ° (c = 0.53 in 5% TFA-chloroform)

Elemental Analysis for C<sub>5 3</sub> H<sub>94</sub> N<sub>4</sub> O<sub>15</sub> S • 0.5 H<sub>2</sub> O:

Calcd. C, 59.58; H, 8.96; N, 5.24; S, 3.00

Found C, 59.36; H, 8.91; N, 5.20; S, 2.95

Example 34

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Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu-Glu-D-Glu-OH (Compound 29)

In substantially the same manner as in Example 6, (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-Glu-(O¹Bu)-Glu(O¹Bu)-D-Glu(O¹Bu)-O¹Bu (29a), (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu-O¹Bu)-Glu(O¹Bu)-O¹Bu (29b) and (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu-Glu-D-Glu-OH (Compound 29) were produced.

		Materials (g)	Reaction Con	ditions F	roducts (g)
	a)	GC-2	P-16 5	90 mg	29a
5		(0.75)	HONB 1	66 mg	(1.27)
			DIC 1	46 µl	
			DMF	7.5 ml	
			20 °C	16 h	
10	b)	29a	piperidine	1.2 ml	29ხ
		(1.15)	DCM	10.8 ml	(0.99)
			20 °C	1 h	
15			silica-gel		
			(chloroform-	-methanol)	
			97:3		
20	C)	29b	TFA	5.0 ml	29
20		(0.50)	20 °C	3 h	(0.36)

Compound 29a:  $[\alpha]_0^{24}$  - 12.7 • (c = 0.49 in chloroform)

Elemental Analysis for C<sub>84</sub> H<sub>136</sub> N<sub>4</sub> O<sub>17</sub> S:

Calcd. C, 66.99; H, 9.10; N, 3.72; S, 2.13

Found C, 66.80; H, 9.15; N, 3.65; S, 2.15

Compound 29b:  $[\alpha]_{D}^{24}$  - 20.7 ° (c = 0.51, in chloroform)

Elemental Analysis for C<sub>69</sub> H<sub>126</sub> N<sub>4</sub> O<sub>15</sub> S • H<sub>2</sub> O:

Calcd. C, 63.66; H, 9.91; N, 4.30; S, 2.46

Found C, 63.67; H, 9.88; N, 4.26; S, 2.48

Compound 29:  $[\alpha]_D^{24}$  - 16.1 ° (c = 0.52 in 5% TFA-chloroform)

Elemental Analysis for C<sub>53</sub>H<sub>94</sub>N<sub>4</sub>O<sub>15</sub>S+H<sub>2</sub>O:

Calcd. C, 59.08; H, 8.98; N, 5.20; S, 2.98

Found C, 58.85; H, 8.90; N, 5.17; S, 2.81

Example 35

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Production of (2R,6R)-2-Fmoc-amino-6-hydroxy-7-PamO-4-THT-O¹Bu (GC-5a), (2R,6R)-2-Fmoc-amino-6-hexanoyloxy-7-PamO-4-THT-O¹Bu (GC-5b) and (2R,6R)-2-Fmoc-amino-6-hexanoyloxy-7-PamO-4-THT-OH (GC-5)

In substantially the same manner as in Example 2, (2R,6R)-2-Fmoc-amino-6-hydroxy-7-PamO-4-THT-O'Bu (GC-5a), (2R,6R)-2-Fmoc-amino-6-hexanoyloxy-7-PamO-4-THT-O'Bu (GC-5b) and (2R,6R)-2-Fmoc-amino-6-hexanoyloxy-7-PamO-4-THT-OH (GC-5) were prepared.

		Materials (g)	Re	eaction Condit	ions	Pr	oducts (g	()
	a)	(Fmoc-(R)-Cys-O <sup>t</sup> Bu) <sub>2</sub>	1)	zinc	1.2	5 g	GC-5a	
5		(3.83)	aci	d mixt. soln.	13	ml	(5.07)	
•				DCM	25	m1		
				20°C	30	min		
			2)	(S)-O-palmito	ylgl;	ycidol	suppleme	nted
10					12.	0 g		
				50°C		8 h		
	b)	GC-5a	hex	anoic acid	1.9	6 g	GC-5b	
15		(4.80)	DIC	:	2.6	4 ml	(5.23)	
			DMA	ΔP	32	9 mg		
			THE	?	90	m1		
20			209	°C	17	h		
20	c)	GC-5b	TFA	7	20	m1	GC-5	
		(5.08)	209	°C	1.	5 h	(4.72	•

<sup>25</sup> Compound GC-5a: m.p. 57.5-58.8 °C  $[\alpha]_0^{20}$  - 2.9 ° (c = 0.55 in chloroform)

Elemental Analysis for C <sub>41</sub> H <sub>61</sub> NO <sub>7</sub> S:						
	C, 69.16; C, 68.95;		N, 1.97; N, 1.83;			

Compound GC-5b:  $[\alpha]_D^{20} + 1.1 \circ (c = 0.52, in chloroform)$ 

Elemental Analysis for C <sub>47</sub> H <sub>71</sub> NO <sub>8</sub> S:						
Calcd. Found	C, 69.68; C, 69.78;		N, 1.73; N, 1.78;	-		

Compound GC-5:  $[\alpha]_D^{20} + 14.0^{\circ}$  (c = 0.56 in chloroform)

Elemental Analysis for C <sub>43</sub> H <sub>63</sub> NO <sub>8</sub> S:					
Calcd. Found	C, 68.49; C, 68.27;			S, 4.25 S, 4.26	

50 Example 36

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Production of (2R,6R)-2-amino-6-hexanoyloxy-7-PamO-4-THT-Gly-Gly-Gly-Glu-OH (Compound 30)

In substantially the same manner as in Example 6, (2R,6R)-2-Fmoc-amino-6-hexanoyloxy-7-PamO-4-THT-Gly-Gly-Glu(O¹Bu)-O¹Bu (30a), (2R,6R)-2-amino-6-hexanoyloxy-7-PamO-4-THT-Gly-Gly-Gly-Glu-(O¹Bu)-O¹Bu (30b) and (2R,6R)-2-amino-6-hexanoyloxy-7-PamO-4-THT-Gly-Gly-Gly-Glu-OH (Compound 30) were produced.

		Materials (g)	Reaction Cor	nditions Pro	oducts (g)
	a)	GC-5	P-4 1	L.88 g	30a
		(3.00)	HONB 7	784 mg	(3.32)
5			DIC 0	).69 ml	
			DMF	25 ml	
			20 °C	16 h	
10	b·)	30a	piperidine	3.0 ml	30b
		(3.20)	DCM	30 ml	(2.26)
			20 °C	2 h	
15			silica-gel		
			(chloroform	n-methanol)	
			50:1, 20:1		
	c)	30b	TFA	5 ml	30
20		(0.53)	20 °C	2 h	(0.46)

Compound 30a:  $\left[\alpha\right]_{D}^{22}$  - 6.4° (c = 0.53 in chloroform)

Elemental Analysis for C<sub>62</sub>H<sub>95</sub>N<sub>5</sub>O<sub>14</sub>S:

Calcd. C, 63.84; H, 8.21; N, 6.00; S, 2.75

Found C, 63.67; H, 8.09; N, 5.89; S, 2.84

Compound 30b:  $[\alpha]_D^{22} - 8.2 \circ (c = 0.56, in chloroform)$ 

Elemental Analysis for C<sub>47</sub>H<sub>85</sub>N<sub>5</sub>O<sub>12</sub>S:

Calcd. C, 59.78; H, 9.07; N, 7.42; S, 3.40

Found C, 59.64; H, 8.87; N, 7.42; S, 3.37

Compound 30:  $[\alpha]_{\rm D}^{22}$  + 9.5 ° (c = 0.57 in 5% TFA-chloroform)

Elemental Analysis for C<sub>39</sub>H<sub>69</sub>N<sub>5</sub>O<sub>12</sub>S•1.5H<sub>2</sub>O:

Calcd. C, 54.53; H, 8.45; N, 8.15; S, 3.73

Found C, 54.56; H, 8.09; N, 8.21; S, 3.69

Example 37

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Production of (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-Glu-Gly-D-Glu-OH (Compound 31)

The compound 11a (600 mg) was dissolved in TFA (6.0 ml), and the solution was left standing for 2 hours at 20 °C. The reaction mixture was concentrated, and the concentrate was suspended in acetonitrile. The suspension was subjected to filtration to collect the compound 31 as a white powdery product (517 mg, yield 98%).

 $[\alpha]_0^{25}$  - 16.5 • (c = 0.58, in 5% TFA-chloroform)

Elemental Analysis for C <sub>65</sub> H <sub>100</sub> N <sub>4</sub> O <sub>15</sub> S•H <sub>2</sub> O:					
Calcd.	C, 63.60;	H, 8.38;	N, 4.56;	S, 2.61	
Found	C, 63.83;	H, 8.47;	N, 4.46;	S, 2.52	

Example 38

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Production of (2R,6R)-2-acetylamino-6,7-bis(PamO)-4-THT-Glu-Gly-D-Glu-OH (Compound 32)

Starting from the compound 11b, (2R,6R)-2-acetylamino-6,7-bis(PamO)-4-THT-Glu-Gly-D-Glu-OH (Compound 32) was produced via (2R,6R)-2-acetylamino-6,7-bis(PamO)-4-THT-Glu(O¹Bu)-Gly-D-Glu(O¹Bu)-O¹Bu (32a).

a) The compound 11b (700 mg) was dissolved in DCM (7.0 ml), to which was added acetic anhydride (86  $\mu$ l), and the mixture was stirred for one hour at 20 °C. The reaction mixture was concentrated. The concentrate was suspended in acetonitrile. The suspension was subjected to filtration to collect the compound 32a as a white powdery product (580 mg).

 $[\alpha]_D^{24}$  - 12.8° (c = 0.49, in chloroform)

Elemental Analysis for C <sub>64</sub> H <sub>116</sub> N <sub>4</sub> O <sub>14</sub> S•0.5H <sub>2</sub> O:					
Calcd.	C, 63.70;	H, 9.77;	N, 4.64;	S, 2.66	
Found	C, 63.80;	H, 9.79;	N, 4.54;	S, 2.55	

b) A TFA solution (5.3 ml) of the compound 32a (530 mg) was processed in substantially the same manner as in the deprotection reaction in Example 37 to give the compound 32 as a white powdery product (455 mg).

 $[\alpha]_0^{25}$  - 16.9 ° (c = 0.52, in 5% TFA-chloroform)

L	Elemental Analysis for C <sub>52</sub> H <sub>92</sub> N <sub>4</sub> O <sub>14</sub> S•H <sub>2</sub> O:					
	Calcd.	C, 59.63;	H, 9.05;	N, 5.35;	S, 3.06	
	Found	C, 59.59;	H, 8.99;	N, 5.15;	S, 2.91	

Example 39

Production of (2R,6R)-2-hexanoylamino-6-hexanoyloxy-7-PamO-4-THT-Gly-Gly-Gly-Glu-OH (Compound 33)

Starting from the compound 30b, (2R,6R)-2-hexanoylamino-6-hexanoyloxy-7-PamO-4-THT-Gly-Gly-Gly-Glu-OH (Compound 33) was produced via (2R,6R)-2-hexanoylamino-6-hexanoyloxy-7-PamO-4-THT-Gly-Gly-Gly-Gly-Glu(O¹Bu)-O¹Bu (33a).

a) The compound 30b (550 mg) was dissolved in DCM (10 ml). To the solution were added, under ice-cooling, HOBT (87 mg), WSC (123 mg) and hexanoic acid (80 µl). The mixture was stirred for 18 hours at 20 °C. The reaction mixture was concentrated, and the concentrate was dissolved in ethyl acetate, followed by washing with a 10% aqueous solution of ammonium chloride, a 2% aqueous solution of sodium hydrogencarbonate and water, successively. The ethyl acetate layer was dried over anhydrous sodium sulfate, which was then concentrate. The concentrate was suspended in acetonitrile, and the suspension was subjected to filtration to collect (2R,6R)-2-hexanoylamino-6-hexanoyloxy-7-PamO-4-THT-Gly-Gly-Gly(O¹Bu)-O¹Bu (33a) as a white powdery product (534 mg, yield 88%).

The compound 33a:  $[\alpha]_D^{22}$  - 8.0 ° (c = 0.56, in chloroform)

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Elemental Analysis for C <sub>53</sub> H <sub>95</sub> N <sub>5</sub> O <sub>13</sub> S:					
	C, 61.07; C, 61.02;				

b) The compound 33a (480 mg) was dissolved in TFA (5.0 ml), and the solution was stirred for 2 hours at 20 °C. The reaction mixture was concentrated, and the concentrate was suspended in acetonitrile. The suspension was subjected to filtration to collect the compound 33 (408 mg, yield 95%).

The compound 33 :  $[\alpha]_D^{22}$  - 12.7 • (c = 0.64, in 5% TFA - chloroform)

Elemental Analysis for C <sub>45</sub> H <sub>79</sub> N <sub>5</sub> O <sub>13</sub> S•0.5H <sub>2</sub> O:					
Calcd.	C, 57.55;	Н, 8.59;	N, 7.46;	S, 3.41	
Found	C, 57.74;	Н, 8.54;	N, 7.41;	S, 3.43	

Example 40

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Production of (2R,6R)-2-Fmoc-amino-6,7-bis(SteO)-4-THT-O¹Bu (GC-6a), and (2R,6R)-2-Fmoc-amino-6,7-bis(SteO)-4-THT-OH (GC-6)

In substantially the same manner as in Example 2, (2R,6R)-2-Fmoc-amino-6,7-bis(SteO)-4-THT-O<sup>t</sup>Bu (GC-6a), and (2R,6R)-2-Fmoc-amino-6,7-bis(SteO)-4-THT-OH (GC-6) were produced.

		Materials	Reaction (	Conditions	Products
	a)	GC-2a	Stearic acid	5.58 g	GC-6a
30	•	(2.65 g)	DIC	3.07 ml	(3.20 g)
30			DMAP	275 mg	
			THF	45 ml	
			20°C	14 h	
35	b)	GC-6a	TFA	15 ml	GC-6
	•	(3.00 g)	20°C	2.0 h	(2.75 g)

Compound GC-6a: m.p. 64.2-65.9 °C  $[\alpha]_D^{24} + 0.2$  ° (c = 0.92, in chloroform)

Elemental Analysis for C <sub>61</sub> H <sub>99</sub> NO <sub>8</sub> S:					
Calcd.	C, 72.79;	H, 9.91;	N, 1.39;	S, 3.19	
Found	C, 72.60,	H, 10.10;	N, 1.66;	S, 3.37	

Compound GC-6: m.p. 92.0-92.8 °C  $[\alpha]_D^{24} + 11.0$  ° (c = 0.67, in chloroform)

Elemental Analysis for C <sub>57</sub> H <sub>91</sub> NO <sub>8</sub> S:					
	C, 72.03; C, 72.21;		N, 1.47; N, 1.57,	S, 3.37 S, 3.24	

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## Example 41

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Production of (2R,6R)-2-Fmoc-amino-6,7-bis(MyrO)-4-THT-O'Bu (GC-7a) and (2R,6R)-2-Fmoc-amino-6,7-bis(MyrO)-4-THT-OH (GC-7)

In substantially the same manner as in Example 2, (2R,6R)-2-Fmoc-amino-6,7-bis(MyrO)-4-THT-O<sup>t</sup>Bu (GC-7a) and (2R,6R)-2-Fmoc-amino-6,7-bis(MyrO)-4-THT-OH (GC-7) were produced.

10		Materials	Reaction C	Conditions	Products
	a)	GC-2a	myristic acid	4.22 g	GC-7a
		(2.50 g)	DIC	2.89 ml	(2.58 g)
			DMAP	258 mg	
15			THF	45 ml	
			20°C	14 h	•
	b)	GC-7a	TFA	20 ml	GC-7
20		(3.00 g)	20°C	2.0 h	(1.93 g)

Compound GC-7a: m.p.  $49.2-50.9 \,^{\circ}$  C  $[\alpha]_{0}^{22} + 0.5 \,^{\circ}$  (c = 0.83, in chloroform)

Elemental Analysis for C<sub>53</sub>H<sub>83</sub>NO<sub>8</sub>S:

Calcd. C, 71.18; H, 9.35; N, 1.57; S, 3.59
Found C, 70.97, H, 9.24; N, 1.62; S, 3.52

Compound GC-7: m.p. 82.8-83.5 ° C  $[\alpha]_D^{22}$  + 12.7 ° (c = 0.58, in chloroform)

Elemental Analysis for C<sub>49</sub>H<sub>75</sub>NO<sub>8</sub>S•0.25H<sub>2</sub>O:

Calcd. C, 69.84; H, 9.03; N, 1.66; S, 3.81

Found C, 69.85, H, 9.09; N, 1.62, S, 3.78

Example 42

Production of (2R,6R)-2-amino-6,7-bis(SteO)-4-THT-Gly-Glu-Glu-OH (Compound 34)

In substantially the same manner as in Example 6, (2R,6R)-2-Fmoc-amino-6,7-bis(SteO)-4-THT-Gly-Glu-(O¹Bu)-Glu(O¹Bu)-O¹Bu (34a), (2R,6R)-2-amino-6,7-bis(SteO)-4-THT-Gly-Glu(O¹Bu)-Glu(O¹Bu)-O¹Bu(34b) and (2R,6R)-2-amino-6,7-bis(SteO)-4-THT-Gly-Glu-Glu-OH (Compound 34) were produced.

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a) GC-6 P-11 1.47 g 34a (2.53 g) HONB 525 mg (3.7)	75 a\
(2.53  g) HONB $525  mg$ (3.	75 ~ \
(2.00.3)	,, 9)
DIC 459 μℓ	
DMF 20 ml	
20°C 16 h	
b) 34a piperidine 3.6 ml 34b	
(3.56 g) DCM 30 ml (2.7	78 g)
20°C 2.0 h	
silica-gel	
(chroloform-methanol)	
20:1	
20 c) 34b TFA 6.0 ml 34	
(0.59 g) 20°C 2.0 h (0.5	51 g)

Compound 34a:  $[\alpha]_D^{24}$ -6.6 • (c = 0.62, in chloroform)

Elemental Analysis for C<sub>81</sub> H<sub>132</sub>N<sub>4</sub>O<sub>15</sub>S:

Calcd. C, 67.84; H, 9.28; N, 3.91; S, 2.24

Found C, 67.59, H, 9.54; N, 4.27; S, 2.46

Compound 34b:  $[\alpha]_D^{24}$ -10.7 • (c = 0.75, in chloroform)

Elemental Analysis for C<sub>66</sub>H<sub>122</sub>N<sub>4</sub>O<sub>13</sub>S•H<sub>2</sub>O:

Calcd. C, 64.46; H, 10.16; N, 4.56; S, 2.61

Found C, 64.46, H, 10.07; N, 4.83, S, 2.64

Compound 34:  $[\alpha]_D^{24} + 10.8^{\circ}$  (c = 0.65, 5% in TFA-chloroform)

Elemental Analysis for C<sub>54</sub>H<sub>98</sub>N<sub>4</sub>O<sub>13</sub>S•4H<sub>2</sub>O:

Calcd. C, 58.14; H, 9.58; N, 5.02; S, 2.87
Found C, 58.48, H, 9.20; N, 5.20, S, 2.72

Example 43

Production of (2R,6R)-2-amino-6,7-bis(MyrO)-4-THT-Gly-Glu-OH (Compound 35)

In substantially the same manner as in Example 6, (2R,6R)-2-Fmoc-amino-6,7-bis(MyrO)-4-THT-Gly-Glu(O¹Bu)-Glu(O¹Bu)-O¹Bu (35a), (2R,6R)-2-amino-6,7-bis(MyrO)-4-THT-Gly-Glu(O¹Bu)-Glu(O¹Bu)-Glu(O¹Bu)-O¹Bu (35b) and (2R,6R)-2-amino-6,7-bis(MyrO)-4-THT-Gly-Glu-Glu-OH (Compound 35) were produced.

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		Materials	Reaction	Conditions	Products
	a)	GC-7	P-11	1.21 g	35a
5		(1.83 g)	HONB	432 mg	(2.22 g)
			DIC	378 µg	
			DMF	20 ml	
			20°C	18 h	
10	b)	35a	piperidine	3.6 ml	35b
		(2.10 g)	DCM	30 ml	(1.58 g)
			20°C	2.0 h	
15			silica-g	el	•
			(chrolof	orm-methanol)	
			20:1		
20	C)	35b	TFA	10 ml	35
		(1.00 g)	20°C	2.0 h	(0.85 g)

Compound 35a:  $[\alpha]_D^{22}$ -7.6 ° (c = 0.71, in chloroform)

Elemental Analysis for C <sub>73</sub> H <sub>116</sub> N <sub>4</sub> O <sub>15</sub> S:						
Calcd. Found	_, ,	H, 8.85; H, 8.97;	N, 4.24; N, 4.38;	_,		

Compound 35b:  $[\alpha]_D^{22}$ -12.0 • (c = 1.11, in chloroform)

Elemental Analysis for C <sub>58</sub> H <sub>106</sub> N₄ O <sub>13</sub> S:					
	C, 63.36; C, 63.14,				

Compound 35:  $[\alpha]_D^{22} + 10.9^{\circ}$  (c = 0.86, in 5% TFA-chloroform)

Elemental Analysis for C <sub>46</sub> H <sub>82</sub> N <sub>4</sub> O <sub>13</sub> S•4H <sub>2</sub> O:					
Calcd.	C, 55.07;	H, 9.04;	N, 5.58;	S, 3.20	
Found	C, 54.96,	H, 9.12;	N, 5.25,	S, 2.94	

# Example 44

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Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-NH(CH<sub>2</sub>)<sub>7</sub> CO-Glu-OH hydrochloride (Compound 36)

a) To a solution of GC-2 (179 mg) synthesized in Example 3 and P-17 (80 mg) synthesized in Reference Example 17 in DMF (4 ml) were added TEA (0.033 ml) and DEPC (49 mg). The mixture was stirred for 90 minutes at 20 °C. To the reaction mixture was added water, which was subjected to extraction with ethyl acetate. The extract solution was washed with a 5% aqueous solution of citric acid, a saturated aqueous solution of sodium hydrogencarbonate and a saturated aqueous saline solution, followed by drying over anhydrous sodium sulfate. The solvent was distilled off, and the residue was purified by means of a silica gel column chromatography (hexane:ethyl acetate = 1:1) to give (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-NH(CH<sub>2</sub>)<sub>7</sub> CO-Glu(O¹Bu)-O¹Bu (36a) (105 mg, yield 41%) as a colorless waxy

compound.

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IR (KBr) v: 3300, 1730, 1685, 1660, 1645 cm<sup>-1</sup>

 $^{1}$ H-NMR (CDCl<sub>3</sub>) δ: 0.88 (6H,t,J=7.0Hz), 1.03-1.38 (56H,m), 1.44 (9H,s), 1.47 (9H,s), 1.49-2.40 (12H,m), 2.79 (2H,d,J=5.2Hz), 2.93 (2H,d,J=7.0Hz), 3.17-3.32 (2H,m), 4.00-4.56 (7H,m), 5.17-5.32 (1H,m), 5.78-5.90 (1H,m), 6.19 (1H,d,J=8.0Hz), 6.44-6.56 (1H,m), 7.26-7.46 (4H,m), 7.61 (2H,d,J=7.4Hz), 7.78 (2H,d,J=7.4Hz)

b) To the compound 36a (104 mg) obtained thus above was added piperidine (2 ml), and the mixture was stirred for 30 minutes at room temperature. The reaction mixture was then concentrated. The concentrate was purified by means of a silica gel column chroamtography (hexane:ethyl acetate = 1:1 ~ chloroform:methanol = 19:1) to give (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-NH(CH<sub>2</sub>)<sub>7</sub> CO-Glu(O¹Bu)-O¹Bu (36b) (80 mg, yield 93%) as a colorless waxy compound.

IR (KBr) v: 3320, 1735, 1650 cm<sup>-1</sup>

<sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ: 0.88 (6H,t,J=6.8Hz), 1.03-1.38 (52H,m), 1.44 (9H,s), 1.47 (9H,s), 1.49-2.43 (12H,m), 2.20 (2H,t,H=7.8Hz), 2.31 (2H,t,J=7.8Hz), 2.33 (2H,t,J=7.4Hz), 2.72 (1H,dd,J=8.8, 13.6Hz), 2.75 (2H,d,J=6.0Hz), 3.12 (1H,dd,J=3.8, 13.6Hz), 3.15-3,29 (2H,m), 3.48 (1H,dd,J=3.8, 8.8Hz), 4.14 (1H,dd,J=6.0, 11.8Hz), 4.35 (1H,dd,J=3.6, 11.8Hz), 4.41-4.55 (1H,m), 5.09-5.22 (1H,m), 6.15 (1H,d,J=8.0Hz), 7.40 (1H,t,J=5.8Hz).

c) To the compound 36b (80 mg) obtained as above was added a 4N HCl ethyl acetate solution (4 ml), which was stirred for 4 hours at 20 °C. The solvent was distilled off to give (2R,6R)-2-amino-6,7-bis-(PamO)-4-THT-NH(CH<sub>2</sub>)<sub>7</sub> CO-Glu-OH hydrochloride (36) (74 mg, yield 100%) as a white powdery product, m.p. 53-56 °C.

 $[\alpha]_D^{23} + 6.0$ ° (c = 0.50, in chloroform);

Elemental Analysis for C <sub>52</sub> H <sub>89</sub> N <sub>4</sub> O <sub>11</sub> SC1 • 2H <sub>2</sub> O:					
Calcd.	C, 59.49;	H, 8.93;	N, 5.34;	S, 3.05	
Found	C, 59.62,	H, 8.77;	N, 5.37,	S, 3.14	

IR (KBr)  $\nu$ : 3250, 1740, 1715, 1640 cm<sup>-1</sup>

<sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 0.88 (6H,t,J=6.8Hz), 1.03-1.50 (56H,m), 1.50-1.80 (6H,m), 1.80-3.55 (16H,m), 4.03-4.70 (4H,m), 5.10-5.35 (1H,m)

### Example 45

Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-NH(CH<sub>2</sub>)<sub>11</sub>CO-Glu-OH hydrochloride (Compound 37)

In substantially the same manner as in Example 44, (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-NH- $(CH_2)_{11}$ CO-Glu(O<sup>1</sup>Bu)-O<sup>1</sup>Bu (37a), (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-NH( $CH_2)_{11}$ CO-Glu(O<sup>1</sup>Bu)-O<sup>1</sup>Bu (37b) and (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-NH( $CH_2)_{11}$ CO-Glu-OH hydrochloride (Compound 37) were produced.

		Materials	Reaction	n Conditions	Products
45	a)	GC-2	P-18	91 mg	37a
		(179 mg)	DEPC	49 mg	(52 mg)
			TEA	33 µl	
50			DMF	4 ml	
50			20°C	90 min	
	b)	37a	piperidine	2 ml	37b
		(104 mg)	20°C	30 min	(31 mg)
55	c)	37b	4N HCl ethyl	acetate	37
		(31 mg)		4 ml	(29 mg)
			20°C	4.0 h	

Compound 37a: IR (KBr) v: 3280, 1730, 1685, 1645 cm<sup>-1</sup>

'H-NMR (CDCl<sub>3</sub>) δ: 0.88 (6H,t,J=7.0Hz), 0.95-1.38 (64H,m), 1.44 (9H,s), 1.47 (9H,s), 1.50-2.40 (12H,m), 2.79 (2H,d,J=5.2Hz), 2.86-2.96 (1H,m), 3.18-3.32 (2H,m), 4.08-4.55 (6H,m), 5.18-5.33 (1H,m), 5.73-5.85 (1H,m), 6.14 (1H.d.J = 7.8Hz), 6.40-6.53 (1H,m), 7.25-7.45 (4H,m), 7.61 (2H,d.J = 7.4Hz), 7.78 (2H,d.J = 7.0Hz) Compound 37b: IR (KBr)  $\nu$ : 3360, 3310, 1730, 1650 cm<sup>-1</sup> <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 0.88 (6H,t,J=6.4Hz), 1.03-1.39  $(56H,m), \quad 1.44 \quad (9H,s), \quad 1.47 \quad (9H,s), \quad 1.52-1.72 \quad (12H,m), \quad 1.72-2.43 \quad (2H,m), \quad 2.20 \quad (2H,t,J=8.0Hz), \quad 2.31 \quad (2H,m), \quad 2.20 \quad (2H,t,J=8.0Hz), \quad$  $(2H,t,J=7.8Hz),\ 2.33\ (2H,t,J=7.6Hz),\ 2.73\ (1H,dd,J=8.7\ 13.0Hz),\ 2.75\ (2H,d,J=7.4Hz),\ 3.11\ (1H,dd,J=3.7,d)$ 13.0Hz), 3.23 (2H,dt,J=7.4, 7.4Hz), 3.48 (1H,dd,J=3.7, 8.7Hz), 4.14 (1H,dd,J=6.2, 12.0Hz), 4.35  $(1H,dd,J=3.2,\ 12.0Hz),\ 4.42-4.56\ (1H,m),\ 5.08-5.22\ (1H,m),\ 6.15\ (1H,d,J=7.8Hz),\ 7.39\ (1H,t,J=7.4Hz)$ 

Compound 37: m.p. 61-64 ° C  $[\alpha]_D^{23} + 4.2 \circ (c = 0.50, in chloroform);$ 

Elemental Analysis for C <sub>55</sub> H <sub>104</sub> N <sub>3</sub> O <sub>10</sub> SC1 • 2H <sub>2</sub> O:					
Calcd.	C, 61.68;	Н, 10.16;	N, 3.92;	S, 2.99	
Found	C, 62.02,	Н, 9.33;	N, 3.73,	S, 2.64	

IR (KBr) v: 3250, 1740, 1720, 1640 cm<sup>-1</sup>

 $^1\text{H-NMR}$  (CDCl<sub>3</sub>)  $\delta$ : 0.88 (6H,t,J=6.8Hz), 1.03-1.50 (64H,m), 1.50-1.80 (6H,m), 1.80-3.55 (16H,m), 4.03-4.70 (4H,m), 5.10-5.35 (1H,m)

#### Example 46

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Production of (2R,6R)-4-(2-amino-6,7-bis(PamO)-4-THT-amino)benzoyl-Glu-OH hydrochloride (Compound

a) To a solution of the compound P-19 (85 mg) in pyridine (1.5 ml) was added phosphorus trichloride (0.01 ml). The mixture was stirred for 2 hours at 20 °C, to which was added the compound GC-2 (100 mg), followed by stirring for 3 hours at 20 °C. To the reaction mixture was added water, which was subjected to extraction with ethyl acetate. The extract was washed with a 5% aqueous solution of citric acid, a saturated aqueous solution of sodium hydrogencarbonate and a saturated aqueous saline solution, followed by drying over anhydrous sodium sulfate. The solvent was the distilled off, and the residue was purified by means of a silica gel column chromatography (chloroform:hexane = 7:1) to give (2R,6R)-4-(2-Fmoc-amino-6,7-bis(PamO)-4-THT-amino)benzoyl-Glu(O<sup>t</sup>Bu)-O<sup>t</sup>Bu (38a) (107 mg, yield 84%) as a colorless waxy compound.

IR (KBr) v: 3300, 1730, 1685, 1660, 1640, 1600 cm<sup>-1</sup>

¹H-NMR (CDCl<sub>3</sub>) δ: 0.88 (6H,t,J=7.0Hz), 1.10-1.38 (48H,m), 1.42 (9H,s), 1.49 (9H,s), 1.55-1.83 (4H,m), 1.90-2.58 (8H,m), 2.75-2.86 (2H,m), 3.02 (2H,d,J = 6.6Hz), 4.21-4.72 (8H,m), 5.24-5.49 (1H,m), 5.74-5.86(1H,m), 7.01 (2H,d,J=7.6Hz), 7.32 (2H,d,J=7.6Hz), 7.35-7.87 (8H,m), 8.70  $(1H,br\ s)$ 

- b) To the compound 38a (107 mg) obtained as above was added piperidine (2 ml). The mixture was stirred for 3 hours at 20 °C, then the reaction mixture was concentrated. The concentrate was purified by means of a silica gel column chromatography (hexane:ethyl acete = 1:7) to give (2R,6R)-4-(2-amino-6,7bis(PamO)-4-THT-amino)benzoyl-Glu(O'Bu)-O'Bu(38b) (66 mg, yield 68%) as a colorless waxy product. iR (KBr) v: 3430, 3380, 1735, 1640 cm<sup>-1</sup>
- 45 ¹H-NMR (CDCl<sub>3</sub>) δ: 0.88 (6H,t,J=6.6Hz), 1.05-1.37 (48H,m), 1.42 (9H,s), 1.49 (9H,s), 1.52-1.82 (4H,m), 1.93-2.53 (8H,m), 2.70-4.00 (5H,m), 4.05-4.72 (3H,m), 5.10-5.24 (1H,m), 7.23-7.95 (4H,m), 9.71 (1H,br s) c) To the compound 38b (66 mg) obtained as above was added 4N HCl ethyl acetate solution (3 ml), which was stirred for 4 hours at 20 °C. The solvent was distilled off to give compound 38 (61 mg, yield 100%) as a colorless powdery product, m.p. 94.0-96.0 °C.

50  $[\alpha]_D^{23} + 6.2 \circ (c = 0.50, in chloroform);$ 

Elemental Analysis for C <sub>50</sub> H <sub>86</sub> N <sub>3</sub> O <sub>10</sub> SC1 • 2H <sub>2</sub> O:						
Calcd.	C, 60.49;	H, 9.14;	N, 4.23;	S, 3.23		
Found	C, 60.39,	H, 8.98;	N, 4.19,	S, 3.18		

IR (KBr) v: 3400, 1750, 1630, 1610 cm<sup>-1</sup>

<sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 0.88 (6H,t,J=6.8Hz), 1.02-1.48 (48H,m), 1.48-1.68 (4H,m), 2.00-2.85 (13H,m), 4.05-5.32 (4H,m), 7.40-8.00 (4H,m)

### Example 47

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Production of (2R,6R)-4-(2-amino-6,7-bis(PamO)-4-THT-Gly-amino)benzoyl-Glu-OH hydrochloride (Compound 39)

In substantially the same manner as in Example 44, (2R,6R)-4-(2-Fmoc-amino-6,7-bis(PamO)-4-THT-Gly-amino)benzoyl-Glu(O¹Bu)-O¹Bu (39a), (2R,6R)-4-(2-amino-6,7-bis(PamO)-4-THT-Gly-amino)benzoyl-Glu-(O¹Bu)-O¹Bu (39b) and (2R,6R)-4-(2-amino-6,7-bis(PamO)-4-THT-Gly-amino)benzoyl-Glu-OH hydrochloride (Compound 39) were produced.

20		Materials Reaction Conditions		Conditions	Products	
	a)	GC-2	P-20	87 mg	39a	
25						
		(179 mg)	DEPC	49 mg	(146 mg)	
			TEA	. 33 µl		
30			DMF	4 ml		
			20°C	90 min		
	b)	39a	piperidine	3 ml	39b	
35		(146 mg)	20°C	4.5 h	(87 mg)	
	c)	39b	4N HCl ethyl	acetate		
		(87 mg)		4 ml	39	
			20°C	4.0 h	(78 mg)	

Compound 39a: IR (KBr) v: 3300, 1735, 1730, 1700, 1655, 1640 cm<sup>-1</sup>

<sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ: 0.88 (6H,t,J=6.6Hz), 1.02-1.84 (64H,m), 1.42 (9H,s), 1.49 (9H,s), 1.84-3.15 (13H,m), 4.00-4.73 (9H,m), 5.15-5.28 (1H,m), 5.80-5.92 (1H,m), 6.92-7.13 (2H,m), 7.26-7.83 (12H,m), 8.53 (1H,br s) Compound 39b: IR (KBr)  $\nu$ : 3350, 1735, 1650, 1600 cm<sup>-1</sup>

<sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ: 0.88 (6H,t,J=7.0Hz), 1.52-1.76 (4H,m), 1.95-2.53 (8H,m), 2.75 (2H,d,J=6.6Hz), 2.92 (1H,dd,J=7.8, 13.6Hz), 3.11 (1H,dd,J=4.2, 13.6Hz), 3.66 (1H,dd,J=4.2, 7.8Hz), 4.01-4.24 (3H,m), 4.37 (1H,dd,J=3.4, 11.8Hz), 4.57-4.62 (1H,m), 5.08-5.22 (1H,m), 7.00 (1H,d,J=7.6Hz), 7.59 (2H,d,J=8.8Hz), 7.79 (2H,d,J=8.8Hz), 8.19 (1H,t,J=5.6Hz), 8.63 (1H,s) Compound 39: m.p. 100-101 °C IR (KBr)  $\nu$ : 3400, 1735 cm<sup>-1</sup>

Elemental Analysis for C <sub>51</sub> H <sub>96</sub> N <sub>3</sub> O <sub>10</sub> SC1•H <sub>2</sub> O:						
Calcd.	C, 61.44;	H, 9.91;	N, 4.41;	S, 3.22		
Found	C, 61.17,	H, 9.81;	N, 4.26,	S, 3.21		

 $^{1}$ H-NMR (CDCl<sub>3</sub>) δ: 0.88 (6H,t,J = 6.8Hz), 0.93-1.47 (48H,m), 1.47-1.70 (4H,m), 1.70-3.32 (16H,m), 3.70-4.60 (4H,m), 4.60-4.78 (1H,m), 5.17-5.32 (1H,m), 7.47-7.90 (4H,m)

#### Example 48

Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-NH(CH₂)₅ CO-Glu-OH TFA salt (Compound 40)

In substantially the same manner as in Example 44, (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-NH-(CH<sub>2</sub>)<sub>5</sub> CO-Glu(O'Bu)-O'Bu (40a) was synthesized. Then, (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-NH(CH<sub>2</sub>)<sub>5</sub> CO-Glu(O'Bu)-O'Bu (40b) and (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-NH(CH<sub>2</sub>)<sub>5</sub> CO-Glu-OH TFA salt (Compound 40) were produced.

10		Materials	Reactio	n Conditions	Products
	a)	GC-2	P-21	64 mg	40a
		(150 mg)	DEPC	35 mg	(112 mg)
15			TEA	44 mg	
			DMF	3 ml	
			0°C	1 h	

Compound 40a:  $^{1}$ H-NMR (CDCl<sub>3</sub>)  $\delta$ : 0.878 (6H,t,J=6.0Hz), 1.247 (52H,s), 1.433 (9H,s), 1.463 (9H,s), 1.465-1.720 (6H,m), 1.72-2.15 (2H,m), 2.15-2.40 (16H,m), 2.785 (2H,d,J=5.8Hz), 2.927 (2H,d,J=6.2Hz), 3.265 (2H,m), 4.05-4.55 (7H,m), 5.232 (1H,br s), 5.825 (1H,d,J=8.2Hz), 6.236 (1H,d,J=8.2Hz), 6.595 (1H,br s), 7.366 (4H,m), 7.600 (2H,d,J=7.0Hz), 7.766 (2H,d,J=7.0Hz)

25 IR (neat) v: 3300, 2920, 2850, 1735, 1650, 1530, 1440, 1370, 1255, 1245, 1225, 1155 cm<sup>-1</sup>

b) The compound 40a (112 mg) was dissolved in DCM (0.2 ml). To the solution was added piperidine (2 ml), and the mixture was stirred for 2 hours at 20 °C. The reaction mixture was concentrated under reduced pressure. The concentrate was allowed to be adsorbed on a silica gel column (5 g) processed with ammonia. Elution was conducted with chloroform to give the compound 40b as a colorless oily product (83 mg, yield 89.8%).

<sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ: 0.878 (6H,t,J=6.0Hz), 1.254 (52H,s), 1.467 (9H,s), 1.499 (9H,s), 1.47-1.75 (6H,m), 1.72-2.15 (2H,m), 2.15-2.40 (10H,m), 2.60-2.80 (3H,m), 3.107 (1H,dd,J=3.8Hz,13.6Hz), 3.242 (2H,q,J=6.2Hz), 3.475 (1H,dd,J=4.0Hz,6.8Hz), 4.142 (1H,dd,J=6.0Hz,12.0Hz), 4.358 (1H,dd,J=3.4Hz,11.8Hz), 4.472 (2H,m), 5.154 (1H,m), 6.90 (1H,d,J=7.6Hz), 7.424 (1H,t,J=7.2Hz) IR (neat)  $\nu$ : 3300, 2920, 2850, 1735, 1650, 1530, 1440, 1370, 1255, 1245, 1225, 1155 cm<sup>-1</sup>

c) The compound 40b (83 mg) was dissolved in DCM (0.2 ml). To the solution was added TFA (1 ml), and the mixture was stirred for 2 hours at 20 °C. To the reaction mixture was added toluene (1 ml), and the mixture was concentrated under reduced pressure. This process was repeated again to give the compound 40 as a white powdery product (83 mg, yield 100%), m.p. 62-63 °C.

 $[\alpha]_0^{23} + 3.4^{\circ}$  (c = 0.16, in chloroform);

<sup>1</sup>H-NMR (CDCl<sub>3</sub>-TFA) δ: 0.879 (6H,t,J=6.6Hz), 1.256 (52H,s), 1.567 (6H,m), 1.90-2.20 (2H,m), 2.374 (8H,m), 2.597 (2H,t,J=6.4Hz), 2.744 (2H,d,J=7.0Hz), 3.100 (2H,m), 4.130 (1H,dd,J=6.0Hz,12.0Hz), 4.28-4.50 (2H,m), 4.650 (1H,m), 5.170 (1H,m), 7.080 (1H,d,J=7.5Hz), 7.420 (1H,br s) IR (KBr)  $\nu$ : 3320, 3100, 2920, 2850, 1740, 1665, 1550, 1465, 1200 cm<sup>-1</sup>

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Elemental Analysis for C <sub>51</sub> H <sub>92</sub> N <sub>3</sub> O <sub>12</sub> SF <sub>3</sub> • H <sub>2</sub> O:					
Calcd.	C, 58.54;	H, 9.05;	N, 4.02;	S, 3.06	
Found	C, 58.35,	H, 8.91;	N, 4.06,	S, 2.76	

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## Example 49

Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-NH(CH₂)₅ NHCO-Glu-OH TFA salt (Compound 41)

In substantially the same procedure as in Example 48, (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-NH(CH<sub>2</sub>)<sub>6</sub>NHCO-Glu(O¹Bu)-O¹Bu (41a), (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-NH(CH<sub>2</sub>)<sub>6</sub>NHCO-Glu(O¹Bu)-O¹Bu (41b), and (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-NH(CH<sub>2</sub>)<sub>6</sub>NHCO-Glu-OH TFA salt (Compound 41)

were produced.

_		Materials	Reaction	Conditions	Products
5	a)	GC-2	P-22	83 mg	41a
		(150 mg)	DEPC	35 mg	(203 mg)
			TEA	44 mg	
10			DMF	3 ml	•
			0°C	1 h	
	b)	41a	piperidine	2 ml	41b
15		(203 mg)	DCM	0.2 ml	(137 mg)
			20°C	1.5 h	
	C)	41b	TFA	1 ml	41
		(137 mg)	20°C	2 h	(136 mg)
20					

Compound 41a:  $^{1}$ H-NMR (CDCl<sub>3</sub>)  $\delta$ : 0.879 (6H,t,J=7.0Hz), 1.254 (52H,s), 1.430 (9H,s), 1.448 (9H,s), 1.727 (8H,m), 1.75-2.20 (2H,m), 2.325 (6H,m), 2.780 (2H,d,J=5.0Hz), 2.944 (2H,d,J=5.2Hz), 3.05-3.50 (4H,m), 4.00-4.50 (9H,m), 4.860 (1H,t,J=6.0Hz), 5.249 (2H,d,J=8.2Hz), 6.08 (1H,br s), 6.710 (1H,br s), 7.260-7.55 (4H,m), 7.612 (2H,d,J=8.0Hz), 7.767 (2H,d,J=8.0Hz) IR (neat)  $\nu$ : 3300, 2920, 2850, 1730, 1685, 1640, 1560, 1550, 1530, 1460, 1445, 1360, 1250, 1150, 1100, 1030 cm<sup>-1</sup>

Compound 41b:  $^1\text{H-NMR}$  (CDCl<sub>3</sub>)  $\delta$ : 0.879 (6H,t,J=6.2Hz), 1.254 (52H,s), 1.439 (9H,s), 1.461 (9H,s), 1.597 (8H,m), 1.75-2.20 (2H,m), 2.311 (6H,m), 2.745 (2H,m), 3.00-3.45 (4H,m), 3.483 (1H,m), 4.136 (1H,dd,J=6.4Hz,12.4Hz), 4.353 (2H,m), 4.700 (1H,t,J=5.2Hz), 5.078 (1H,d,J=7.6Hz), 5.162 (1H,br s), 7.440 (1H,br s)

IR (neat) v: 3350, 2920, 2850, 1730, 1640, 1560, 1460, 1450, 1390, 1360, 1250, 1260, 1150, 1100, 750, 730 cm<sup>-1</sup> Compound 41: m.p. 44-46 ° C

 $[\alpha]_D^{23}$ -2.9 ° (c = 0.415, in chloroform);

<sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 0.880 (6H,t,J=6.2Hz), 1.255 (52H,s), 1.591 (8H,m), 2.0-2.20 (2H,m), 2.302 (6H,m), 2.779 (2H,m), 3.00-3.45 (3H,m), 3.516 (2H,m), 3.80-4.25 (6H,m), 4.340 (2H,m), 5.200 (1H,br s), 7.710 (1H,br s)

IR (KBr)  $\nu$ : 3400, 2920, 2850, 1740(sh.), 1700, 1670(sh.), 1460, 1420, 1280, 1260, 1240, 1200, 1175, 1130 cm<sup>-1</sup>

Elemental Analysis for C <sub>50</sub> H <sub>95</sub> N <sub>4</sub> O <sub>12</sub> SF <sub>3</sub> :						
ſ	Calcd.	C, 58.11;	H, 9.27;	N, 5.42;	S, 3.10	
	Found	C, 58.19,	H, 8.77;	N, 5.24,	S, 2.78	

## Example 50

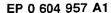
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Production of (2R,6R)-4-(2-amino-6,7-bis(PamO)-4-THT-aminomethyl)benzoyl-Glu-OH hydrochloride (Compound 42)

In substantially the same manner as in Example 44, (2R,6R)-4-(2-Fmoc-amino-6,7-bis(PamO)-4-THT-aminomethyl)benzoyl-Glu(O¹Bu)-O¹Bu (42a), (2R,6R)-4-(2-amino-6,7-bis(PamO)-4-THT-aminomethyl)-benzoyl-Glu(O¹Bu)-O¹Bu (42b) and (2R,6R)-4-(2-amino-6,7-bis(PamO)-4-THT-aminomethyl)benzoyl-Glu-OH hydrochloride (Compound 42) were produced.



		Materials	Reaction	n Conditions	Products
	a)	GC-2	P-23	106 mg	42a
5		(200 mg)	DEPC	55 mg	(247 mg)
			TEA	70 mg	
			DMF	10 ml	
10			20°C	30 min	
	b)	42a pi	peridine	2 ml	42b
		(243 mg)	DCM	0.5 ml	(195 mg)
			20°C	30 min	
15	c)	42b 4N HCl	ethyl aceta	ate 4 ml	42
		(112 mg)	20°C	2 h	(88 mg)

20 Compound 42a: IR (neat)  $\nu$ : 3300, 2920, 2850, 1730, 1660, 1530, 1500, 1445, 1360, 1240, 1150 cm<sup>-1</sup> <sup>1</sup>H-NMR (CDCl<sub>2</sub>)  $\delta$ : 0.88 (6H,t,J = 6.8Hz), 1.25 (48H,s), 1.42 (9H,s), 1.49 (9H,s), 1.40-1.65 (4H,m), 1.95-2.50 (8H,m), 2.77 (2H,d,J = 6.6Hz), 2.90-3.00 (2H,m), 4.05-4.55 (8H,m), 4.66 (1H,m), 5.24 (1H,m), 5.78 (1H,br), 6.98 (1H,br), 7.02 (1H,d,J = 7.4Hz), 7.25-7.45 (6H,m), 7.58 (2H,d,J = 7.4Hz), 7.76 (2H,d,J = 7.2Hz), 7.78 (2H,d,J = 8.4Hz)

<sup>25</sup> Compound 42b: IR (neat) v: 3350, 2920, 2850, 1730, 1650, 1535, 1520, 1500, 1460, 1450, 1360, 1250, 1150 cm<sup>-1</sup>

<sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ: 0.88 (6H,t,J = 6.8Hz), 1.25 (48H,s), 1.42 (9H,s), 1.49 (9H,s), 1.45-1.65 (4H,m), 1.73 (2H,br s), 1.90-2.50 (8H,m), 2.75 (2H,d,J = 6.4Hz), 2.81 (1H,dd,J = 13.4, 8.4Hz), 3.14 (1H,dd,J = 13.4,3.8Hz), 3.57 (1H,dd,J = 8.4,4.0Hz), 4.14 (1H,dd,J = 12.0,6.2Hz), 4.36 (1H,dd,J = 12.0,3.2Hz), 4.49 (2H,d,J = 6.2Hz), 4.66 (1H,m), 5.16 (1H,m), 7.02 (1H,d,J = 7.4Hz), 7.35 (2H,d,J = 8.2Hz), 7.79 (2H,d,J = 8.2Hz), 7.83 (1H,br)

Compound 42: m.p. 79-80 ° C

[ $\alpha$ ]<sub>0</sub><sup>20</sup> + 15.1° (c = 0.305, in chloroform); IR (KBr)  $\nu$ : 3450, 2920, 2850, 1730, 1710, 1690, 1670, 1640, 1630, 1620, 1560, 1540, 1500, 1460 cm<sup>-1</sup> <sup>1</sup>H-NMR (CDCl<sub>3</sub>-TFA)  $\delta$ : 0.88 (6H,t,J = 6.8Hz), 1.25 (48H,s), 1.40-1.65 (4H,m), 2.00-3.30 (12H,m), 4.00-4.80 (6H,m), 5.22 (1H,m), 7.00-8.10 (6H,m).

Elemental Analysis for C <sub>51</sub> H <sub>87</sub> N <sub>3</sub> O <sub>10</sub> S•HCl•H <sub>2</sub> O:					
Calcd			N, 4.25; N, 4.24,	S, 3.24, S, 3.44,	CI, 3.59 CI, 3.79

## Example 51

Production of (2R,6R)-4-(N-(2-Fmoc-amino-6,7-bis(PamO)-4-THT)-N-(carboxymethyl)aminomethyl)benzoyl-Glu-OH (Compound 43)

In substantially the same manner as in Example 44, (2R,6R)-4-(N-(2-Fmoc-amino-6,7-bis(PamO)-4-THT)-N-(t-butyloxycarbonylmethyl)aminomethyl)benzoyl-Glu(O<sup>t</sup>Bu)-O<sup>t</sup>Bu (43a), and (2R,6R)-4-(N-(2-Fmoc-amino-6,7-bis(PamO)-4-THT)-N-(carboxymethyl)aminomethyl)benzoyl-Glu-OH (Compound 43) were produced.

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		Materials	Reactio	n Conditions	Products
	a)	GC-2	P-24	62 mg	43a
5		(100 mg)	DEPC	30 mg	(134 mg)
Ū			TEA	48 mg	
			DMF	6 ml	
			20°C	1 h	
10	b)	43a	TFA	1 ml	43
		(110 mg)	DCM	0.5 ml	(70 mg)

Compound 43a: IR (neat) ν: 2920, 2850, 1730, 1650, 1360, 1250, 1150 cm<sup>-1</sup>

<sup>1</sup>H-NMR (CDCl<sub>3</sub>) δ: 0.88 (6H,t,J=6.8Hz), 1.25 (48H,s), 1.40 (1/2x9H,s), 1.42 (9H,s), 1.44 (1/2x9H,s), 1.49 (9H,s), 1.45-1.70 (4H,m), 2.00-2.50 (8H,m), 2.65-3.20 (4H,m), 3.70-4.50 (8H,m), 4.60-4.80 (3H,m), 5.17 (1H,m), 5.70-5.80 (1H,m), 7.00-7.10 (1H,m), 7.25-7.45 (6H,m), 7.61 (2H,d,J=7.4Hz), 7.75-7.90 (4H,m) Compound 43: m.p. 58-59 °C

 $[\alpha]_D^{20}$ -12.6 • (c = 0.25, in chloroform);

IR (KBr)  $\nu$ : 2921, 2852, 1739, 1640, 1540, 1465, 1450, 1250, 1220, 1160 cm<sup>-1</sup> <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 0.87 (6H,t,J=6.6Hz), 1.25 (48H,s), 1.40-1.70 (4H,m), 2.00-2.50 (8H,m), 2.60-3.00 (4H,m), 3.90-4.50 (8H,m), 5.17 (1H,m), 6.27 (1H,m), 7.15-7.45 (6H,m), 7.50-7.85 (6H,m)

Elemental Analysis for C <sub>68</sub> H <sub>100</sub> N <sub>3</sub> O <sub>14</sub> S•H <sub>2</sub> O:					
	Calcd.	C, 66.21;	H, 8.33;	N, 3.41;	S, 2.60
	Found	C, 66.44.	H, 8.32;	N, 3.21,	S, 2.66

Example 52

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Production of (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Lys-Gly-OH 2 TFA salt (compound 44)

In substantially the same manner as in Example 6, (2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-THT-Gly-Lys(Boc)-Gly-O¹Bu (44a), (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Lys(Boc)-Gly-O¹Bu (44b) and (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-Lys-Gly-OH 2 TFA salt (compound 44) were produced.

		Materials (g)	Reaction Con	ditions Pro	ducts (g)
	a)	GC-2	P-25 1	.01 mg	44a
5		(1.97)	HONB 4	35 mg	(2.24)
			DIC 3	80 µl	
			DMF	30 ml	
			20 °C	13 h	
10	b)	44a	piperidine	2.0 ml	44b
		(2.00)	DCM	20 ml	(1.23)
			20 °C	2 h	
15			silica-gel		
			(chloroform	-methanol)	
			50:1		
20	c)	44b	TFA	6.0 ml	44
		(0.25)	20 °C	2 h	(0.23)

Compound 44a:  $\left[\alpha\right]_{D}^{25}$  - 10.0° (c = 0.66 in chloroform)

Elemental Analysis for C <sub>72</sub> H <sub>117</sub> N <sub>5</sub> O <sub>13</sub> S:					
i	1 ' '	H, 9.12; H, 9.37;		S, 2.48 S, 2.49	

Compound 44b:  $[\alpha]_{0}^{25}$  - 15.2 • (c = 0.56 in chloroform)

Elemental Analysis for C <sub>57</sub> H <sub>107</sub> N <sub>5</sub> O <sub>11</sub> S:						
Calcd.	C, 63.95;	H, 10.07;	N, 6.54;	S, 3.00		
Found	C, 63.70;	H, 10.20;	N, 6.50;	S, 2.99		

Compound 44:  $[\alpha]_{D}^{25} + 5.5^{\circ}$  (c = 0.62 in 5% TFA-chloroform)

Elemental Analysis for C <sub>48</sub> H <sub>91</sub> N <sub>5</sub> O <sub>9</sub> S•2TFA:					
	Calcd.	C, 54.67;	H, 8.21;	N, 6.13;	S, 2.81
	Found	C, 54.29;	H, 8.19;	N, 6.47;	S, 3.06

Example 53

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## Preparation of sodium salt

1) Compound 12 (5.0 g) was dissolved in 20% (v/v) acetonitrile-0.5% (w/v) sodium hydrogen carbonate aqueous solution (5 L) at 40 °C. After adjusting pH of the solution at 9.5, the solution was subjected to a chromatography on Diaion HP-20 (1.0 L, Mitsubishi Kasei Corp. Japan) which was previously swollen with 20% (v/v) aqueous acetonitrile. The resin was washed with 20% (v/v) aqueous acetonitrile (5 L), followed by development with 40% (v/v) aqueous acetonitrile (4 L) and 60% (v/v) aqueous acetonitrile (5 L). Eluate was concentrated and then freeze-dried to give a powdery product. The powdery product was suspended in acetone (150 ml). Insolubles were collected by filtration to give disodium salt of compound 12 (4.5 g) as a white powder.

FAB-Mass spectrum (M + H) = 959

Elemental Analysis for C <sub>47</sub> H <sub>84</sub> N <sub>4</sub> O <sub>11</sub> SNa <sub>2</sub> • 3H <sub>2</sub> O:							
Calcd.	C, 55.71;	H, 8.95;	N, 5.53;	S, 3.16;	Na, 4.54		
Found	C, 55.90;	H, 9.39;	N, 5.34;	S, 3.16;	Na, 4.78		

2) Compound 19 (11.0 g) was dissolved in 15% (v/v) acetonitrile-0.5% (w/v) sodium hydrogen carbonate aqueous solution (5 L) at  $40 \,^{\circ}$  C. After adjusting pH of the solution at 9.5, the solution was subjected to a chromatography on Diaion HP-20 (1.0 L, Mitsubishi Kasei Corp. Japan) which was previously swollen with 15% (v/v) aqueous acetonitrile. The resin was washed with 15% (v/v) aqueous acetonitrile (5 L), followed by development with 40% (v/v) aqueous acetonitrile (12 L). Eluate was concentrated and then freeze-dried to give a powdery product. The powdery product was suspended in acetone (200 ml). Insolubles were collected by filtration to give disodium salt of compound 23 (9.9 g) as a white powder. FAB-Mass spectrum (M+H) = 902

Elemental Analysis for C <sub>45</sub> H <sub>81</sub> N <sub>3</sub> O <sub>10</sub> SNa <sub>2</sub> • 3H <sub>2</sub> O:							
	C, 56.52; C, 56.61;			S, 3.35; S, 3.35;	Na, 4.81 Na, 5.25		

3) Compound 23 (14.0 g) was dissolved in 5% (v/v) acetonitrile-0.5% (w/v) sodium hydrogen carbonate aqueous solution (7 L) at  $40 \, ^{\circ}$ C. After adjusting pH of the solution at 9.5, the solution was subjected to a chromatography on Diaion HP-20 (1.4 L, Mitsubishi Kasei Corp. Japan) which was previously swollen with 5% (v/v) aqueous acetonitrile. The resin was washed with 5% (v/v) aqueous acetonitrile (7 L), followed by development with 40% (v/v) aqueous acetonitrile (8.5 L). Eluate was concentrated and then freeze-dried to give a powdery product. The powdery product was suspended in acetone (600 ml). Insolubles were collected by filtration to give trisodium salt of compound 23 (11.5 g) as a white powder. FAB-Mass spectrum (M+H) = 1053

Elemental Analysis for C <sub>50</sub> H <sub>87</sub> N <sub>4</sub> O <sub>13</sub> SNa <sub>3</sub> • 4H <sub>2</sub> O:							
Calcd.	C, 53.37;	H, 8.51;	N, 4.98;	S, 2.85;	Na, 6.13		
Found	C, 53.48;	H, 8.81;	N, 4.89;	S, 3.03;	Na, 6.0		

4) Compound 25 (8.0 g) was dissolved in 5% (v/v) acetonitrile-0.5% (w/v) sodium hydrogen carbonate aqueous solution (4 L) at 40 °C. After adjusting pH of the solution at 9.5, the solution was subjected to a chromatography on Diaion HP-20 (1.5 L, Mitsubishi Kasei Corp. Japan) which was previously swollen with 5% (v/v) aqueous acetonitrile. The resin was washed with 5% (v/v) aqueous acetonitrile (7.5 L), followed by development with 40% (v/v) aqueous acetonitrile (6 L). Eluate was concentrated and then freeze-dried to give a powdery product. The powdery product was suspended in acetone (200 ml). Insolubles were collected by filtration to give trisodium salt of compound 25 (7.0 g) as a white powder. FAB-Mass spectrum (M+H)=1053

Elemental Analysis for C <sub>50</sub> H <sub>87</sub> N <sub>4</sub> O <sub>13</sub> SNa <sub>3</sub> •2.5H <sub>2</sub> O:							
Calcd.	C, 54.68;	H, 8.44;	N, 5.10;	S, 2.92;	Na, 6.28		
Found	C, 54.48;	H, 8.64;	N, 5.09;	S, 2.79;	Na, 6.13		

5) Compound 7 (200 mg) was dissolved in 30% (v/v) methanol-0.5% (w/v) sodium hydrogen carbonate aqueous solution (300 ml) at 40°C. After adjusting pH of the solution at 9.5, the solution was subjected to a chromatography on Diaion HP-20 (50 ml, Mitsubishi Kasei Corp. Japan) which was previously swollen with 30% (v/v) aqueous methanol. The resin was washed with 30% (v/v) aqueous methanol (200 ml) and 50% (v/v) aqueous methanol (200 ml), followed by development with 80% (v/v) aqueous methanol (600 ml). Eluate was concentrated and then freeze-dried to give a powdery product. The powdery product was suspended in acetone (8 ml). Insolubles were collected by filtration to give

disodium salt of compound 7 (75 mg) as a white powder.

Elemental Analysis for C <sub>4 9</sub> H <sub>8 7</sub> H <sub>5</sub> O <sub>1 2</sub> SNa <sub>2</sub> • 2.5H <sub>2</sub> O:							
Calcd.	C, 55.45;	H, 8.74;	N, 6.60;	S, 3.02			
Found	C, 55.47;	H, 8.55;	N, 6.48;	S, 3.11			

Structural formulae of the compounds obtained in the above examples are shown in Table 1.

Table 1

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	Compound	Exam	ple Structural Formulae
	No.	No	
15	1	6	(2R,6S)-2-amino-6,7-bis(PamO)-4-THT-Gly-
			Gly-Gly-Glu-Thr-Thr-OH
	2	7	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-
20			Gly-Gly-Glu-Thr-Thr-OH
	3	8	(2S,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-
			Gly-Gly-Glu-Thr-Thr-OH
25	4	9	(2S,6S)-2-amino-6,7-bis(PamO)-4-THT-Gly-
			Gly-Gly-Glu-Thr-Thr-OH
	5	10	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-

			Gly-Gly-Glu-Thr-OH
	6	11	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu-
5			Gly-Glu-Gly-D-Glu-OH
J	7	12	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-
			Gly-Gly-Glu-OH
	8	13	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-
10		-	Gly-Gly-D-Glu-OH
	9	14	(2R,6S)-2-amino-6,7-bis(PamO)-4-THT-Gly-
			Gly-Gly-Glu-OH
	10	15	(2R,6S)-2-amino-6,7-bis(PamO)-4-THT-Gly-
15			Gly-Gly-D-Glu-OH
	11	16	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu-
			Gly-D-Glu-OH
20	12	17	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-
			Gly-Glu-OH
	13	18	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-
			Gly-Gly-OH
25	14	19	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-
			Glu-OH
	15	20	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu-
30			ОН
	16	21	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-
			Gly-Gly-Asp-OH
	17	22	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-
35			Gly-D-Glu-OH
	18	23	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-
			Gly-OH
40	19	24	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-
			D-Glu-OH
	20	25	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-D-
			Glu-OH
45	21	26	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Asp-
			ОН
	22	27	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-D-
50			Asp-OH
	23	28	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-

			Glu-Glu-OH
	24	29	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-
5			Glu-D-Glu-OH
	25	30	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu-
			Gly-Glu-OH
	26	31	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu-
10	•		Glu-OH
	27	32	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu-
			D-Glu-OH
15	28	33	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu-
			Glu-Glu-OH
	29	34	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Glu-
			Glu-D-Glu-OH
20	30	36	(2R,6R)-2-amino-6-hexanoyloxy-7-PamO-4-
			THT-Gly-Gly-Glu-OH
	31	37	(2R,6R)-2-Fmoc-amino-6,7-bis(PamO)-4-
25			THT-Glu-Gly-D-Glu-OH
25	32	38	(2R,6R)-2-acetylamino-6,7-bis(PamO)-4-
			THT-Glu-Gly-D-Glu-OH
	33	39	(2R,6R)-2-hexanoylamino-6-hexanoyloxy-7-
30			PamO-4-THT-Gly-Gly-Glu-OH
	34	42	(2R,6R)-2-amino-6,7-bis(SteO)-4-THT-Gly-
			Glu-Glu-OH
	35	43	(2R,6R)-2-amino-6,7-bis(MyrO)-4-THT-Gly-
35			Glu-Glu-OH
	36	44	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-
			NH(CH <sub>2</sub> ),CO-Glu-OH hydrochloride
40	37	45	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-
	4.40		NH(CH <sub>2</sub> ) <sub>11</sub> CO-Glu-OH hydrochloride
	38	46	(2R, 6R)-4-(2-amino-6, 7-bis(PamO)-4-THT-
			amino)benzoyl-Glu-OH hydrochloride
45	39	47	(2R, 6R)-4-(2-amino-6, 7-bis(PamO)-4-THT-
			Gly-amino)benzoyl-Glu-OH hydrochloride
	40	48	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-
50			NH(CH <sub>2</sub> ) <sub>5</sub> CO-Glu-OH TFA salt
	41	49	(2R,6R)-2-amino-6,7-bis(PamO)-4-THT-

NH(CH<sub>2</sub>)<sub>6</sub>NHCO-Glu-OH TFA salt

42 50 (2R,6R)-4-(2-amino-6,7-bis(PamO)-4-THT-aminomethyl)benzoyl-Glu-OH hydrochloride

43 51 (2R,6R)-4-(N-(2-Fmoc-amino-6,7-bis(PamO)-4-THT)-N-(carboxymethyl)aminomethyl)benzoyl-Glu-OH

44 52 (2R,6R)-2-amino-6,7-bis(PamO)-4-THT-Gly-

Retention time of Compounds 1 to 15 in high performance liquid chromatography is shown as follows:

Column:

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YMC-Pack A-602 NH<sub>2</sub> (Yamamura Chemical Laboratories, Japan)

Mobile phase:

85% methanol/0.02M phosphate buffer (pH 4.8)

Flow rate:

1.0 ml/min. UV, 214 nm

Detecting method: Retention time (minute):

Compound 1, 17.3;

Lys-Gly-OH 2TFA salt

Compound 2, 17.1; Compound 3, 15.9; Compound 4, 15.8; Compound 5, 18.5; Compound 7, 20.7; Compound 8, 19.7; Compound 9, 20.7; Compound 10, 19.8; Compound 12, 19.4; Compound 14, 9.4;

Compound 15, 18.4

Biological activities of the compound (I) are described as follows.

## Experimental Example 1

Actions of compound (I) produced in the foregoing Examples on enhancing proliferation of bone marrow cells of mice are shown in Table 2.

Table 2

Actions on enhanc	Actions on enhancing proliferation of bone marrow cells of mice					
Compound No.	Minimal Effective Concentration (MEC, ng/ml)*1					
1	0.625					
2	<0.156					
8	<0.156					
12	<0.156					
14	0.156					
36	<0.156					
44	<0.156					

<sup>&</sup>lt;sup>\*1</sup> Assuming that proliferation in the group, to which no test compound was added, was 1, concentrations at which 1.3 or more times as much proliferation was observed were taken.

#### Method of determination:

To an RPMI 1640 culture medium [Bio-Wittaker Inc. (hereinafter abbreviated as BW), U.S.A.] containing 2 x 10<sup>6</sup>/ml of bone marrow cells of BALB/c mice, 2mM of L-glutamine, 20 ug/ml of gentamicin (Flow Laboratories, Inc., Scotland), 10% fetal calf serum (BW, USA) was added a test compound in an adequate concentration, which was incubated at 37 °C for 3 days in 5% carbon dioxide in air followed by determination of proliferation of the bone marrow cells by the MTT reduction method [Tada et al., Journal of Immunological Methods, Vol. 93, p.157, 1986].

#### 10 Experimental Example 2

Actions of compound (I) on enhancing the number of spleen cells of mice are shown in Table 3.

Table 3

|--|

Actions on enhancing the number of spleen cells						
Drug Dosage Number of spleen (mg/kg/day) cells (%)						
Cyclophosphamide singly		33.4				
Cyclophosphamide + compound 8	0.1	78.6				

<sup>&</sup>lt;sup>\*1</sup> Number of cells in the mouse administered intraperitoneally with physiological saline solution containing 2% gum arabic is assumed as 100%.

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#### Method of Determination:

A BALB/c mouse was injected intraperitoneally with cyclophosphamide dissolved in physiological saline solution at a dose of 200 mg/kg. From the day after next, the animal was administered intraperitoneally for 4 consecutive days with the test compound suspended in physiological saline solution containing 2% gun arabic. On the day following completion of the administration, number of trypan blue chromophobic cells of spleen was counted.

#### 35 Experimental Example 3

Compound No.

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Actions of compound (I) on enhancing the number of leukocytes in mice are shown in Table 4.

Dosage (mg/kg/day)

0.063

0.031

0.13

Table 4

Actions on enhancing the number of leukocytes

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\*1) The value is shown by percentage relative to the number of leukocytes (assumed as 100%) of the mouse orally administered with physiological saline solution, in place of cyclophosphamide, at a dose of 0.2 ml relative to 20 g of body weight, then subcutaneously administered, from the next day of the oral administration, with 5% glucose once a day for 5 days. Incidentally, the average value and standard deviation of leukocyte numbers of mice orally administered with cyclophosphamide at a dose of 150 mg/kg, then from the next day, subcutaneously administered with 0.2 ml of 5% glucose relative to 20 g of body weight once a day for 5 days were 41 ± 11% throughout the experiment.

Number of leukocytes \*1

102

129

118

#### Method of Determination:

Six week old female CDF1/Crj mice (5 animals/group) were subjected to the experiment. Each animal was orally administered with cyclophosphamide dissolved in physiological saline solution at a dose of 150 mg/kg. From the next day, each animal was administered subcutaneously with the compound suspended in 5% glucose at the following dosages for five days once a day. On the next day after completion of the administration, about 100µl of peripheral blood was collected from orbital vein using an EDTA-treated glass capillary. Then, the number of lekocytes was counted using a automatic cell analyzer (Sysmex K-2000, Toa Medical Electronics, Japan)

#### Experimental Example 4

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Actions of compound (I) produced in the foregoing Examples on enhancing development of megakaryocyte colonies are shown in Table 5.

Table 5

Test Compound No.	Concentration (ng/ml)	Degree of megakaryocyte colony formation *1
12	0.4	1.82
	4	2.01

\*1 Number of megakaryocyte colonies in the group to which compound (I) is not added is shown as 1.

## Method of determination:

The production of megakaryocyte colonies was studied according to a plasma clot method [Mizoguti et al., Experimental Hematology, Vol. 7, pp.345 to 351, 1979].

To a NCTC-109 culture medium (Gibco BRL Inc, USA) containing 5 x 10<sup>6</sup>/ml of bone marrow cells of BALB/c mice, 20% fetal calf serum (BW, USA), 1% serum albumin, 0.026 mg/ml CaCl<sub>2</sub>, 0.02 mg/ml L-asparagine, 10% bovine sodium citrated plasma was added a test compound in an adequate concentration and then thoroughly mixed. A 0.4 ml aliquot thereof was placed in the center of plastic dish 35 x 10 mm (A/S Nunc, Denmark) and allowed to clot. To the culture dish was added 0.6 ml of α-MEM culture medium (BW, USA) containing 10% fetal calf serum and then the culture was incubated at 37°C in 5% CO<sub>2</sub> in air during 7 days. After incubation, culture medium surrounding the plasma clot was removed and the plasma clot was dehydrated by placing a piece of filter paper on its surface. Then 1 to 2 drops of 5% glutaraldehyde were dropped on the remaining filter paper over the clot. After ten minutes, the piece of filter paper was removed and then the clot was washed with 0.1 M phosphate buffer (pH 6.0). The clot was subjected acetylcholinesterase (hereinafter abbreviated as AchE) dyeing. To 0.1M phosphate buffers (45 ml) containing acetylthic choline iodide (30 mg) was added 30mM copper sulfate aqueous solution (6 ml) and 0.1M sodium citrate aqueous solution (3 ml) and 5 mM potassium ferricyanide aqueous solution (6 ml) in order. Two ml of each of the resulting solution was added to the plate. After standing at 33°C for 4 to 6 hours, counted the number of the megakaryocte colonies.

## Experimental Example 5

Actions of compound (I) produced in the foregoing Examples on enhancing development of AchE positive cells are shown in Table 6.

#### Table 6

Test Compound No.	Concentration (ng/ml)	Degree of AchE positive cell *1		
7	0.1	1.51		
12	0.1	1.43		
23	0.1	1.67		

<sup>&</sup>quot;1 AchE activity in the group to which compound (I) is not added is shown as 1.

#### Method of determination:

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The AchE activity was studied according to a fluorescence method (Ishibashi et al., Proceedings of the National Academy of Science U.S.A., Vol. 86, pp. 5953 to 5957, 1989).

Nonadherent bone marrow cells of BALB/c mice (1 x 10<sup>6</sup> cells/ml) were suspended in Iscove's modification of Dulbecco's medium (Gibco BRL Inc, USA) containing 1% Neutridoma-SP (Boheringer Mannhaim, Germany). Twenty-five  $\mu$ I of a test compound solution in an adequate concentration was inoculated in a 96-well plate. To each of this solution in the well was added 100  $\mu$ I of the nonadherent bone marrow cells suspension. After incubation at 37°C for 5 days, 25  $\mu$ I of 6% glutaladehyde aqueous solution was added to the well. After standing at 4°C for 30 minutes the mixture was centrifuged (850 x g) at 5°C and the supernatant was removed. After washing the deposit in the well with 100  $\mu$ I of phosphate buffered saline, to the deposit was added 100  $\mu$ I of buffer (pH 7.5) containing 0.2% polyoxyethylene-10-octylphenyl etherd (POPE). 1mM ethylenediaminetetraacetate (EDTA), 0.12M sodium chloride and 50 mM N-2-hydroxyethylpiperadine-N'-2-ethansulfonic acid (HEPES). To this was added 10  $\mu$ I of acetylthiocholine iodide aqueous solution (concentration: 1.6 mg/ml). After incubation at 33°C for 3 hours 10  $\mu$ I of the resulting solution was separated. To this was added 10  $\mu$ I of 0.4 mM 7-diethylamino-3-(4'-maleimidylphenyl)-4-methylcoumrin aqueous solution and 100  $\mu$ I of buffer (pH 5.0) containing 0.2% POPE, 1mM EDTA and 50 mM sodium acetate and fluorescence emission was determined on a fluorometer (excitation wave length: 365 nm, fluorescence wave length: 450 nm).

## Experimental Example 6

#### 5 Toxicity test

No mouse administered intraperitoneally with Compound 3 at a dose of 100 mg/kg was found dead.

## Experimental Example 7

## Toxicity test

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No mouse administered subcutaneously with disodium salt of Compound 7 at a dose of 100 mg/kg was found dead.

Compound (I) or a salt thereof is low in toxicity and can be used safely.

As is clear from the foregoing experimental Examples, Compound (I) or a salt thereof has an activity of remarkably improving hematopoietic disorder, which can be utilized as a therapeutic or prophylactic agent of leukocytopenia caused by radiotherapy or chemotherapy of cancers in mammals (e.g. dog, cat, cow, horse, monkey, man, etc.), as an hematopoietic-stimulating agent in the case of bone marrow transplantation, as an immunological enhancing agent having an action of increasing leukocytes, and, further as a therapeutic agent of thrombocytopenia.

#### Formulation Example 1

The compound 2 (4 g) obtained in Example 8 and mannitol (50 g) were dissolved in sterilized distilled water (1 liter) containing polyethylene glycol 400 (30% w/w). The solution was subjected to filtration under sterilization, and 1 ml each of which was then distributed in one ampoule to prepare intravenous injection containing 4 mg of the compound 2 per ampoule.

## Formulation Example 2

The trisodium salt of compound 23 (40 mg) obtained in Example 53 and mannitol (50 g) were dissolved in sterilized distilled water (1 liter) containing polyethylene glycol 400 (30% w/w). The solution was subjected to filtration under sterilization, and 1 ml each of which was then distributed in one ampoule to prepare intravenous injection containing 40  $\mu$ g of the trisodium salt of compound 23 per ampoule.

#### Claims

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### 10 1. A compound of the formula:

 $OR^2$   $CH_2-S-CH_2-CH-CH_2-O-R^1$  A-CH-CO-X-OH

wherein each of R¹ and R² is hydrogen or aliphatic acyl, A is amino which may be protected, X is an amino acid sequence consisting of 1 to 10 amino acid residues which contain at least one amino acid residue having a water-solubility enhancing group, or a salt thereof.

- 2. A compound according to claim 1, wherein A is amino.
- 3. A compound according to claim 1, wherein A is amino substituted with substituted oxycarbonyl.
- 4. A compound according to any one of claims 1 to 3, wherein the amino acid residue having a water-solubility enhancing group is an acidic amino acid residue.
- 5. A compound according to any one of claims 1 to 3, wherein the amino acid residue having a water-solubility enhancing group is a basic amino acid residue.
- 6. A compound according to any one of claims 1 to 5, wherein aliphatic acyl is  $C_{2-30}$  aliphatic acyl.
- 7. A compound according to any one of claims 1 to 6, wherein at least one of R¹ and R² is aliphatic acyl.
- 8. A compound according to any one of claims 1 to 7, wherein R<sup>1</sup> is aliphatic acyl.
- 40 9. A compound according to any one of claims 1 to 7, wherein R<sup>2</sup> is aliphatic acyl.
  - **10.** A compound according to claim 1, wherein the compound is (2R,6R)-2-amino-6,7-bis-(hexadecanoyloxy)-4-thiaheptanoyl-glycyl-glutamyl-glutamic acid.
- 45 11. A compound according to claim 1, wherein the compound is (2R,6R)-2-amino-6,7-bis-(hexadecanoyloxy)-4-thiaheptanoyl-glycyl-glycyl-glutamic acid.
  - 12. A compound according to claim 1, wherein the compound is (2R,6R)-2-amino-6,7-bis-(hexadecanoyloxy)-4-thiaheptanoyl-glutamyl-glycyl-glutamic acid.
  - 13. A compound according to claim 1, wherein the compound is (2R,6R)-2-amino-6,7-bis-(hexadecanoyloxy)-4-thiaheptanoyl-glycyl-D-glutamic acid.
  - 14. A compound as claimed in any one of claims 1 to 13 for the use as a therapeutic agent.

15. An immuno-stimulating composition having a leukocyte-increasing action, which comprises a compound or a salt thereof as claimed in any one of claims 1 to

- 16. An immuno-stimulating composition according to claim 15, wherein at least one of R¹ and R² is aliphatic acyl.
- 17. A composition for treating thrombocytopenia, which comprises a compound or a salt thereof as claimed in any one of claims 1 to 13.
  - 18. A method of producing the compound or a salt thereof as claimed in claim 1, which comprises subjecting a compound of the formula:

$$OR^{2}$$

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 $CH_{2}$ -S- $CH_{2}$ -CH- $CH_{2}$ -O- $R^{1}$ 

|
 $A$ -CH- $CO$ -X'- $OR^{3}$ 

wherein each of R<sup>1</sup> and R<sup>2</sup> is hydrogen or aliphatic acyl, R<sup>3</sup> is a protecting group, A is amino which may be protected, X' is an amino acid sequence consisting of 1 to 10 optionally protected amino acid residues which contain at least one optionally protected amino acid residue having a water-solubility enhancing group, or its salt, to a deprotection reaction.

- 19. Use of a compound according to any one of claims 1 to 13 for the preparation of a therapeutic agent for immuno stimulation with a leukocyte-increasing action.
- 20. Use of a compound according to any one of claims 1 to 13 for the preparation of a therapeutic agent for treating thrombocytopenia.

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## **EUROPEAN SEARCH REPORT**

Application Number EP 93 12 0970

DOCUMENTS CONSIDERED TO BE RELEVANT				
Category	<del></del>	dication, where appropriate,	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.CL5)
x	EP-A-0 000 330 (CIB 1979 * claims 1,4,5 *	A-GÉIGY AG) 24 January	1,4-9, 14-18	C07K5/06 C07K5/08 C07K5/10 C07K7/06
X	1980	A-GEIGY AG) 3 September; claim 1; examples 7,9	1,4-9, 14-18	A61K37/02
x	EP-A-0 114 787 (CIB 1984 * claim 1; examples	27,31 *	1,4-9, 14-18	
Р,Х	EP-A-0 548 024 (CIB 1993 * claim 1; examples		1,4,6-9, 14-18	
X	immunogen preparati	ynthesis of novel ive erylcysteinyl ful intermediates for ons'	1,4-9, 14-18	TECHNICAL FIELDS SEARCHED (Int.Cl.5) CO7K A61K
Y	* figure 3; table 1 *peptides 4,7,9* * figure 3; table 1		1-18	
Y	INT.J.PEPTIDE PROTE vol. 38, no. 6, 19 pages 545 - 554 J.W.METZGER ET AL. N(alpha)-Fmoc prote S-(2,3-dihydroxypro application in pept * page 547; example	91 'Synthesis of cted derivatives of pyl)-cysteine and their ide synthesis'	1-18	
	The present search report has be	een drawn up for all claims		
	Place of search	Date of completion of the search	<del></del>	Executar
	MUNICH	31 March 1994	Def	fner, C-A
CATEGORY OF CITED DOCUMENTS  T: theory or principal E: earlier patent & after the filling of the filling of the same category  A: technological background  O: non-written disclosure  P: intermediate document  document  A: technological background  A: technological background  O: non-written disclosure  document  document  document  document  document		cument, but publication or other reasons	ished on, or	



# **EUROPEAN SEARCH REPORT**

Application Number EP 93 12 0970

Category	Citation of document with of relevant p	indication, where appropriate,	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.CL.5)
Y	PEPTIDE CHEMISTRY pages 361 - 366 M.KURIMURA ET AL. Synthesis and MitogLipopeptide WS1279 * page 362; example	'Stereospecific genic Activity of and its Derivatives'	1-18	
		•		
				TECHNICAL FIELDS SEARCHED (Int.Cl.5)
		, <del></del>		
	-			
	•			
	The present search report has b	een drawn up for all claims		
Place of search MUNICH		Date of completion of the search 31 March 1994	Deffner, C-A	
X : partic Y : partic docui	ATEGORY OF CITED DOCUMES  cularly relevant if taken alone cularly relevant if combined with and nent of the same category logical background	NTS T: theory or print E: earlier patent efter the fillin other D: document cit	ciple underlying the i document, but publis g date ed in the application ad for other reasons	nvention

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